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(54) Title: INHIBITORS OF FACTOR Xa

(57) Abstract: Novel compounds, their salts and compositions related thereto having activity against mammalian factor Xa are disclosed. The compounds are useful *in vitro* or *in vivo* for preventing or treating coagulation disorders.

INHIBITORS OF FACTOR Xa

Related Applications

This application claims benefit of priority under 35 USC § 119(e) to U.S. Provisional Application No. 60/135,820 filed on May 24, 1999, which is herein incorporated in its entirety by reference.

Field of the Invention

This invention relates to novel compounds which are potent and highly selective inhibitors of isolated factor Xa or when assembled in the prothrombinase complex. These compounds show selectivity for factor Xa versus other proteases of the coagulation (e.g. thrombin, fVIIa, fIXa) or the fibrinolytic cascades (e.g. plasminogen activators, plasmin). In another aspect, the present invention relates to novel monoamidino-containing compounds, their pharmaceutically acceptable salts, and pharmaceutically acceptable compositions thereof which are useful as potent and specific inhibitors of blood coagulation in mammals. In yet another aspect, the invention relates to methods for using these inhibitors as therapeutic agents for disease states in mammals characterized by coagulation disorders.

Background of the Invention

Hemostasis, the control of bleeding, occurs by surgical means, or by the physiological properties of vasoconstriction and coagulation. This invention is particularly concerned with blood coagulation and ways in which it assists in maintaining the integrity of mammalian circulation after injury, inflammation, disease, congenital defect, dysfunction or other disruption. Although platelets and blood coagulation are both involved in thrombus formation, certain components of the coagulation cascade are primarily responsible for the amplification or acceleration of the processes involved in platelet aggregation and fibrin deposition.

Thrombin is a key enzyme in the coagulation cascade as well as in hemostasis. Thrombin plays a central role in thrombosis through its ability to catalyze the conversion of fibrinogen into fibrin and through its potent platelet activation activity. Direct or indirect inhibition of thrombin activity has been the focus of a variety of recent anticoagulant strategies as reviewed by Claeson, G.,

"Synthetic Peptides and Peptidomimetics as Substrates and Inhibitors of Thrombin and Other Proteases in the Blood Coagulation System", Blood Coag. Fibrinol. 5, 411-436 (1994). Several classes of anticoagulants currently used in the clinic directly or indirectly affect thrombin (i.e. heparins, low-molecular weight heparins, heparin-like compounds and coumarins).

A prothrombinase complex, including Factor Xa (a serine protease, the activated form of its Factor X precursor and a member of the calcium ion binding, gamma carboxyglutamyl (Gla)-containing, vitamin K dependent, blood coagulation glycoprotein family), converts the zymogen prothrombin into the active procoagulant thrombin. Unlike thrombin, which acts on a variety of protein substrates as well as at a specific receptor, factor Xa appears to have a single physiologic substrate, namely prothrombin. Since one molecule of factor Xa may be able to generate up to 138 molecules of thrombin (Elodi et al., *Thromb. Res.* 15, 617-619 (1979)), direct inhibition of factor Xa as a way of indirectly inhibiting the formation of thrombin may be an efficient anticoagulant strategy. Therefore, it has been suggested that compounds which selectively inhibit factor Xa may be useful as *in vitro* diagnostic agents, or for therapeutic administration in certain thrombotic disorders, see *e.g.*, WO 94/13693.

Polypeptides derived from hematophagous organisms have been reported which are highly potent and specific inhibitors of factor Xa. United States Patent 4,588,587 describes anticoagulant activity in the saliva of the Mexican leech, *Haementeria officinalis*. A principal component of this saliva was shown to be the polypeptide factor Xa inhibitor, antistasin (ATS), by Nutt, E. *et al.*, "The Amino Acid Sequence of Antistasin, a Potent Inhibitor of Factor Xa Reveals a Repeated Internal Structure", J. Biol. Chem., 263, 10162-10167 (1988). Another potent and highly specific inhibitor of Factor Xa, called tick anticoagulant peptide (TAP), has been isolated from the whole body extract of the soft tick *Ornithidoros moubata*, as reported by Waxman, L., *et al.*, "Tick Anticoagulant Peptide (TAP) is a Novel Inhibitor of Blood Coagulation Factor Xa" Science, 248, 593-596 (1990).

Factor Xa inhibitory compounds which are not large polypeptide-type inhibitors have also been reported including: Tidwell, R.R. *et al.*, "Strategies for Anticoagulation With Synthetic Protease Inhibitors. Xa Inhibitors Versus Thrombin Inhibitors", Thromb. Res., 19, 339-349 (1980); Turner, A.D. *et al.*, "p-Amidino Esters as Irreversible Inhibitors of Factor IXa and Xa and Thrombin", Biochemistry,

25, 4929-4935 (1986); Hitomi, Y. *et al.*, "Inhibitory Effect of New Synthetic Protease Inhibitor (FUT-175) on the Coagulation System", Haemostasis, 15, 164-168 (1985); Sturzebecher, J. *et al.*, "Synthetic Inhibitors of Bovine Factor Xa and Thrombin. Comparison of Their Anticoagulant Efficiency", Thromb. Res., 54, 245-252 (1989); Kam, C.M. *et al.*, "Mechanism Based Isocoumarin Inhibitors for Trypsin and Blood Coagulation Serine Proteases: New Anticoagulants", Biochemistry, 27, 2547-2557 (1988); Hauptmann, J. *et al.*, *"Comparison* of the Anticoagulant and Antithrombotic Effects of Synthetic Thrombin and Factor Xa Inhibitors", Thromb. Haemost., 63, 220-223 (1990); and the like.

Others have reported Factor Xa inhibitors which are small molecule organic compounds, such as nitrogen containing heterocyclic compounds which have amidino substituent groups, wherein two functional groups of the compounds can bind to Factor Xa at two of its active sites. For example, WO 98/28269 describes pyrazole compounds having a terminal C(=NH)-NH_2 group; WO 97/21437 describes benzimidazole compounds substituted by a basic radical which are connected to a naththyl group via a straight or branched chain alkylene,-C(=O) or -$S(=O)_2$ bridging group; WO 99/10316 describes compounds having a 4-phenyl-N-alkylamidino-piperidine and 4-phenoxy-N-alkylamidino-piperidine group connected to a 3-amidinophenyl group via a carboxamidealkyleneamino bridge; and EP 798295 describes compounds having a 4-phenoxy-N-alkylamidino-piperidine group connected to an amidinonaphthyl group via a substituted or unsubstituted sulfonamide or carboxamide bridging group.

There exists a need for effective therapeutic agents for the regulation of hemostasis, and for the prevention and treatment of thrombus formation and other pathological processes in the vasculature induced by thrombin such as restenosis and inflammation. In particular, there continues to be a need for compounds which selectively inhibit factor Xa or its precursors. Compounds that have different combinations of bridging groups and functional groups than compounds previously discovered are needed, particularly compounds which selectively or preferentially bind to Factor Xa. Compounds with a higher degree of binding to Factor Xa than to thrombin are desired, especially those compounds having good bioavailability and/or solubility.

Summary of the Invention

The present invention relates to novel compounds which inhibit factor Xa, their pharmaceutically acceptable isomers, salts, hydrates, solvates and prodrug derivatives, and pharmaceutically acceptable compositions thereof which have particular biological properties and are useful as potent and specific inhibitors of blood coagulation in mammals. In another aspect, the invention relates to methods of using these inhibitors as diagnostic reagents or as therapeutic agents for disease states in mammals which have coagulation disorders, such as in the treatment or prevention of any thrombotically mediated acute coronary or cerebrovascular syndrome, any thrombotic syndrome occurring in the venous system, any coagulopathy, and any thrombotic complications associated with extracorporeal circulation or instrumentation, and for the inhibition of coagulation in biological samples.

In certain embodiments, this invention relates to novel compounds which are potent and highly selective inhibitors of isolated factor Xa when assembled in the prothrombinase complex. These compounds show selectivity for factor Xa versus other proteases of the coagulation cascade (e.g. thrombin, etc.) or the fibrinolytic cascade, and are useful as diagnostic reagents as well as antithrombotic agents.

In a preferred embodiment, the present invention provides a compound of the formula I:

A-Y-D-E-G-J-Z-L

wherein:

A is selected from:

(a) C_1-C_6-alkyl;

(b) C_3-C_8-cycloalkyl;

(c) $-NR^2R^3$, $R^3C(=NR^2)-$, $R^2R^3N-C(=NR^2)-$, $R^2R^3N-C(=NR^2)-N(R^3)-$

(d) phenyl, which is independently substituted with 0-2 R^1 substituents;

(e) naphthyl, which is independently substituted with 0-2 R^1 substituents;and

(f) a monocyclic or fused bicyclic heterocyclic ring system having from 5 to 10 ring atoms, wherein 1-4 ring atoms of the ring system are selected from N, O and S, and wherein the ring system may be subsituted from 0-2 R^1 substituents;

R^1 is selected from:

Halo, C_{1-4}alkyl, C_{2-6}alkenyl, C_{2-6}alkynyl, C_{3-8}cycloalkyl, C_{0-4}alkylC_{3-8}cycloalkyl,-CN, $-NO_2$, R2-C(=NR3)-, R2R3N-C(=NR2)-, $(CH_2)_mNR^2R^3$, $SO_2NR^2R^3$, SO_2R^2, CF_3, OR^2, and a 5-6 membered aromatic heterocyclic system containing from 1-4 heteroatoms selected from N, O and S, wherein from 1-4 hydrogen atoms on the aromatic heterocyclic system may be independently replaced with a member selected from the group consisting of halo, C_1-C_4-alkyl, -CN C_{1-4}alkyl, C_{2-6}alkenyl, C_{2-6}alkynyl, C_{3-8}cycloalkyl, C_{0-4}alkylC_{3-8}cycloalkyl and $-NO_2$;

R^2 and R^3 are independently selected from the group consisting of:

H, OR2, NR2R3, C_{1-4}alkyl, C_{2-6}alkenyl, C_{2-6}alkynyl, C_{3-8}cycloalkyl, C_{0-4}alkylC_{3-8}cycloalkyl, C_{0-4}alkylphenyl and C_{0-4}alkylnaphthyl, wherein from 1-4 hydrogen atoms on the ring atoms of the phenyl and naphthyl moieties may be independently replaced with a member selected from the group consisting of halo, C_{1-4}alkyl, C_{2-6}alkenyl, C_{2-6}alkynyl, C_{3-8}cycloalkyl, C_{0-4}alkylC_{3-8}cycloalkyl, -CN, and $-NO_2$;

m is an integer of 0-2;

Y is a member selected from the group consisting of:

a direct link, -CH2-, -C(=O)-, -N(R^4)-, -N(R^4)CH2-, -C=N(R4)-, -C(=O)-N(R^4)-, -N(R^4)-C(=O)-, $-SO_2$-, -O-, $-SO_2$-N(R^4)- and –N(R^4)-SO_2-;

R^4 is selected from:

H, C_{1-4}alkyl, C_{2-6}alkenyl, C_{2-6}alkynyl, C_{3-8}cycloalkyl, C_{0-4}alkylC_{3-8}cycloalkyl, C_{0-4}alkylphenyl and C_{0-4}alkylnaphthyl, wherein from 1-4 hydrogen atoms on the ring atoms of the phenyl and naphthyl moieties may be independently replaced with a member selected from the group consisting of halo, C_{1-4}alkyl, C_{2-6}alkenyl, C_{2-6}alkynyl, C_{3-8}cycloalkyl, C_{0-4}alkylC_{3-8}cycloalkyl, -CN, and $-NO_2$;.

D is a direct link or is a member selected from the group consisting of:

(a) phenyl, which is independently substituted with 0-2 R^{1a} substituents;

(b) naphthyl, which is independently substituted with 0-2 R^{1a} substituents; and

(c) a monocyclic or fused bicyclic heterocyclic ring system having from 5 to 10 ring atoms, wherein 1-4 ring atoms of the ring system are selected from N, O and S, and wherein the ring system may be subsituted from 0-2 R^{1a} substituents;

R^{1a} is selected from:

Halo, C_{1-4}alkyl, C_{2-6}alkenyl, C_{2-6}alkynyl, C_{3-8}cycloalkyl, C_{0-4}alkylC_{3-8}cycloalkyl, -CN, $-NO_2$, $(CH_2)_mNR^{2a}R^{3a}$, $SO_2NR^{2a}R^{3a}$, SO_2R^{2a}, CF_3, OR^{2a}, and a 5-6 membered aromatic heterocyclic system containing from 1-4 heteroatoms selected from N, O and S, wherein from 1-4 hydrogen atoms on the aromatic heterocyclic system may be independently replaced with a member selected from the group consisting of halo, C_{1-4}alkyl, C_{2-6}alkenyl, C_{2-6}alkynyl, C_{3-8}cycloalkyl, C_{0-4}alkylC_{3-8}cycloalkyl, -CN and $-NO_2$;

R^{2a} and R^{3a} are independently selected from the group consisting of:

H, C_{1-4}alkyl, C_{2-6}alkenyl, C_{2-6}alkynyl, C_{3-8}cycloalkyl, C_{0-4}alkylC_{3-8}cycloalkyl, C_{0-4}alkylphenyl and C_{0-4}alkylnaphthyl, wherein from 1-4 hydrogen atoms on the ring atoms of the phenyl and naphthyl moieties may be independently replaced with a member selected from the group consisting of halo, C_{1-4}alkyl, C_{2-6}alkenyl, C_{2-6}alkynyl, C_{3-8}cycloalkyl, C_{0-4}alkylC_{3-8}cycloalkyl, -CN and $-NO_2$;.

E is a member selected from the group consisting of:

$-N(R^5)-C(=O)-$, $-C(=O)-N(R^5)-$, $-N(R^5)-C(=O)-N(R^6)-$, $-SO_2-N(R^5)-$, $-N(R^5)-SO_2-N(R^6)-$ and $-N(R^5)-SO_2-N(R^6)-C(=O)-$;

R^5 and R^6 are independently selected from:

H, C_{1-4}alkyl, C_{2-6}alkenyl, C_{2-6}alkynyl, C_{3-8}cycloalkyl, C_{0-4}alkylC_{3-8}cycloalkyl, C_{0-4}alkylphenyl, C_{0-4}alkylnaphthyl, C_{0-4}alkylheteroaryl, C_{1-4}alkylCOOH and C_{1-4}alkylCOOC_{1-4}alkyl, wherein from 1-4 hydrogen atoms on the ring atoms

of the phenyl, naphthyl and heteroaryl moieties may be independently replaced with a member selected from the group consisting of halo, C_{1-4}alkyl, C_{2-6}alkenyl, C_{2-6}alkynyl, C_{3-8}cycloalkyl, C_{0-4}alkylC_{3-8}cycloalkyl, -CN and $-NO_2$;

G is selected from:

$-CR^7R^8-$ and $-CR^{7a}R^{8a}-CR^{7b}R^{8b}-$

wherein R^7, R^8, R^{7a}, R^{8a}, R^{7b} and R^{8b} are independently a member selected from from the group consisting of:

hydrogen, C_{1-4}alkyl, C_{2-6}alkenyl, C_{2-6}alkynyl, C_{3-8}cycloalkyl, C_{0-4}alkyl-C_{3-8}cycloalkyl, C_{0-4}alkylphenyl, C_{0-4}alkylnaphthyl $-C_{0-4}$alkylCOOR^9, $-C_{0-4}$alkylC(=O)NR^{9}R^{10}, $-C_{0-4}$alkylC(=O)NR^9-CH_2-CH_2-O-R^{10}, $-C_{0-4}$alkylC(=O)NR^9(-CH_2-CH_2-O-R^{10}-)$_2$, -N(R^9)COR^{10}, -N(R^9)C(=O)R^{10}, -N(R^9)SO_2R^{10}, and a naturally occurring or synthetic amino acid side chain, wherein from 1-4 hydrogen atoms on the ring atoms of the phenyl and naphthyl moieties may be independently replaced with a member selected from the group consisting of halo, C_{1-4}alkyl, C_{2-6}alkenyl, C_{2-6}alkynyl, C_{3-8}cycloalkyl, C_{0-4}alkyl-C_{3-8}cycloalkyl, -CN and $-NO_2$;

R^9 and R^{10} are independently selected from:

H, C_{1-4}alkyl, C_{0-4}alkylphenyl and C_{0-4}alkylnaphthyl, wherein from 1-4 hydrogen atoms on the ring atoms of the phenyl and naphthyl moieties may be independently replaced with a member selected from the group consisting of halo, C_{1-4}alkyl, C_{2-6}alkenyl, C_{2-6}alkynyl, C_{3-8}cycloalkyl, C_{0-4}alkyl-C_{3-8}cycloalkyl, -CN and $-NO_2$, and wherein R^9 and R^{10} taken together can form a 5-8 membered heterocylic ring;

J is a member selected from the group consisting of:

a direct link, $-C(=O)-N(R^{11})-(CH_2)_{0-2}$, $-N(R^{11})-(CH_2)_{0-2}-C(=O)-$, and $-N(R^{11})-(CH_2)_{0-2}$;

R^{11} is a member selected from the group consisting of:

hydrogen, C_{1-4}alkyl, C_{2-6}alkenyl, C_{2-6}alkynyl, C_{3-8}cycloalkyl, C_{0-4}alkyl-C_{3-8}cycloalkyl, C_{0-4}alkylphenyl, C_{0-4}alkylnaphthyl, C_{0-4}alkylheterocyclic ring having from 1 to 4 hetero ring atoms selected from the group consisting of N,

O and S, CH_2COOC_{1-4}alkyl, CH_2COOC_{1-4}alkylphenyl and CH_2COOC_{1-4}alkylnaphthyl;

G and J together can form a cyclic ring systems.

Z is a member selected from the group consisting of:

(a) phenyl, which is independently substituted with 0-2 R^{1b} substituents;

(b) naphthyl, which is independently substituted with 0-2 R^{1b} substituents; and

(c) a monocyclic or fused bicyclic heterocyclic ring system having from 5 to 10 ring atoms, wherein 1-4 ring atoms of the ring system are selected from N, O and S, and wherein the ring system may be subsituted from 0-2 R^{1b} substituents;

R^{1b} is selected from:

Halo, C_{1-4}alkyl, C_{2-6}alkenyl, C_{2-6}alkynyl, C_{3-8}cycloalkyl, C_{0-4}alkylC_{3-8}cycloalkyl, -CN, $-NO_2$, $NR^{2b}R^{3b}$, $SO_2NR^{2b}R^{3b}$, SO_2R^{2b}, CF_3, OR^{2b}, O-CH_2-Ph, O-CH_2-OPh, O-CH_2-CH_2-OR^{2b}, O-CH_2-$COOR^{2b}$, N(R^{2b})-CH_2-CH_2-OR^{2b}, N(-CH_2-CH_2-OR^{2b})$_2$, N(R^{2b})-C(=O)R^{3b}, N(R^{2b})-SO_2-R^{3b}, and a 5-6 membered aromatic heterocyclic system containing from 1-4 heteroatoms selected from N, O and S, wherein from 1-4 hydrogen atoms on the aromatic heterocyclic system may be independently replaced with a member selected from the group consisting of halo, C_{1-4}alkyl, C_{2-6}alkenyl, C_{2-6}alkynyl, C_{3-8}cycloalkyl, C_{0-4}alkylC_{3-8}cycloalkyl, -CN and $-NO_2$;

R^{2b} and R^{3b} are independently selected from the group consisting of:

H, C_{1-4}alkyl, C_{2-6}alkenyl, C_{2-6}alkynyl, C_{3-8}cycloalkyl, C_{0-4}alkylC_{3-8}cycloalkyl, C_{0-4}alkylphenyl and C_{0-4}alkylnaphthyl, wherein from 1-4 hydrogen atoms on the ring atoms of the phenyl and naphthyl moieties may be independently replaced with a member selected from the group consisting of halo, C_{1-4}alkyl, C_{2-6}alkenyl, C_{2-6}alkynyl, C_{3-8}cycloalkyl, C_{0-4}alkylC_{3-8}cycloalkyl, -CN and $-NO_2$;

L is selected from:

H, -CN, C(=O)$NR^{12}R^{13}$, $(CH_2)_nNR^{12}R^{13}$, C(=NR^{12})$NR^{12}R^{13}$, OR^{12}, $NR^{12}R^{13}$,

$-NR^{12}C(=NR^{12})NR^{12}R^{13}$, and $NR^{12}C(=NR^{12})-R^{13}$;

R^{12} and R^{13} are independently selected from:

hydrogen, $-OR^{14}$, $-NR^{14}R^{15}$, C_{1-4}alkyl, C_{0-4}alkylphenyl, C_{0-4}alkylnaphthyl, $COOC_{1-4}$alkyl, $COO-C_{0-4}$alkylphenyl and $COO-C_{0-4}$alkylnaphthyl, wherein from 1-4 hydrogen atoms on the ring atoms of the phenyl and naphthyl moieties may be independently replaced with a member selected from the group consisting of halo, C_{1-4}alkyl, C_{2-6}alkenyl, C_{2-6}alkynyl, C_{3-8}cycloalkyl, C_{0-4}alkylC_{3-8}cycloalkyl, -CN, and $-NO_2$;

R^{14} and R^{15} are independently selected from:

H, C_{1-4}alkyl, C_{2-6}alkenyl, C_{2-6}alkynyl, C_{3-8}cycloalkyl, C_{0-4}alkylC_{3-8}cycloalkyl, C_{0-4}alkylphenyl and C_{0-4}alkylnaphthyl, wherein from 1-4 hydrogen atoms on the ring atoms of the phenyl and naphthyl moieties may be independently replaced with a member selected from the group consisting of halo, C_{1-4}alkyl, C_{2-6}alkenyl, C_{2-6}alkynyl, C_{3-8}cycloalkyl, C_{0-4}alkylC_{3-8}cycloalkyl, -CN, and $-NO_2$;

and all pharmaceutically acceptable isomers, salts, hydrates, solvates and prodrug derivatives thereof.

In certain aspects of this invention, compounds are provided which are useful as diagnostic reagents. In another aspect, the present invention includes pharmaceutical compositions comprising a pharmaceutically effective amount of the compounds of this invention and a pharmaceutically acceptable carrier. In yet another aspect, the present invention includes methods comprising using the above compounds and pharmaceutical compositions for preventing or treating disease states characterized by undesired thrombosis or disorders of the blood coagulation process in mammals, or for preventing coagulation in biological samples such as, for example, stored blood products and samples. Optionally, the methods of this invention comprise administering the pharmaceutical composition in combination with an additional therapeutic agent such as an antithrombotic and/or a thrombolytic agent and/or an anticoagulant.

The preferred compounds also include their pharmaceutically acceptable isomers, hydrates, solvates, salts and prodrug derivatives.

Detailed Description of the Invention

Definitions

In accordance with the present invention and as used herein, the following terms are defined with the following meanings, unless explicitly stated otherwise.

The term "alkenyl" refers to a trivalent straight chain or branched chain unsaturated aliphatic radical. The term "alkinyl" (or "alkynyl") refers to a straight or branched chain aliphatic radical that includes at least two carbons joined by a triple bond. If no number of carbons is specified alkenyl and alkinyl each refer to radicals having from 2-12 carbon atoms.

The term "alkyl" refers to saturated aliphatic groups including straight-chain, branched-chain and cyclic groups having the number of carbon atoms specified, or if no number is specified, having up to 12 carbon atoms. The term "cycloalkyl" as used herein refers to a mono-, bi-, or tricyclic aliphatic ring having 3 to 14 carbon atoms and preferably 3 to 7 carbon atoms.

As used herein, the terms "carbocyclic ring structure " and " C_{3-16} carbocyclic mono, bicyclic or tricyclic ring structure" or the like are each intended to mean stable ring structures having only carbon atoms as ring atoms wherein the ring structure is a substituted or unsubstituted member selected from the group consisting of: a stable monocyclic ring which is aromatic ring ("aryl") having six ring atoms; a stable monocyclic non-aromatic ring having from 3 to 7 ring atoms in the ring; a stable bicyclic ring structure having a total of from 7 to 12 ring atoms in the two rings wherein the bicyclic ring structure is selected from the group consisting of ring structures in which both of the rings are aromatic, ring structures in which one of the rings is aromatic and ring structures in which both of the rings are non-aromatic; and a stable tricyclic ring structure having a total of from 10 to 16 atoms in the three rings wherein the tricyclic ring structure is selected from the group consisting of: ring structures in which three of the rings are aromatic, ring structures in which two of the rings are aromatic and ring structures in which three of the rings are non-aromatic. In each case, the non-aromatic rings when present in the monocyclic, bicyclic or tricyclic ring structure may independently be saturated, partially saturated or fully saturated. Examples of such carbocyclic ring structures include, but are not limited to, cyclopropyl, cyclobutyl, cyclopentyl, cyclohexyl, adamantyl, cyclooctyl, [3.3.0]bicyclooctane, [4.3.0]bicyclononane, [4.4.0]bicyclodecane (decalin),

2.2.2]bicyclooctane, fluorenyl, phenyl, naphthyl, indanyl, adamantyl, or tetrahydronaphthyl (tetralin). Moreover, the ring structures described herein may be attached to one or more indicated pendant groups via any carbon atom which results in a stable structure. The term "substituted" as used in conjunction with carbocyclic ring structures means that hydrogen atoms attached to the ring carbon atoms of ring structures described herein may be substituted by one or more of the substituents indicated for that structure if such substitution(s) would result in a stable compound.

The term "aryl" which is included with the term "carbocyclic ring structure" refers to an unsubstituted or substituted aromatic ring, substituted with one, two or three substituents selected from loweralkoxy, loweralkyl, loweralkylamino, hydroxy, halogen, cyano, hydroxyl, mercapto, nitro, thioalkoxy, carboxaldehyde, carboxyl, carboalkoxy and carboxamide, including but not limited to carbocyclic aryl, heterocyclic aryl, and biaryl groups and the like, all of which may be optionally substituted. Preferred aryl groups include phenyl, halophenyl, loweralkylphenyl, napthyl, biphenyl, phenanthrenyl and naphthacenyl.

The term "arylalkyl" which is included with the term "carbocyclic aryl" refers to one, two, or three aryl groups having the number of carbon atoms designated, appended to an alkyl group having the number of carbon atoms designated. Suitable arylalkyl groups include, but are not limited to, benzyl, picolyl, naphthylmethyl, phenethyl, benzyhydryl, trityl, and the like, all of which may be optionally substituted.

As used herein, the term "heterocyclic ring" or "heterocyclic ring system" is intended to mean a substituted or unsubstituted member selected from the group consisting of stable monocyclic ring having from 5-7 members in the ring itself and having from 1 to 4 hetero ring atoms selected from the group consisting of N, O and S; a stable bicyclic ring structure having a total of from 7 to 12 atoms in the two rings wherein at least one of the two rings has from 1 to 4 hetero atoms selected from N, O and S, including bicyclic ring structures wherein any of the described stable monocyclic heterocyclic rings is fused to a hexane or benzene ring; and a stable tricyclic heterocyclic ring structure having a total of from 10 to 16 atoms in the three rings wherein at least one of the three rings has from 1 to 4 hetero atoms selected from the group consisting of N, O and S. Any nitrogen and sulfur atoms present in a heterocyclic ring of such a heterocyclic ring structure may be oxidized. Unless indicated otherwise the terms "heterocyclic ring" or "heterocyclic ring

system" include aromatic rings, as well as non-aromatic rings which can be saturated, partially saturated or fully saturated non-aromatic rings. Also, unless indicated otherwise the term "heterocyclic ring system" includes ring structures wherein all of the rings contain at least one hetero atom as well as structures having less than all of the rings in the ring structure containing at least one hetero atom, for example bicyclic ring structures wherein one ring is a benzene ring and one of the rings has one or more hetero atoms are included within the term "heterocyclic ring systems" as well as bicyclic ring structures wherein each of the two rings has at least one hetero atom. Moreover, the ring structures described herein may be attached to one or more indicated pendant groups via any hetero atom or carbon atom which results in a stable structure. Further, the term "substituted" means that one or more of the hydrogen atoms on the ring carbon atom(s) or nitrogen atom(s) of the each of the rings in the ring structures described herein may be replaced by one or more of the indicated substituents if such replacement(s) would result in a stable compound. Nitrogen atoms in a ring structure may be quaternized, but such compounds are specifically indicated or are included within the term "a pharmaceutically acceptable salt" for a particular compound. When the total number of O and S atoms in a single heterocyclic ring is greater than 1, it is preferred that such atoms not be adjacent to one another. Preferably, there are no more that 1 O or S ring atoms in the same ring of a given heterocyclic ring structure.

Examples of monocylic and bicyclic heterocylic ring systems, in alphabetical order, are acridinyl, azocinyl, benzimidazolyl, benzofuranyl, benzothiofuranyl, benzothiophenyl, benzoxazolyl, benzthiazolyl, benztriazolyl, benztetrazolyl, benzisoxazolyl, benzisothiazolyl, benzimidazalinyl, carbazolyl, 4aH-carbazolyl, carbolinyl, chromanyl, chromenyl, cinnolinyl, decahydroquinolinyl, 2H,6H-1,5,2-dithiazinyl, dihydrofuro[2,3-b]tetrahydrofuran, furanyl, furazanyl, imidazolidinyl, imidazolinyl, imidazolyl, 1H-indazolyl, indolinyl, indolizinyl, indolyl, 3H-indolyl, isobenzofuranyl, isochromanyl, isoindazolyl, isoindolinyl, isoindolyl, isoquinolinyl (benzimidazolyl), isothiazolyl, isoxazolyl, morpholinyl, naphthyridinyl, octahydroisoquinolinyl, oxadiazolyl, 1,2,3-oxadiazolyl, 1,2,4-oxadiazolyl, 1,2,5-oxadiazolyl, 1,3,4-oxadiazolyl, oxazolidinyl, oxazolyl, oxazolidinyl, pyrimidinyl, phenanthridinyl, phenanthrolinyl, phenazinyl, phenothiazinyl, phenoxathiinyl, phenoxazinyl, phthalazinyl, piperazinyl, piperidinyl, pteridinyl, purinyl, pyranyl, pyrazinyl, pyroazolidinyl, pyrazolinyl, pyrazolyl, pyridazinyl, pryidooxazole, pyridoimidazole, pyridothiazole, pyridinyl, pyridyl, pyrimidinyl,

pyrrolidinyl, pyrrolinyl, 2H-pyrrolyl, pyrrolyl, quinazolinyl, quinolinyl, 4H-quinolizinyl, quinoxalinyl, quinuclidinyl, tetrahydrofuranyl, tetrahydroisoquinolinyl, tetrahydroquinolinyl, 6H-1,2,5-thiadazinyl, 1,2,3-thiadiazolyl, 1,2,4-thiadiazolyl, 1,2,5-thiadiazolyl, 1,3,4-thiadiazolyl, thianthrenyl, thiazolyl, thienyl, thienothiazolyl, thienooxazolyl, thienoimidazolyl, thiophenyl, triazinyl, 1,2,3-triazolyl, 1,2,4-triazolyl, 1,2,5-triazolyl, 1,3,4-triazolyl and xanthenyl. Preferred heterocyclic ring structures include, but are not limited to, pyridinyl, furanyl, thienyl, pyrrolyl, pyrazolyl, pyrrolidinyl, imidazolyl, indolyl, benzimidazolyl, 1H-indazolyl, oxazolinyl, or isatinoyl. Also included are fused ring and spiro compounds containing, for example, the above heterocylic ring structures.

As used herein the term "aromatic heterocyclic ring system" has essentially the same definition as for the monocyclic and bicyclic ring systems except that at least one ring of the ring system is an aromatic heterocyclic ring or the bicyclic ring has an aromatic or non-aromatic heterocyclic ring fused to an aromatic carbocyclic ring structure.

The terms "halo" or "halogen" as used herein refer to Cl, Br, F or I substituents. The term "haloalkyl", and the like, refer to an aliphatic carbon radicals having at least one hydrogen atom replaced by a Cl, Br, F or I atom, including mixtures of different halo atoms. Trihaloalkyl includes trifluoromethyl and the like as preferred radicals, for example.

The term "methylene" refers to $-CH_2-$.

The term "pharmaceutically acceptable salts" includes salts of compounds derived from the combination of a compound and an organic or inorganic acid. These compounds are useful in both free base and salt form. In practice, the use of the salt form amounts to use of the base form; both acid and base addition salts are within the scope of the present invention.

"Pharmaceutically acceptable acid addition salt" refers to salts retaining the biological effectiveness and properties of the free bases and which are not biologically or otherwise undesirable, formed with inorganic acids such as hydrochloric acid, hydrobromic acid, sulfuric acid, nitric acid, phosphoric acid and the like, and organic acids such as acetic acid, propionic acid, glycolic acid, pyruvic acid, oxalic acid, maleic acid, malonic acid, succinic acid, fumaric acid, tartaric acid,

citric acid, benzoic acid, cinnamic acid, mandelic acid, methanesulfonic acid, ethanesulfonic acid, p-toluenesulfonic acid, salicyclic acid and the like.

"Pharmaceutically acceptable base addition salts" include those derived from inorganic bases such as sodium, potassium, lithium, ammonium, calcium, magnesium, iron, zinc, copper, manganese, aluminum salts and the like. Particularly preferred are the ammonium, potassium, sodium, calcium and magnesium salts. Salts derived from pharmaceutically acceptable organic nontoxic bases include salts of primary, secondary, and tertiary amines, substituted amines including naturally occurring substituted amines, cyclic amines and basic ion exchange resins, such as isopropylamine, trimethylamine, diethylamine, triethylamine, tripropylamine, ethanolamine, 2-diethylaminoethanol, trimethamine, dicyclohexylamine, lysine, arginine, histidine, caffeine, procaine, hydrabamine, choline, betaine, ethylenediamine, glucosamine, methylglucamine, theobromine, purines, piperizine, piperidine, N-ethylpiperidine, polyamine resins and the like. Particularly preferred organic nontoxic bases are isopropylamine, diethylamine, ethanolamine, trimethamine, dicyclohexylamine, choline, and caffeine.

"Biological property" for the purposes herein means an *in vivo* effector or antigenic function or activity that is directly or indirectly performed by a compound of this invention that are often shown by *in vitro* assays. Effector functions include receptor or ligand binding, any enzyme activity or enzyme modulatory activity, any carrier binding activity, any hormonal activity, any activity in promoting or inhibiting adhesion of cells to an extracellular matrix or cell surface molecules, or any structural role. Antigenic functions include possession of an epitope or antigenic site that is capable of reacting with antibodies raised against it.

In the compounds of this invention, carbon atoms bonded to four non-identical substituents are asymmetric. Accordingly, the compounds may exist as diastereoisomers, enantiomers or mixtures thereof. The syntheses described herein may employ racemates, enantiomers or diastereomers as starting materials or intermediates. Diastereomeric products resulting from such syntheses may be separated by chromatographic or crystallization methods, or by other methods known in the art. Likewise, enantiomeric product mixtures may be separated using the same techniques or by other methods known in the art. Each of the asymmetric carbon atoms, when present in the compounds of this invention, may be in one of two configurations (R or S) and both are within the scope of the present invention.

Preferred Embodiments

In a preferred embodiment, the present invention provides a compound according to the formula I:

A-Y-D-E-G-J-Z-L

wherein:

A is selected from:

(a) C_1-C_6-alkyl;

(b) C_3-C_8-cycloalkyl;

(c) $-NR^2R^3$, $R^3C(=NR^2)-$, $R^2R^3N-C(=NR^2)-$, $R^2R^3N-C(=NR^2)-N(R^3)-$

(d) phenyl, which is independently substituted with 0-2 R^1 substituents;

(e) naphthyl, which is independently substituted with 0-2 R^1 substituents; and

(f) a monocyclic or fused bicyclic heterocyclic ring system having from 5 to 10 ring atoms, wherein 1-4 ring atoms of the ring system are selected from N, O and S, and wherein the ring system may be subsituted from 0-2 R^1 substituents;

R^1 is selected from:

halo, C_{1-4}alkyl, $R^2-C(=NR^3)-$, $R^2R^3N-C(=NR^2)-$, -CN, $(CH_2)_mNR^2R^3$, $SO_2NR^2R^3$, SO_2R^2, CF_3, OR^2, and a 5-6 membered aromatic heterocyclic system containing from 1-4 heteroatoms selected from N, O and S;

R^2 and R^3 are independently selected from the group consisting of:

H, C_{1-4}alkyl and C_{0-4}alkylaryl,

m is an integer of 0-2;

Y is a member selected from the group consisting of:

a direct link, -CH2-, $-C(=O)-$, $-N(R^4)-$, $-N(R^4)CH2-$, $-C=N(R^4)-$, $-C(=O)-N(R^4)-$, $-N(R^4)-C(=O)-$, $-SO_2-$, -O-, $-SO_2-N(R^4)-$ and $-N(R^4)-SO_2-$;

R^4 is selected from:

H, C_{1-4}alkyl and C_{0-4}alkylaryl;.

D is absent or is a member selected from the group consisting of:

(a) aryl, which is independently substituted with 0-2 R^{1a} substituents; and

(b) a monocyclic or fused bicyclic heterocyclic ring system having from 5 to 10 ring atoms, wherein 1-4 ring atoms of the ring system are selected from N, O and S, and wherein the ring system may be subsituted from 0-2 R^{1a} substituents;

R^{1a} is selected from:

Halo, C_{1-4}alkyl, -CN, $-NO_2$, $(CH_2)_mNR^{2a}R^{3a}$, $SO_2NR^{2a}R^{3a}$, SO_2R^{2a}, CF_3, OR^{2a}, and a 5-6 membered aromatic heterocyclic ring containing from 1-4 heteroatoms selected from N, O and S;

R^{2a} and R^{3a} are independently selected from the group consisting of:

H, C_{1-4}alkyl and C_{0-4}alkylaryl;

E is a member selected from the group consisting of:

$-N(R^5)-C(=O)-$, $-C(=O)-N(R^5)-$, $-N(R^5)-C(=O)-N(R^6)-$, $-SO_2-N(R^5)-$, $-N(R^5)-SO_2-N(R^6)-$ and $-N(R^5)-SO_2-N(R^6)-C(=O)-$;

R^5 and R^6 are independently selected from:

H, C_{1-4}alkyl, C_{0-4}alkylaryl, C_{0-4}alkylheteroaryl, C_{1-4}alkylCOOH and C_{1-4}alkylCOOC_{1-4}alkyl;

G is selected from:

$-CR^7R^8-$ and $-CR^{7a}R^{8a}-CR^{7b}R^{8b}-$

wherein R^7, R^8, R^{7a}, R^{8a}, R^{7b} and R^{8b} are independently a member selected from from the group consisting of:

hydrogen, C_{1-4}alkyl, C_{0-4}alkyl-C_{3-8}cycloalkyl, C_{0-4}alkylaryl, $-C_{0-4}$alkylCOOR^9, $-C_{0-4}$alkylC(=O)NR^9R^{10}, $-N(R^9)COR^{10}$, $-N(R^9)C(=O)R^{10}$, $-N(R^9)SO_2R^{10}$, and common amino acid side chains;

R^9 and R^{10} are independently selected from:

H, C_{1-4}alkyl and C_{0-4}alkylaryl;

J is a member selected from the group consisting of:

a direct link, -C(=O)-N(R^{11})-$(CH_2)_{0-2}$, -N(R^{11})-$(CH_2)_{0-2}$-C(=O)-, and -N(R^{11})-$(CH_2)_{0-2}$;

R^{11} is a member selected from the group consisting of:

hydrogen, C_{1-4}alkyl, C_{2-6}alkenyl, C_{2-6}alkynyl, C_{3-8}cycloalkyl, C_{0-4}alkylaryl, C_{0-4}alkylheterocyclics, CH_2COOC_{1-4}alkyl, CH_2COOC_{1-4}alkylaryl;

G and J together can form a cyclic ring systems.

Z is a member selected from the group consisting of:

(a) aryl, which is independently substituted with 0-2 R^{1b} substituents;and

(b) a monocyclic or fused bicyclic heterocyclic ring system having from 5 to 10 ring atoms, wherein 1-4 ring atoms of the ring system are selected from N, O and S, and wherein the ring system may be subsituted from 0-2 R^{1b} substituents;

R^{1b} is selected from:

halo, C_{1-4}alkyl, -CN, -NO_2, $NR^{2b}R^{3b}$, $SO_2NR^{2b}R^{3b}$, SO_2R^{2b}, CF_3, OR^{2b}, O-CH_2-CH_2-OR^{2b}, O-CH_2-$COOR^{2b}$, N(R^{2b})-CH_2-CH_2-OR^{2b}, N(-CH_2-CH_2-OR^{2b})$_2$, N(R^{2b})-C(=O)R^{3b}, N(R^{2b})-SO_2-R^{3b}, and a 5-6 membered aromatic heterocyclic ring containing from 1-4 heteroatoms selected from N, O and S;

R^{2b} and R^{3b} are independently selected from the group consisting of:

H, C_{1-4}alkyl and C_{0-4}alkylaryl;

L is selected from:

H, -CN, C(=O)$NR^{12}R^{13}$, $(CH_2)_nNR^{12}R^{13}$, C(=NR^{12})$NR^{12}R^{13}$, OR^{12}, -NR^{12}C(=NR^{12})$NR^{12}R^{13}$ and NR^{12}C(=NR^{12})-R^{13};

R^{12} and R^{13} are independently selected from:

hydrogen, -OR^{14}, -$NR^{14}R^{15}$, C_{1-4}alkyl, C_{0-4}alkylaryl $COOC_{1-4}$alkyl, and COO-C_{0-4}alkylaryl;

R^{14} and R^{15} are independently selected from:

H and C_{1-4}alkyl; and

and all pharmaceutically acceptable isomers, salts, hydrates, solvates and prodrug derivatives thereof.

In a further preferred embodiment, the present invention provides a compound according to the formula I:

A-Y-D-E-G-J-Z-L

wherein:

A is selected from:

(a) phenyl, which is independently substituted with 0-2 R^1 substituents;

(b) a monocyclic or fused bicyclic heterocyclic ring system having from 5 to 10 ring atoms, wherein 1-4 ring atoms of the ring system are selected from N, O and S, and wherein the ring system may be subsituted from 0-2 R^1 substituents; and

(c) $-NR^2R^3$, $R^3C(=NR^2)-$, $R^2R^3N-C(=NR^2)-$, $R^2R^3N-C(=NR^2)-N(R^3)-$

R^1 is selected from:

halo, $R^2-C(=NR^3)-$, $R^2R^3N-C(=NR^2)-$, $(CH_2)_mNR^2R^3$, $SO_2NR^2R^3$ and SO_2R^2;

R^2 and R^3 are independently selected from the group consisting of:

H and C_{1-4}alkyl;

Y is a member selected from the group consisting of:

a direct link, -CH2-, -C(=O)-, $-N(R^4)-$, $-N(R^4)$CH2-, and -C=N(R4)-,

D is a member selected from the group consisting of:

(a) phenyl, which is independently substituted with 0-2 R^{1a} substituents; and

(b) a monocyclic or fused bicyclic heterocyclic ring system having from 5 to 10 ring atoms, wherein 1-4 ring atoms of the ring system are

selected from N, O and S, and wherein the ring system may be subsituted from 0-2 R^{1a} substituents;

R^{1a} is selected from:

Halo and C_{1-4}alkyl;

R^{2a} and R^{3a} are independently selected from the group consisting of:

H, C_{1-4}alkyl, C_{0-4}alkylaryl;

E is a member selected from the group consisting of:

$-N(R^5)-C(=O)-$

R^5 is independently selected from:

H, C_{1-4}alkyl, C_{0-4}alkylaryl and C_{0-4}alkylheteroaryl;

G is selected from:

$-CR^7R^8-$ and $-CR^{7a}R^{8a}-CR^{7b}R^{8b}-$

wherein R^7, R^8, R^{7a}, R^{8a}, R^{7b} and R^{8b} are independently a member selected from from the group consisting of:

hydrogen, C_{1-4}alkyl, C_{0-4}alkyl-C_{3-8}cycloalkyl, C_{0-4}alkylaryl, $-C_{0-4}$alkylCOOR^9, $-C_{0-4}$alkylC(=O) NR^9R^{10}, $-C_{0-4}$alkylC(=O)NR^9-CH_2-CH_2-O-R^{10}, $-C_{0-4}$alkylC(=O)NR^9(-CH_2-CH_2-O-R^{10}-)$_2$, $-N(R^9)COR^{10}$, $-N(R^9)C(=O)R^{10}$, $-N(R^9)SO_2R^{10}$, and common amino acid side chains;

R^9 and R^{10} are independently selected from:

H and C_{1-4}alkyl, wherein the NR^9R^{10} group of R^7, R^8, R^{7a}, R^{8a}, R^{7b} and R^{8b} is optionally cyclized to form a 5-8 membered heterocyclic group;

J is a member selected from the group consisting of:

$-N(R^{11})-C(=O)-(CH_2)_{0-2}$, and $-N(R^{11})-(CH_2)_{0-2}$;

R^{11} is a member selected from the group consisting of:

hydrogen, C_{1-4}alkyl, C_{2-6}alkenyl, C_{0-4}alkylaryl and a C_{0-4}alkylheterocyclic ring;

G and J together can form a cyclic ring systems.

Z is a member selected from the group consisting of:

(a) phenyl, which is independently substituted with 0-2 R^{1b} substituents;

(b) an aromatic heterocyclic ring having from 5 to 10 ring atoms, wherein 1-4 ring atoms are selected from N, O and S, and wherein the ring may be subsituted independently by from 0-2 R^{1b} substituents; and

(c) a fused aromatic bicyclic heterocyclic ring system having from 5 to 10 ring atoms, wherein 1-4 ring atoms of the ring system are selected from N, O and S, wherein the bicyclic ring system may be subsituted from 0-2 R^{1b} substituents;

R^{1b} is selected from:

halo, C_{1-4}alkyl, OH, OBn, O-CH_2-CH_2-OH, O-CH_2-CH_2-OCH_3, O-CH_2-COOH, O-CH_2-C(=O)-O-CH_3, NH_2, NH-CH_2-CH_2-O-CH_3, NH-C(=O)-O-CH_3, and NH-SO_2-CH_3;

L is selected from:

H, C(=O)$NR^{12}R^{13}$, $(CH_2)_nNR^{12}R^{13}$ and C(=NR^{12})$NR^{12}R^{13}$;

R^{12} and R^{13} are independently selected from:

hydrogen and C_{1-4}alkyl;

and all pharmaceutically acceptable isomers, salts, hydrates, solvates and prodrug derivatives thereof.

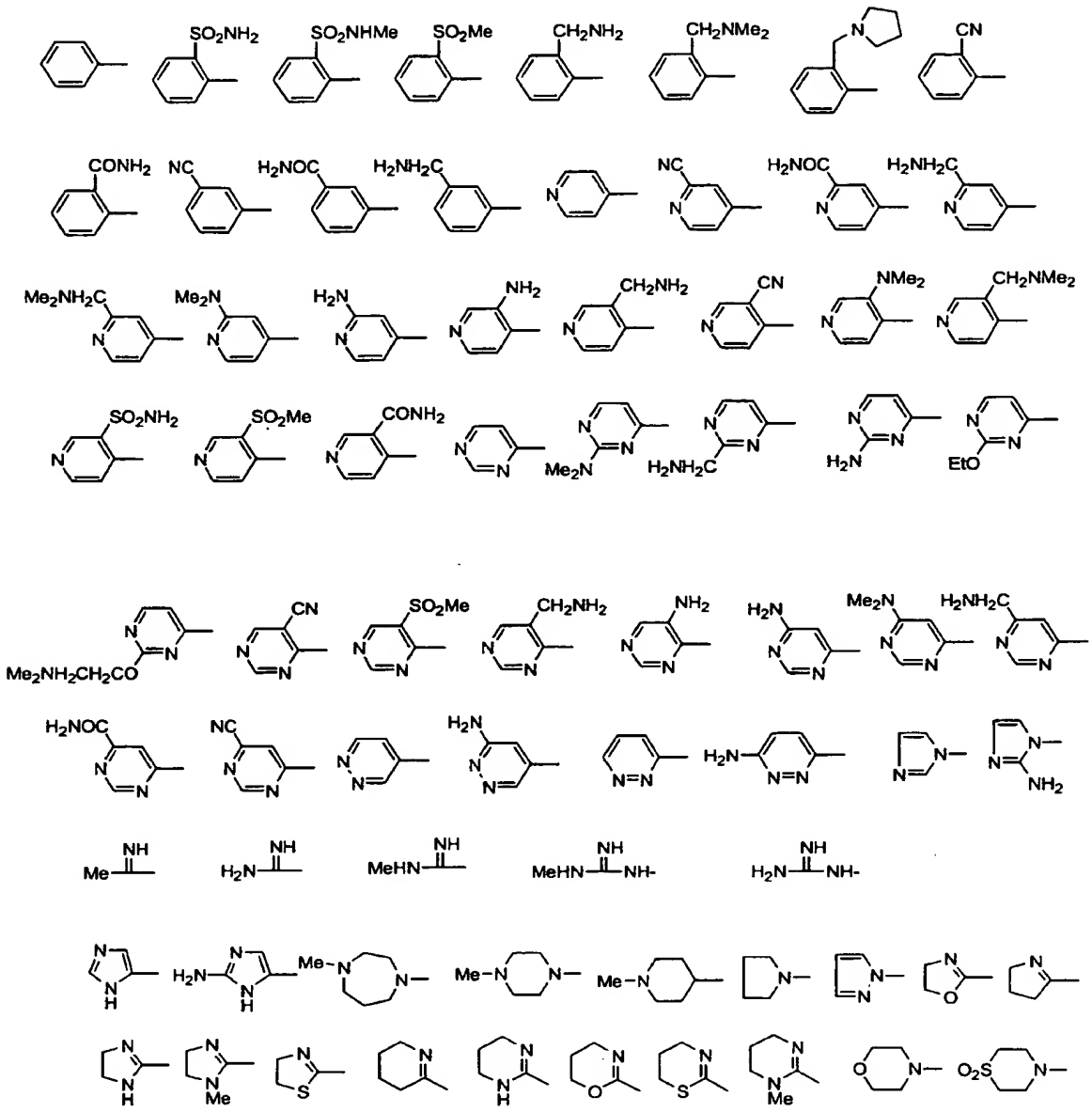
In a further preferred embodiment, the present invention provides a compound according to formula I:

where is Y?

A-D-E-G-J-Z-L

wherein

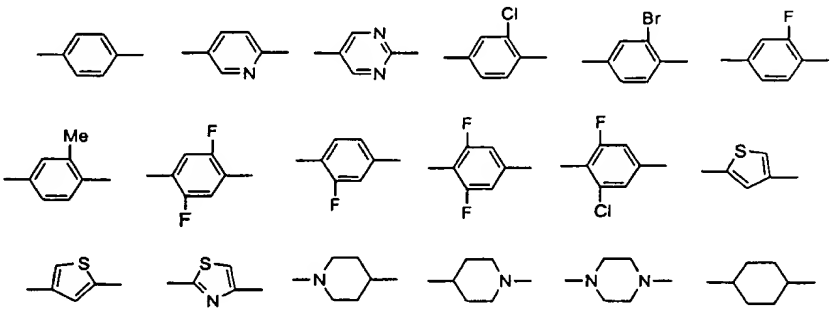
A is a member selected from the groups consisting of:



Y is selected from the group consisting of:

a direct link, -CO-, -SO2-, -N(Me)-, -N(Me)-CH2-, CH2-, C(=NH)-, and –C(=NMe)-

D is a direct link or a member selected from the group consisting of:



E is a member selected from the group consisting of::

-C(=O)-NH-, -NH-C(=O)-,

G is selected from:

-CH-(-NH_2)-CH_2-, -CH-(-NH(C(=O)-CH_3))-CH_2-, -CH-(-NH(C(=O)-Ph))-CH_2-, -CH-(C(=O)-OR^8)-, -CH(-R^7)-, -CH_2-CH(C(=O)-OR^8)-, and -CH_2-CH(C(=O)-N(-R^8, -R^8))-;

R^7 is a member selected from the group consisting of :

H, C-14alkyl, phenyl, Bn, and cycohexyl;

R^8 is a member selected from the group consisting of:

H, C_{1-6}alkyl, and C_{3-6}cycloalkyl;

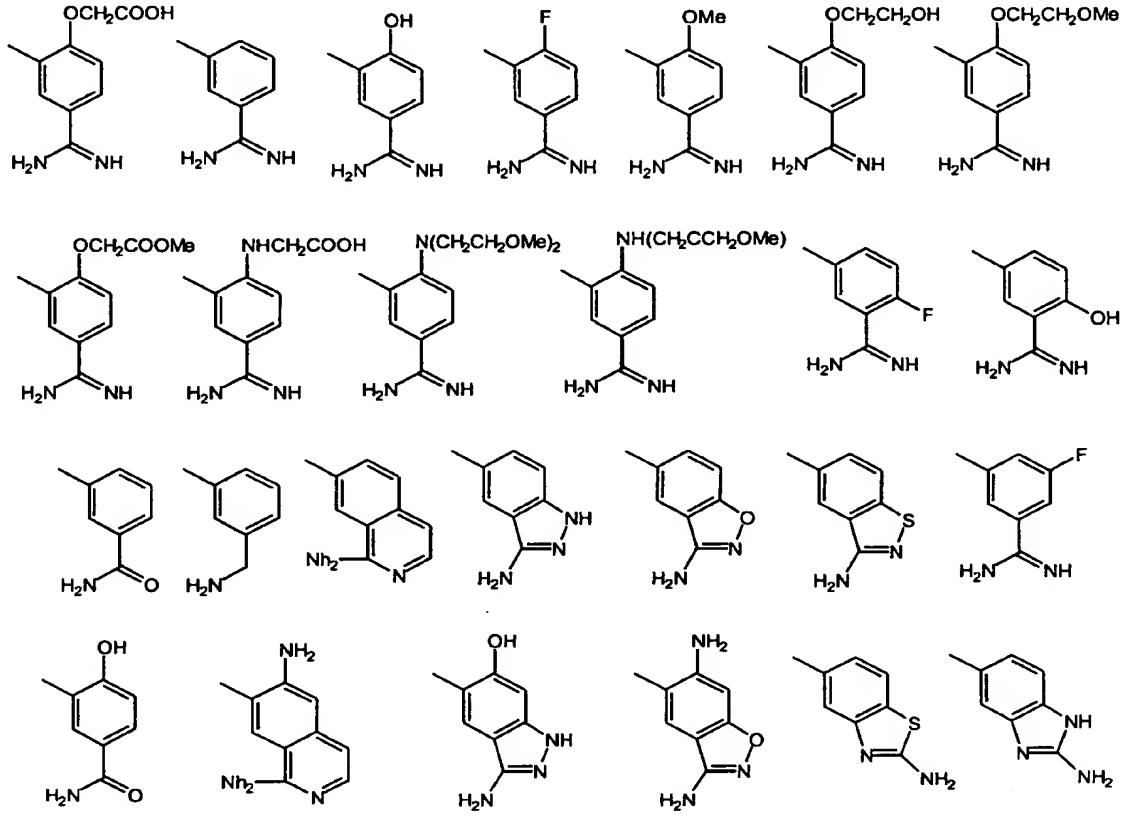
J is a member selected from the group consisting of;

-C(=O)-N(R^{11})-$(CH_2)_{0-2}$, -N(R^{11})-$(CH_2)_{0-2}$-C(=O)-, and -N(R^{11})-$(CH_2)_{0-2}$;

R^{11} is a member selected from the group consisting of:

H, methyl, phenyl and benzyl; and

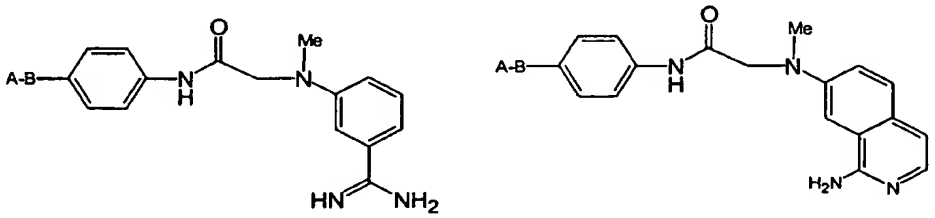
Z and L taken together are a member selected from the group consisting of:



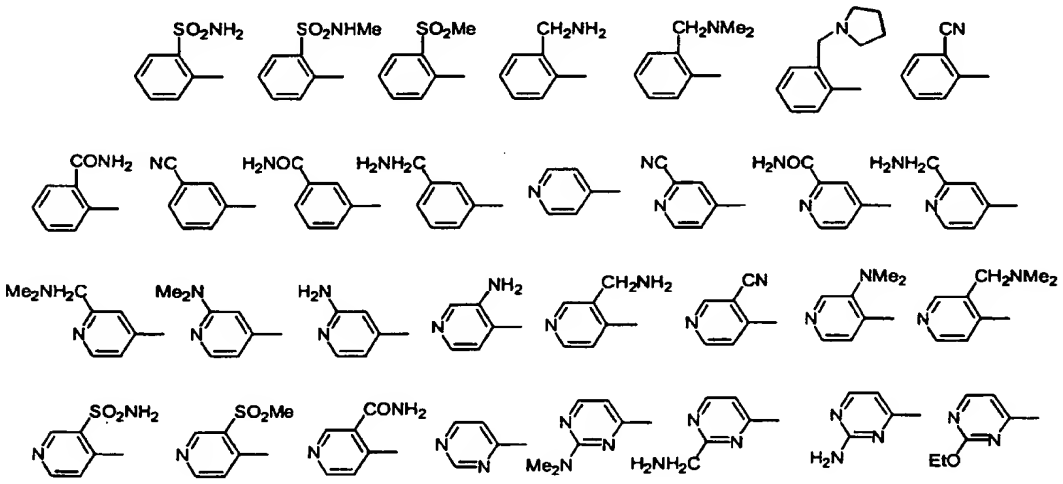
and all pharmaceutically acceptable isomers, salts, hydrates, solvates and prodrug derivatives thereof.

The following non-limiting tables illustrate representative compounds of the present invention:

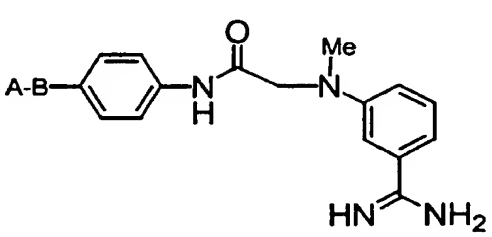
The compounds of the following formula:

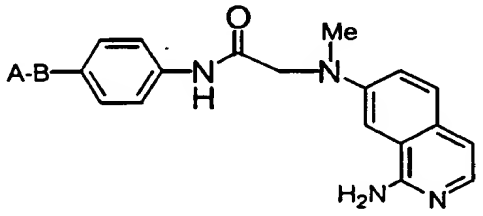


Where A-B is



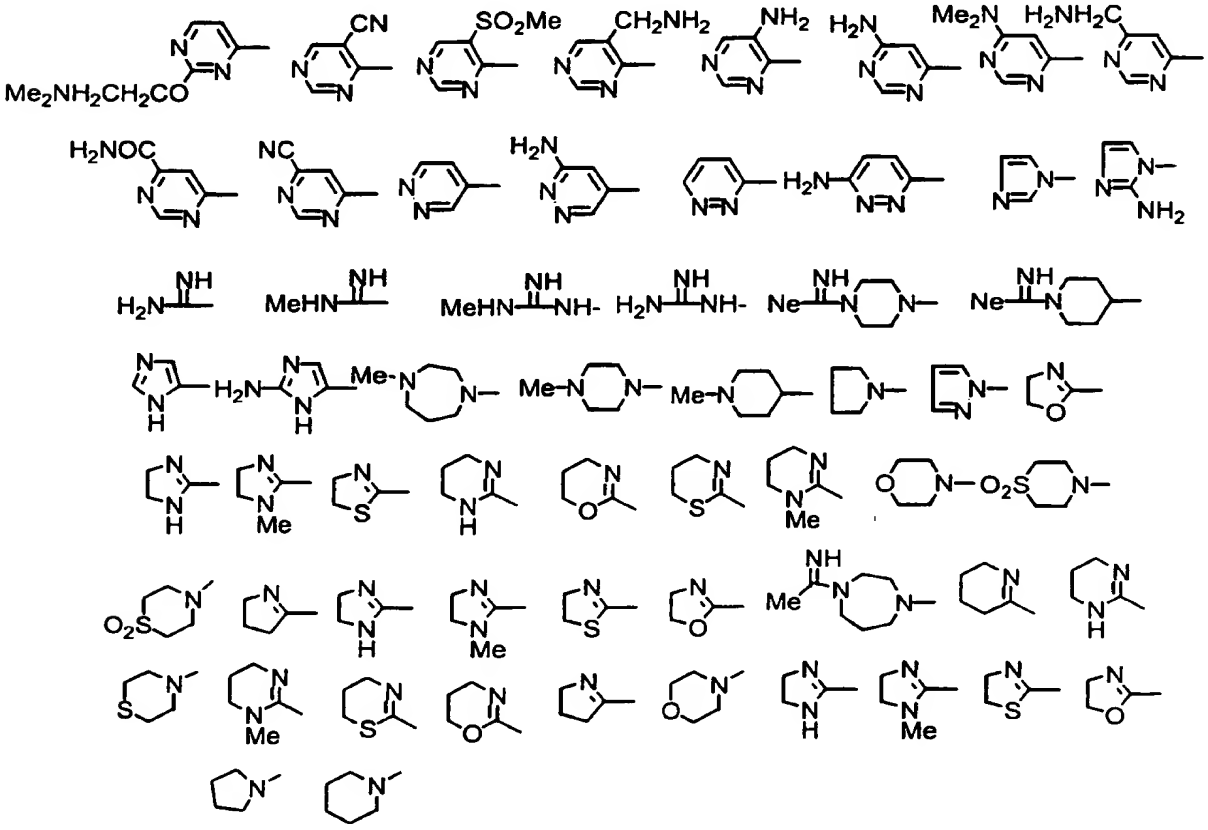
The compounds of the following formula:



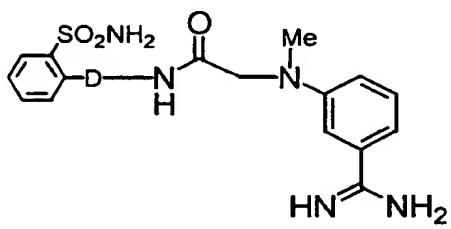


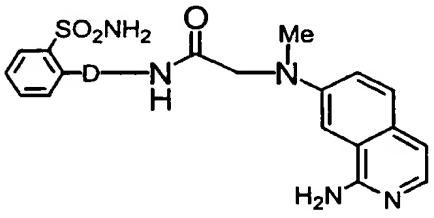
Where B is direct link, CH2, -CO-, -SO2, -C(=NH)-, -NH-, -N(Me)-, -N(Me)CH2-

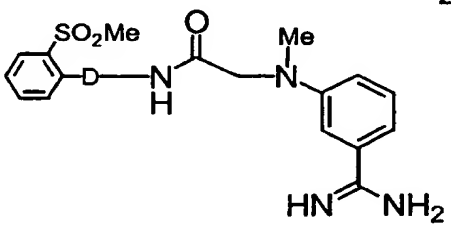
A is selected from:

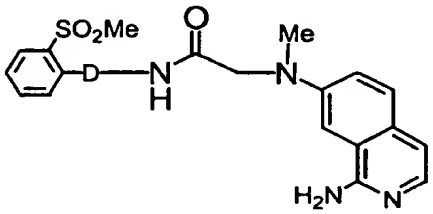


The compounds of the following formula:

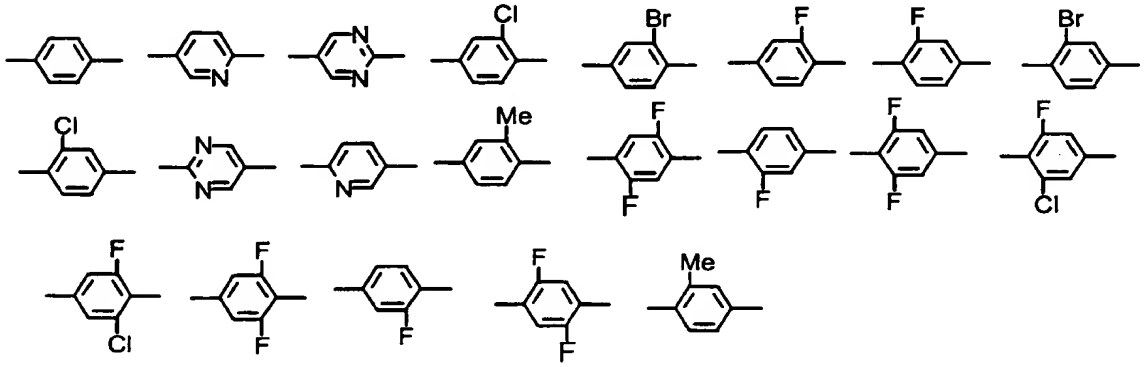




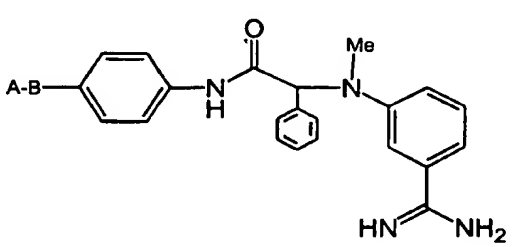


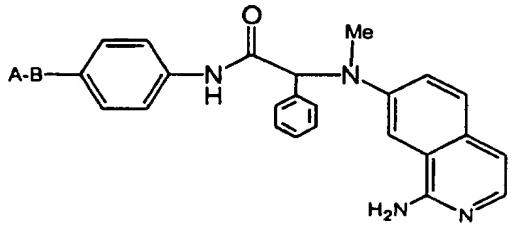


Where D is selected from

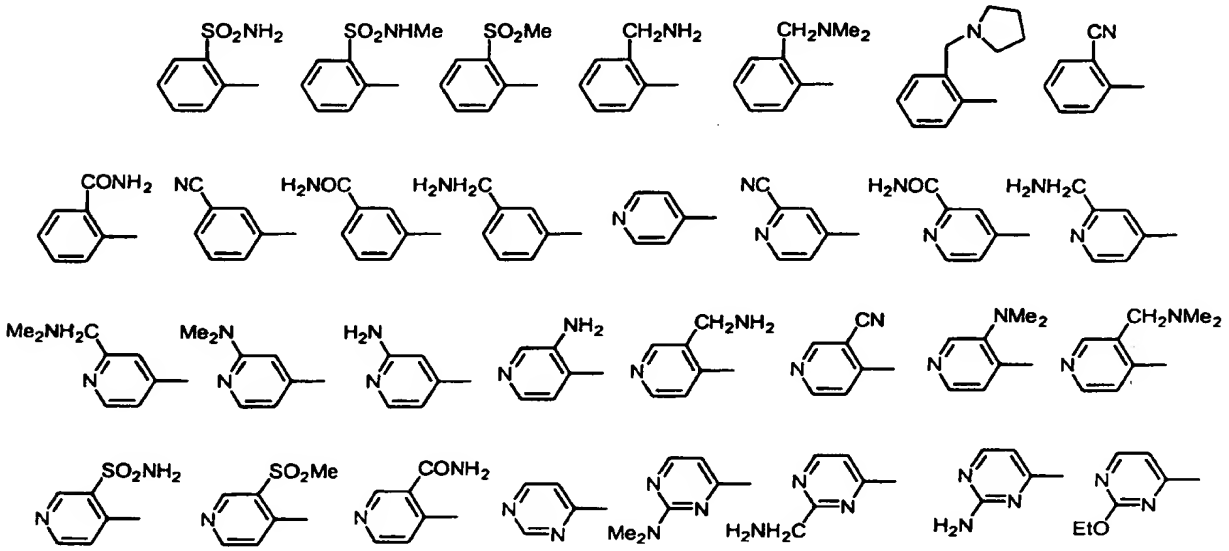


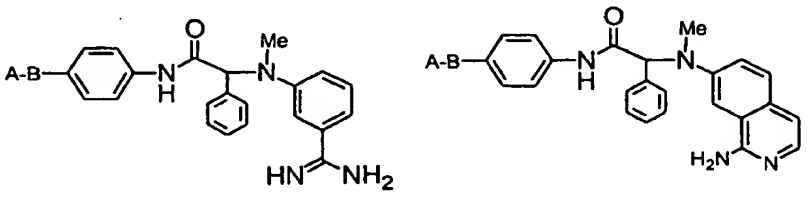
The compounds of the following formula:





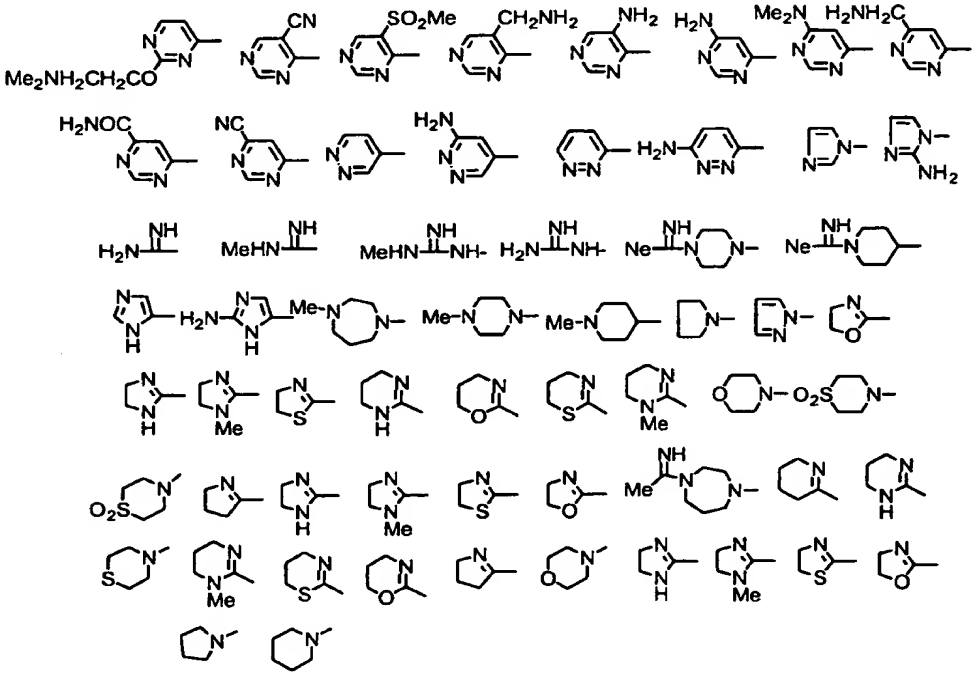
Where A-B is



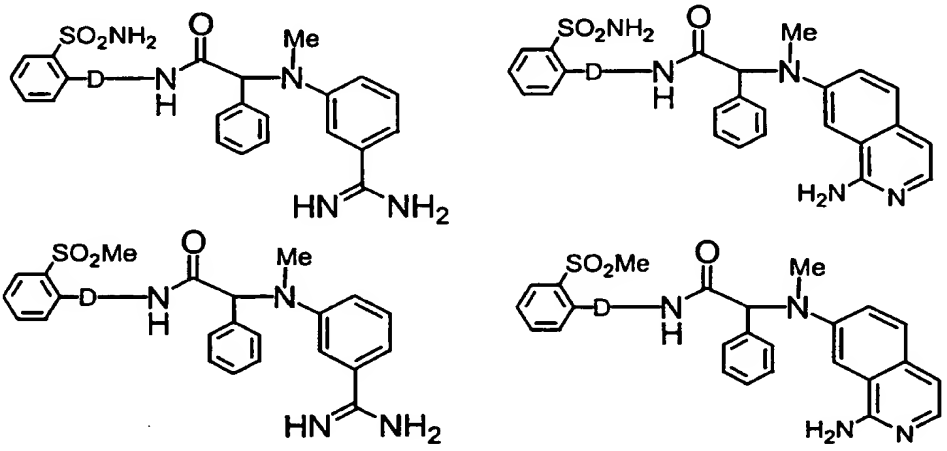


Where B is direct link, CH2, -CO-, -SO2, -C(=NH)-, -NH-, -N(Me)-, -N(Me)CH2-

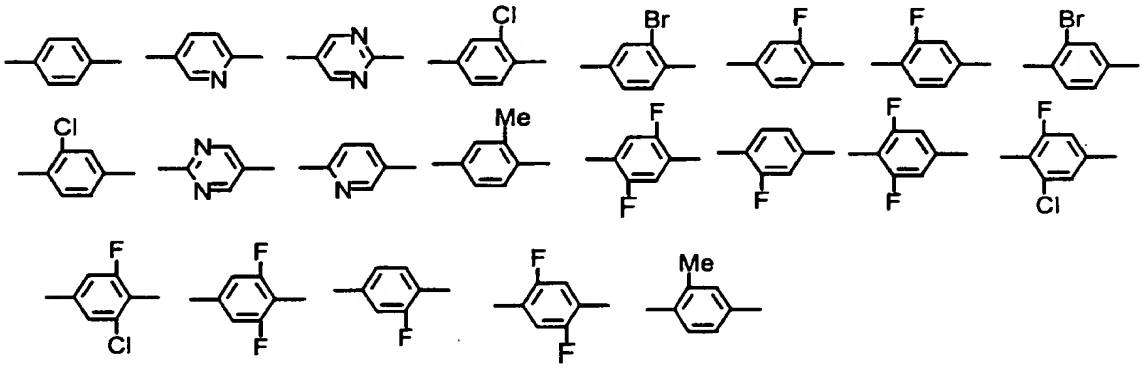
A is selected from:



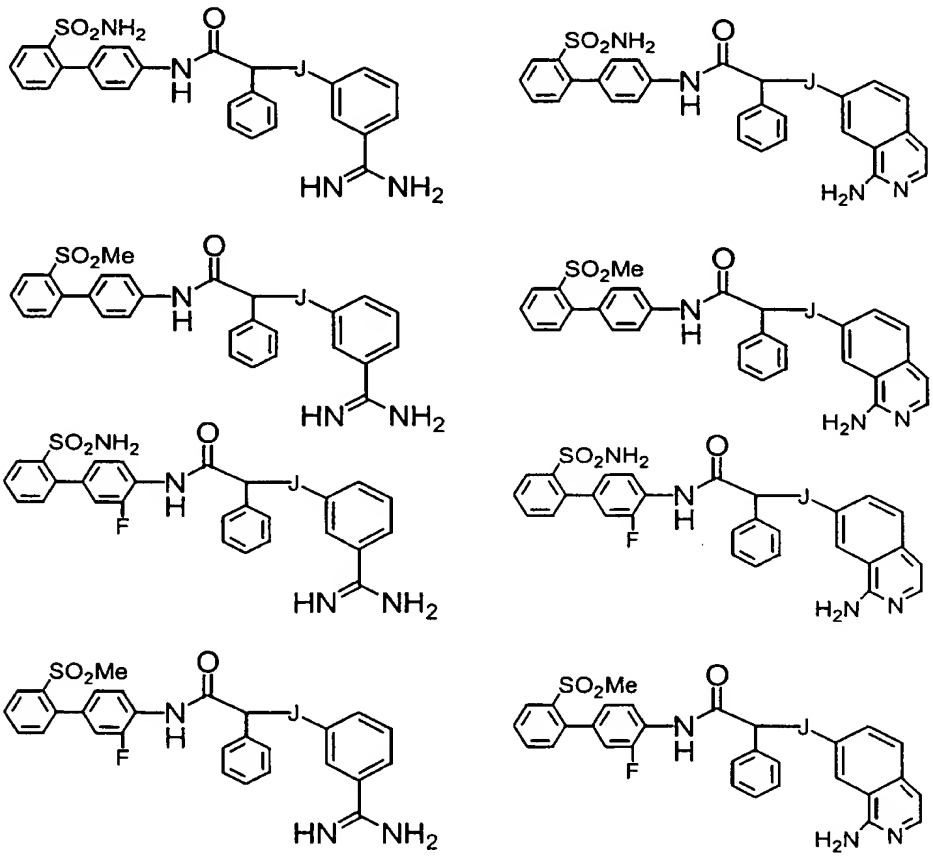
The compounds of the following formula where the preferred D is illustarted.



Where D is selected from



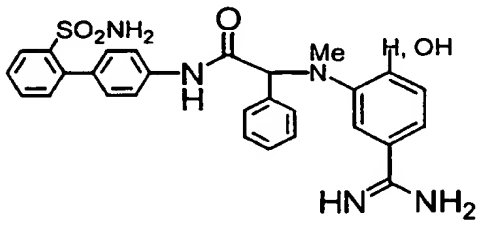
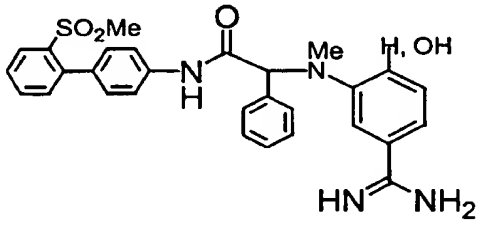
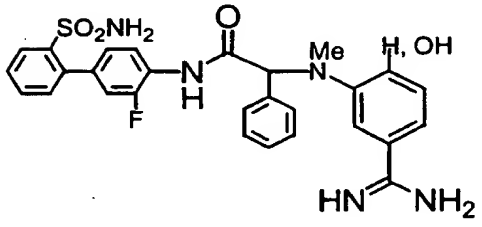
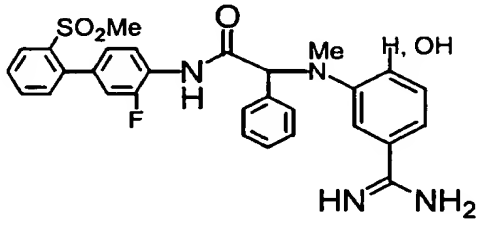
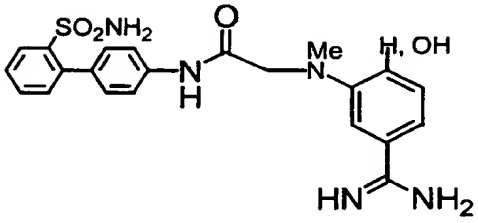
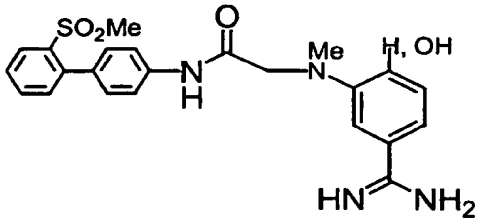
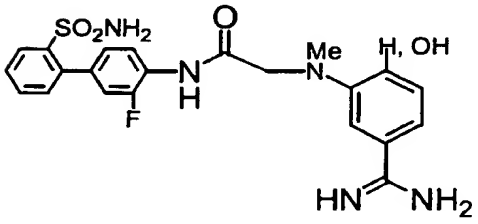
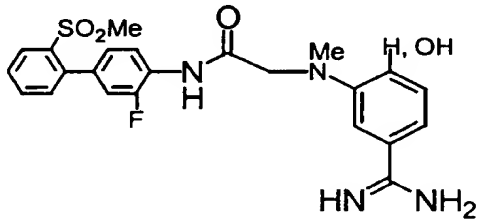
The compounds of the following formula where the preferred J is illustarted.



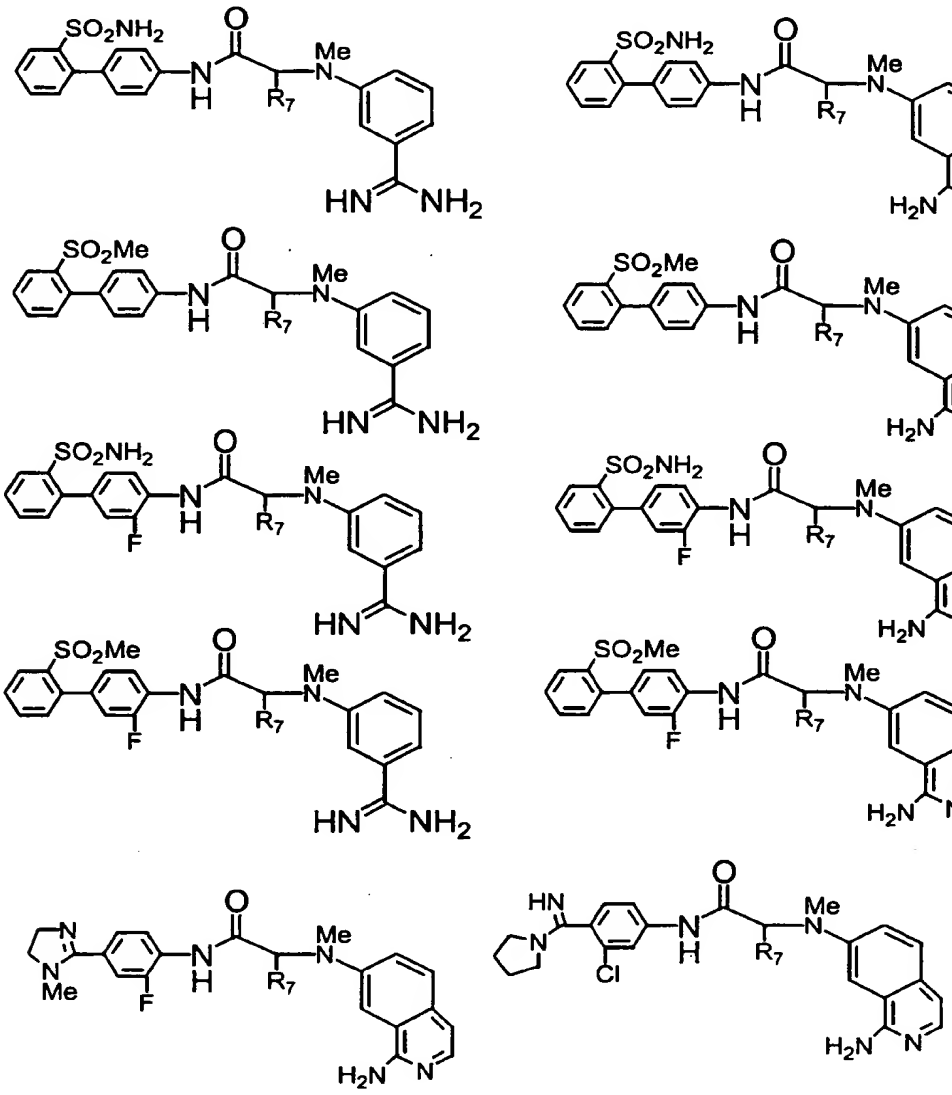
J is selected from:

NH, -N(Me)-, -N(Et)-, -N(Bn)-, -NHCO-, -NHCH2-, -N(Me)CH2-

The compounds of the following structures:

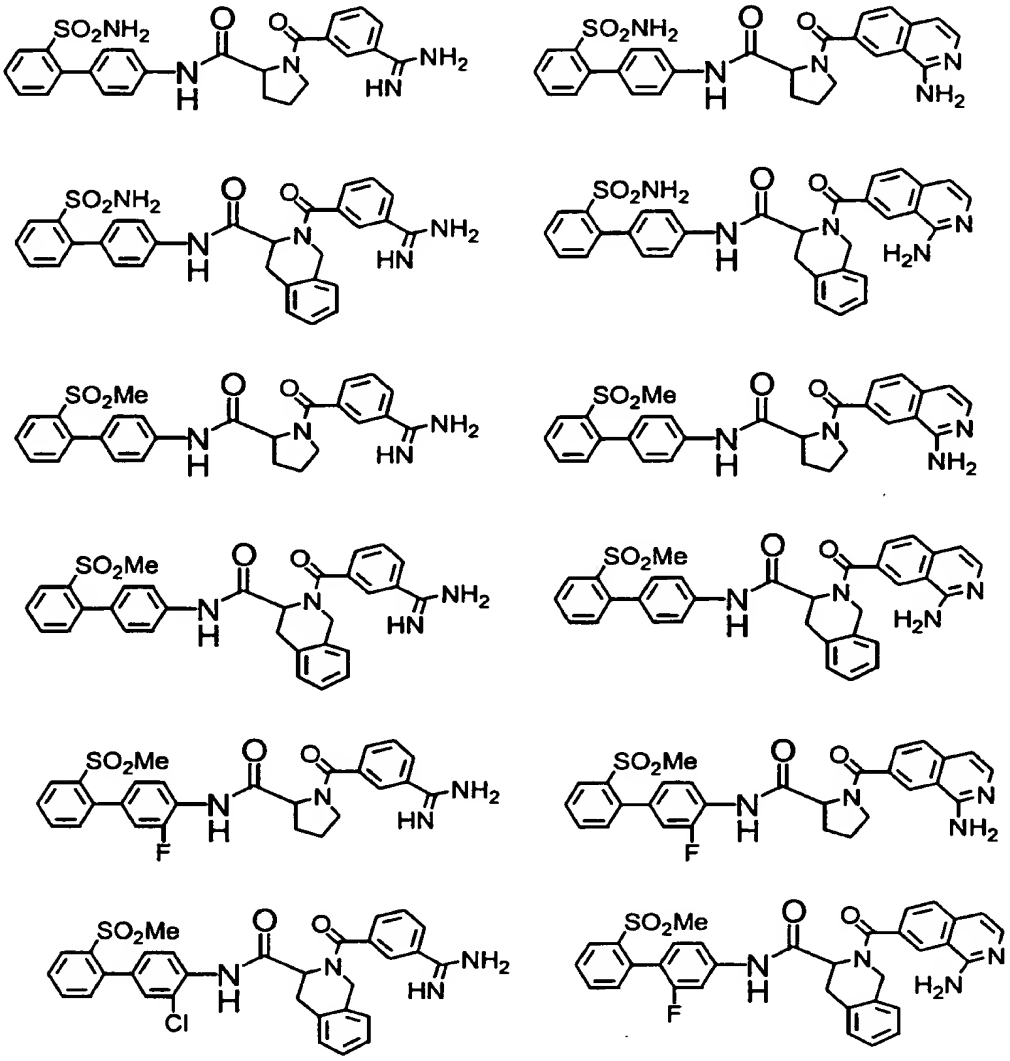









The compounds of the following structures:

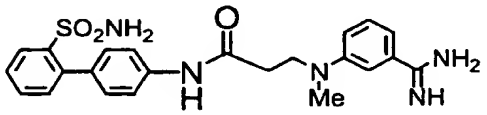
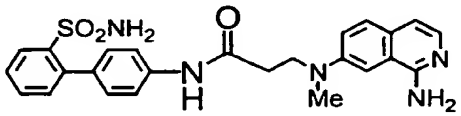
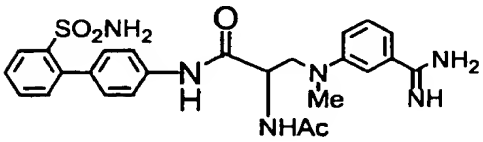
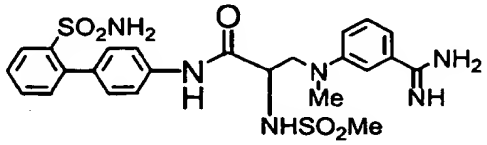


R7 is selected from: H, Me, Et, phenyl, benzyl

The compounds of the following formula where G and J form cyclic structures.



The compounds of the following formula:

This invention also encompasses all pharmaceutically acceptable isomers, salts, hydrates and solvates of the compounds of formulas I, II and III. In addition, the compounds of formulas I, II and III can exist in various isomeric and tautomeric forms, and all such forms are meant to be included in the invention, along with pharmaceutically acceptable salts, hydrates and solvates of such isomers and tautomers.

The compounds of this invention may be isolated as the free acid or base or converted to salts of various inorganic and organic acids and bases. Such salts are within the scope of this invention. Non-toxic and physiologically compatible salts are particularly useful although other less desirable salts may have use in the processes of isolation and purification.

A number of methods are useful for the preparation of the salts described above and are known to those skilled in the art. For example, the free acid or free base form of a compound of one of the formulas above can be reacted with one or more molar equivalents of the desired acid or base in a solvent or solvent mixture in which the salt is insoluble, or in a solvent like water after which the solvent is removed by evaporation, distillation or freeze drying. Alternatively, the free acid or base form of the product may be passed over an ion exchange resin to form the desired salt or one salt form of the product may be converted to another using the same general process.

Prodrug Derivatives of Compounds

This invention also encompasses prodrug derivatives of the compounds

contained herein. The term "prodrug" refers to a pharmacologically inactive derivative of a parent drug molecule that requires biotransformation, either spontaneous or enzymatic, within the organism to release the active drug. Prodrugs are variations or derivatives of the compounds of this invention which have groups cleavable under metabolic conditions. Prodrugs become the compounds of the invention which are pharmaceutically active *in vivo*, when they undergo solvolysis under physiological conditions or undergo enzymatic degradation. Prodrug compounds of this invention may be called single, double, triple etc., depending on the number of biotransformation steps required to release the active drug within the organism, and indicating the number of functionalities present in a precursor-type form. Prodrug forms often offer advantages of solubility, tissue compatibility, or delayed release in the mammalian organism (see, Bundgard, Design of Prodrugs, pp. 7-9, 21-24, Elsevier, Amsterdam 1985 and Silverman, The Organic Chemistry of Drug Design and Drug Action, pp. 352-401, Academic Press, San Diego, CA, 1992). Prodrugs commonly known in the art include acid derivatives well known to practitioners of the art, such as, for example, esters prepared by reaction of the parent acids with a suitable alcohol, or amides prepared by reaction of the parent acid compound with an amine, or basic groups reacted to form an acylated base derivative. Moreover, the prodrug derivatives of this invention may be combined with other features herein taught to enhance bioavailability.

As mentioned above, the compounds of this invention find utility as therapeutic agents for disease states in mammals which have disorders of coagulation such as in the treatment or prevention of unstable angina, refractory angina, myocardial infarction, transient ischemic attacks, thrombotic stroke, embolic stroke, disseminated intravascular coagulation including the treatment of septic shock, deep venous thrombosis in the prevention of pulmonary embolism or the treatment of reocclusion or restenosis of reperfused coronary arteries. Further, these compounds are useful for the treatment or prophylaxis of those diseases which involve the production and/or action of factor Xa/prothrombinase complex. This includes a number of thrombotic and prothrombotic states in which the coagulation cascade is activated which include but are not limited to, deep venous thrombosis, pulmonary embolism, myocardial infarction, stroke, thromboembolic complications of surgery and peripheral arterial occlusion.

Accordingly, a method for preventing or treating a condition in a mammal

characterized by undesired thrombosis comprises administering to the mammal a therapeutically effective amount of a compound of this invention. In addition to the disease states noted above, other diseases treatable or preventable by the administration of compounds of this invention include, without limitation, occlusive coronary thrombus formation resulting from either thrombolytic therapy or percutaneous transluminal coronary angioplasty, thrombus formation in the venous vasculature, disseminated intravascular coagulopathy, a condition wherein there is rapid consumption of coagulation factors and systemic coagulation which results in the formation of life-threatening thrombi occurring throughout the microvasculature leading to widespread organ failure, hemorrhagic stroke, renal dialysis, blood oxygenation, and cardiac catheterization.

The compounds of the invention also find utility in a method for inhibiting the coagulation biological samples, which comprises the administration of a compound of the invention.

The compounds of the present invention may also be used in combination with other therapeutic or diagnostic agents. In certain preferred embodiments, the compounds of this invention may be coadministered along with other compounds typically prescribed for these conditions according to generally accepted medical practice such as anticoagulant agents, thrombolytic agents, or other antithrombotics, including platelet aggregation inhibitors, tissue plasminogen activators, urokinase, prourokinase, streptokinase, heparin, aspirin, or warfarin. The compounds of the present invention may act in a synergistic fashion to prevent reocclusion following a successful thrombolytic therapy and/or reduce the time to reperfusion. These compounds may also allow for reduced doses of the thrombolytic agents to be used and therefore minimize potential hemorrhagic side-effects. The compounds of this invention can be utilized *in vivo*, ordinarily in mammals such as primates, (e.g. humans), sheep, horses, cattle, pigs, dogs, cats, rats and mice, or *in vitro*.

The biological properties of the compounds of the present invention can be readily characterized by methods that are well known in the art, for example by the *in vitro* protease activity assays and *in vivo* studies to evaluate antithrombotic efficacy, and effects on hemostasis and hematological parameters, such as are illustrated in the examples.

Diagnostic applications of the compounds of this invention will typically

utilize formulations in the form of solutions or suspensions. In the management of thrombotic disorders the compounds of this invention may be utilized in compositions such as tablets, capsules or elixirs for oral administration, suppositories, sterile solutions or suspensions or injectable administration, and the like, or incorporated into shaped articles. Subjects in need of treatment (typically mammalian) using the compounds of this invention can be administered dosages that will provide optimal efficacy. The dose and method of administration will vary from subject to subject and be dependent upon such factors as the type of mammal being treated, its sex, weight, diet, concurrent medication, overall clinical condition, the particular compounds employed, the specific use for which these compounds are employed, and other factors which those skilled in the medical arts will recognize.

Formulations of the compounds of this invention are prepared for storage or administration by mixing the compound having a desired degree of purity with physiologically acceptable carriers, excipients, stabilizers etc., and may be provided in sustained release or timed release formulations. Acceptable carriers or diluents for therapeutic use are well known in the pharmaceutical field, and are described, for example, in Remington's Pharmaceutical Sciences, Mack Publishing Co., (A.R. Gennaro edit. 1985). Such materials are nontoxic to the recipients at the dosages and concentrations employed, and include buffers such as phosphate, citrate, acetate and other organic acid salts, antioxidants such as ascorbic acid, low molecular weight (less than about ten residues) peptides such as polyarginine, proteins, such as serum albumin, gelatin, or immunoglobulins, hydrophilic polymers such as polyvinylpyrrolidinone, amino acids such as glycine, glutamic acid, aspartic acid, or arginine, monosaccharides, disaccharides, and other carbohydrates including cellulose or its derivatives, glucose, mannose or dextrins, chelating agents such as EDTA, sugar alcohols such as mannitol or sorbitol, counterions such as sodium and/or nonionic surfactants such as Tween, Pluronics or polyethyleneglycol.

Dosage formulations of the compounds of this invention to be used for therapeutic administration must be sterile. Sterility is readily accomplished by filtration through sterile membranes such as 0.2 micron membranes, or by other conventional methods. Formulations typically will be stored in lyophilized form or as an aqueous solution. The pH of the preparations of this invention typically will be 3-11, more preferably 5-9 and most preferably 7-8. It will be understood that use of certain of the foregoing excipients, carriers, or stabilizers will result in the

formation of cyclic polypeptide salts. While the preferred route of administration is by injection, other methods of administration are also anticipated such as orally, intravenously (bolus and/or infusion), subcutaneously, intramuscularly, colonically, rectally, nasally, transdermally or intraperitoneally, employing a variety of dosage forms such as suppositories, implanted pellets or small cylinders, aerosols, oral dosage formulations and topical formulations such as ointments, drops and dermal patches. The compounds of this invention are desirably incorporated into shaped articles such as implants which may employ inert materials such as biodegradable polymers or synthetic silicones, for example, Silastic, silicone rubber or other polymers commercially available.

The compounds of the invention may also be administered in the form of liposome delivery systems, such as small unilamellar vesicles, large unilamellar vesicles and multilamellar vesicles. Liposomes can be formed from a variety of lipids, such as cholesterol, stearylamine or phosphatidylcholines.

The compounds of this invention may also be delivered by the use of antibodies, antibody fragments, growth factors, hormones, or other targeting moieties, to which the compound molecules are coupled. The compounds of this invention may also be coupled with suitable polymers as targetable drug carriers. Such polymers can include polyvinylpyrrolidinone, pyran copolymer, polyhydroxy-propyl-methacrylamide-phenol, polyhydroxyethyl-aspartamide-phenol, or polyethyleneoxide-polylysine substituted with palmitoyl residues. Furthermore, compounds of the invention may be coupled to a class of biodegradable polymers useful in achieving controlled release of a drug, for example polylactic acid, polyglycolic acid, copolymers of polylactic and polyglycolic acid, polyepsilon caprolactone, polyhydroxy butyric acid, polyorthoesters, polyacetals, polydihydropyrans, polycyanoacrylates and cross linked or amphipathic block copolymers of hydrogels. Polymers and semipermeable polymer matrices may be formed into shaped articles, such as valves, stents, tubing, prostheses and the like.

Therapeutic compound liquid formulations generally are placed into a container having a sterile access port, for example, an intravenous solution bag or vial having a stopper pierceable by hypodermic injection needle.

Therapeutically effective dosages may be determined by either *in vitro* or *in vivo* methods. For each particular compound of the present invention, individual

determinations may be made to determine the optimal dosage required. The range of therapeutically effective dosages will be influenced by the route of administration, the therapeutic objectives and the condition of the patient. For injection by hypodermic needle, it may be assumed the dosage is delivered into the body's fluids. For other routes of administration, the absorption efficiency must be individually determined for each compound by methods well known in pharmacology. Accordingly, it may be necessary for the therapist to titer the dosage and modify the route of administration as required to obtain the optimal therapeutic effect. The determination of effective dosage levels, that is, the dosage levels necessary to achieve the desired result, will be readily determined by one skilled in the art. Typically, applications of compound are commenced at lower dosage levels, with dosage levels being increased until the desired effect is achieved.

The compounds of the invention can be administered orally or parenterally in an effective amount within the dosage range of about 0.1 to 100 mg/kg, preferably about 0.5 to 50 mg/kg and more preferably about 1 to 20 mg/kg on a regimen in a single or 2 to 4 divided daily doses and/or continuous infusion.

Typically, about 5 to 500 mg of a compound or mixture of compounds of this invention, as the free acid or base form or as a pharmaceutically acceptable salt, is compounded with a physiologically acceptable vehicle, carrier, excipient, binder, preservative, stabilizer, dye, flavor etc., as called for by accepted pharmaceutical practice. The amount of active ingredient in these compositions is such that a suitable dosage in the range indicated is obtained.

Typical adjuvants which may be incorporated into tablets, capsules and the like are binders such as acacia, corn starch or gelatin, and excipients such as microcrystalline cellulose, disintegrating agents like corn starch or alginic acid, lubricants such as magnesium stearate, sweetening agents such as sucrose or lactose, or flavoring agents. When a dosage form is a capsule, in addition to the above materials it may also contain liquid carriers such as water, saline, or a fatty oil. Other materials of various types may be used as coatings or as modifiers of the physical form of the dosage unit. Sterile compositions for injection can be formulated according to conventional pharmaceutical practice. For example, dissolution or suspension of the active compound in a vehicle such as an oil or a synthetic fatty vehicle like ethyl oleate, or into a liposome may be desired. Buffers,

preservatives, antioxidants and the like can be incorporated according to accepted pharmaceutical practice.

Preparation of Compounds

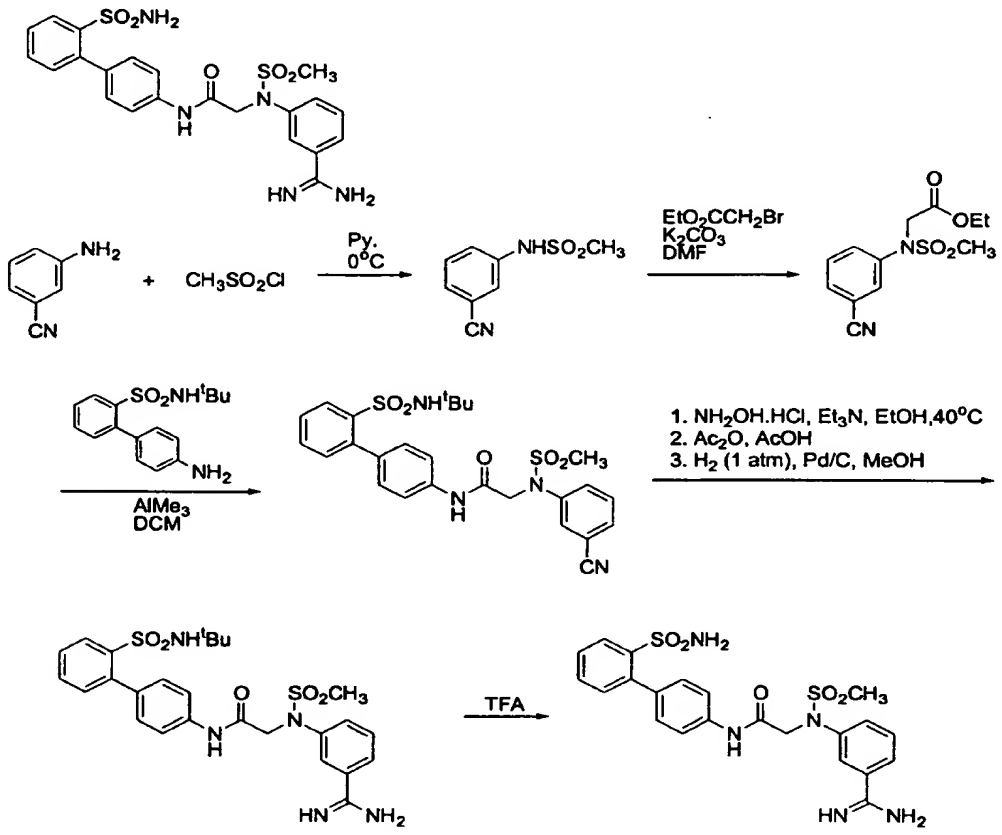
The compounds of the present invention may be synthesized by either solid or liquid phase methods described and referenced in standard textbooks, or by a combination of both methods. These methods are well known in the art. See, Bodanszky, "The Principles of Peptide Synthesis", Hafner, *et al.*, Eds., Springer-Verlag, Berlin, 1984.

Starting materials used in any of these methods are commercially available from chemical vendors such as Aldrich, Sigma, Nova Biochemicals, Bachem Biosciences, and the like, or may be readily synthesized by known procedures.

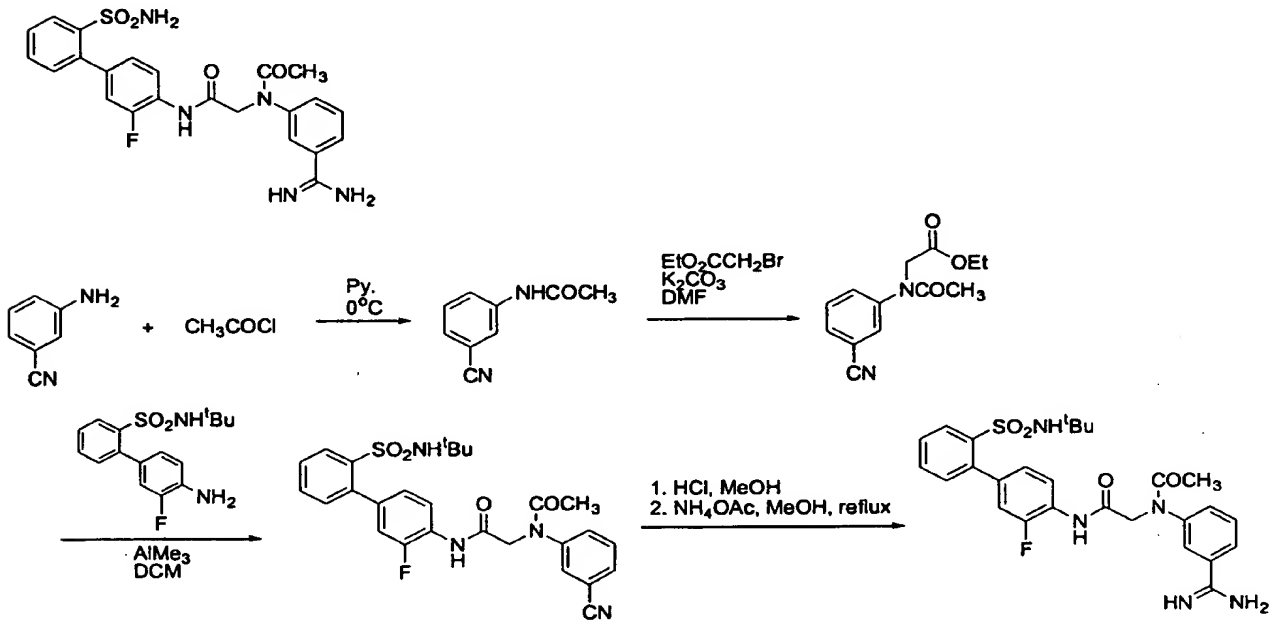
Reactions are carried out in standard laboratory glassware and reaction vessels under reaction conditions of standard temperature and pressure, except where otherwise indicated.

During the synthesis of these compounds, the functional groups of the amino acid derivatives used in these methods are protected by blocking groups to prevent cross reaction during the coupling procedure. Examples of suitable blocking groups and their use are described in "The Peptides: Analysis, Synthesis, Biology", Academic Press, Vol. 3 (Gross, *et al.*, Eds., 1981) and Vol. 9 (1987), the disclosures of which are incorporated herein by reference.

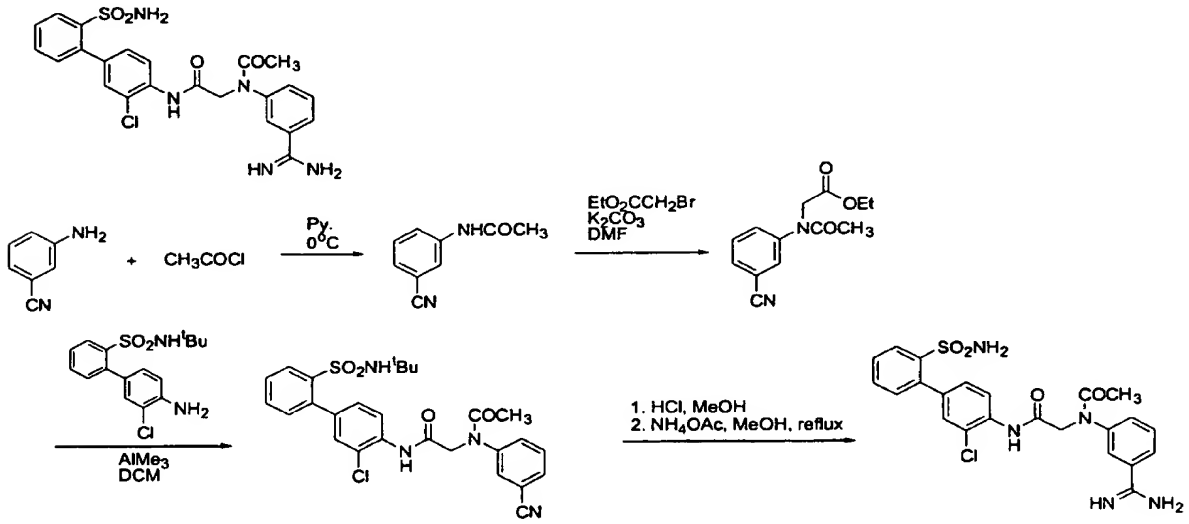
Scheme 1


SO2NH2
O
SO2CH3
N
H
HN
NH2
NH2
CN
+
CH3SO2Cl
Py.
0°C
NHSO2CH3
CN
EtO2CCH2Br
K2CO3
DMF
O
OEt
NSO2CH3
CN
SO2NHtBu
NH2
AlMe3
DCM
1. NH2OH.HCl, Et3N, EtOH,40°C
2. Ac2O, AcOH
3. H2 (1 atm), Pd/C, MeOH
TFA

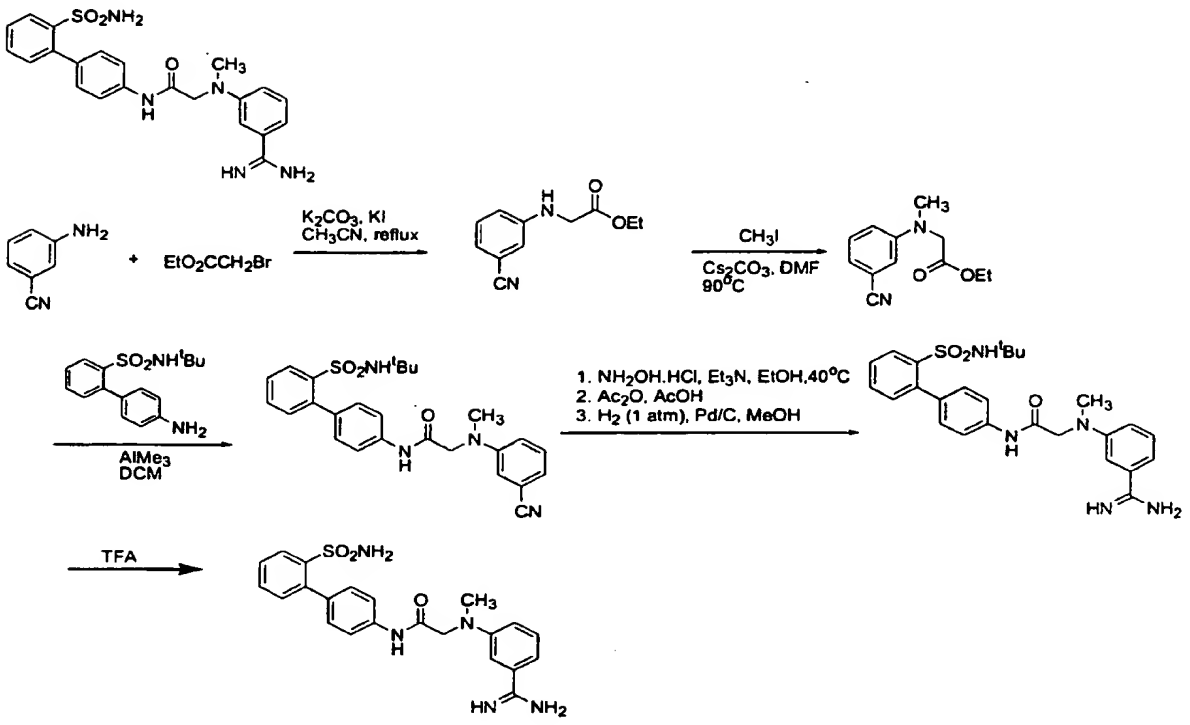
Scheme 2



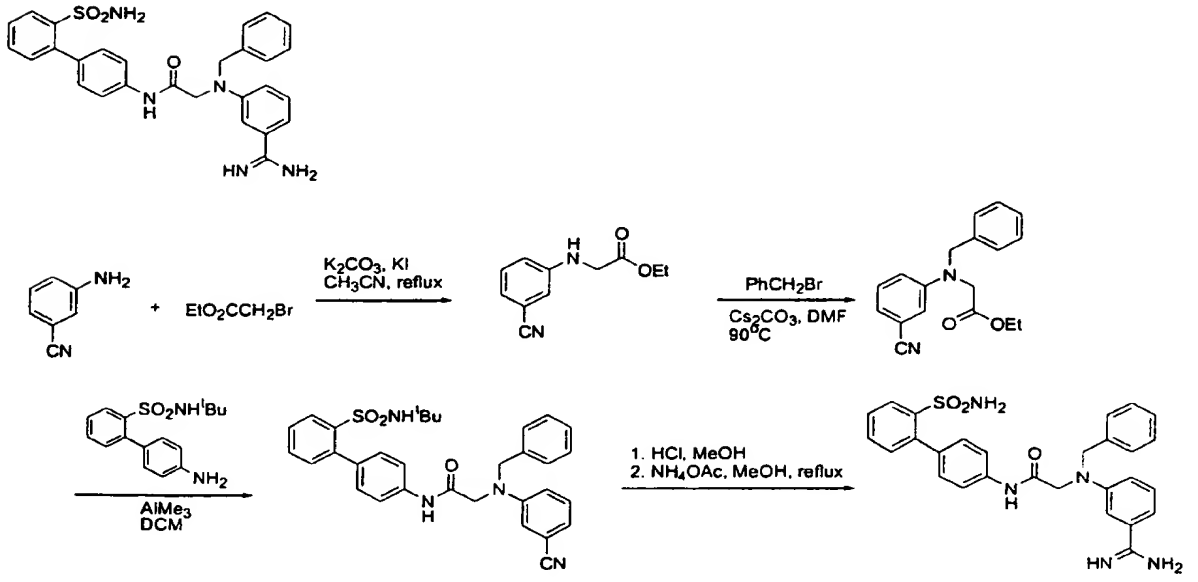
Scheme 3



Scheme 4



Scheme 5



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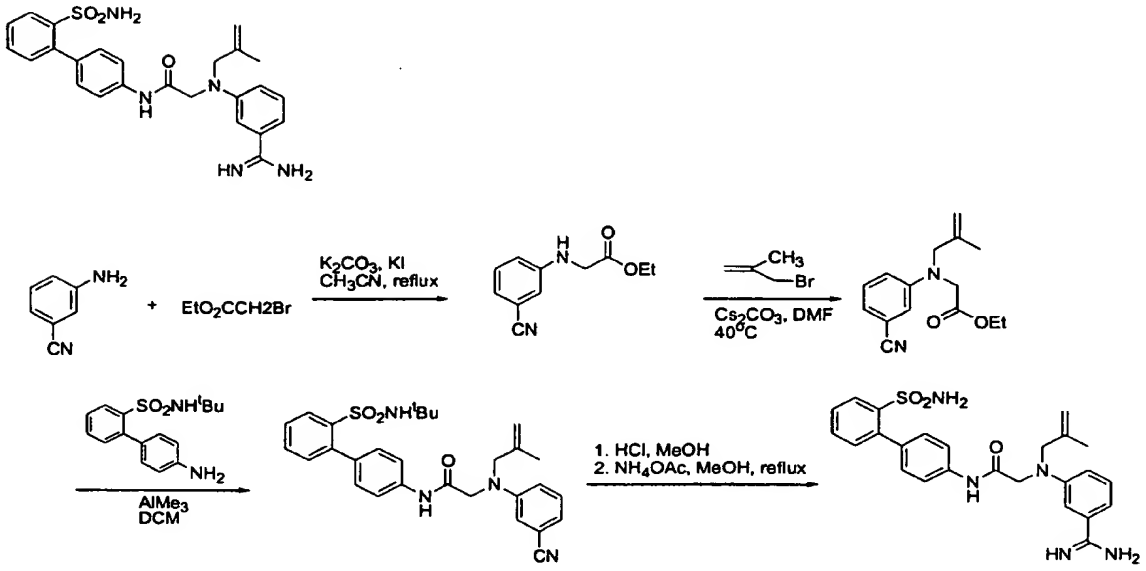
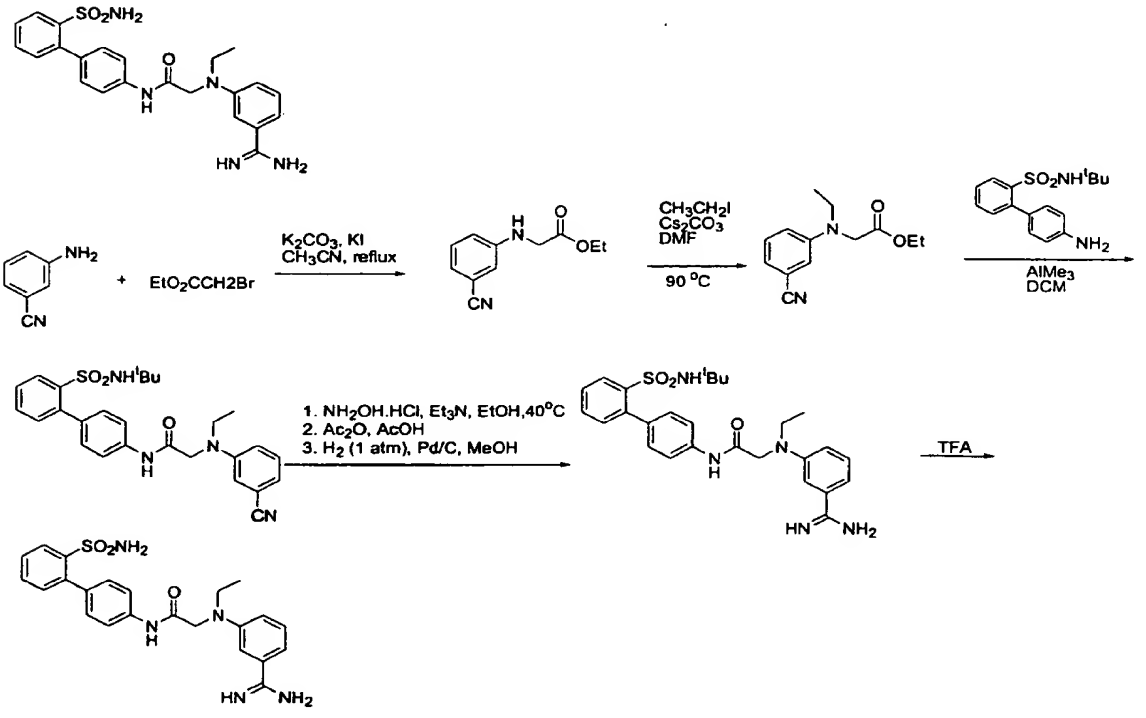
Scheme 6



Scheme 7



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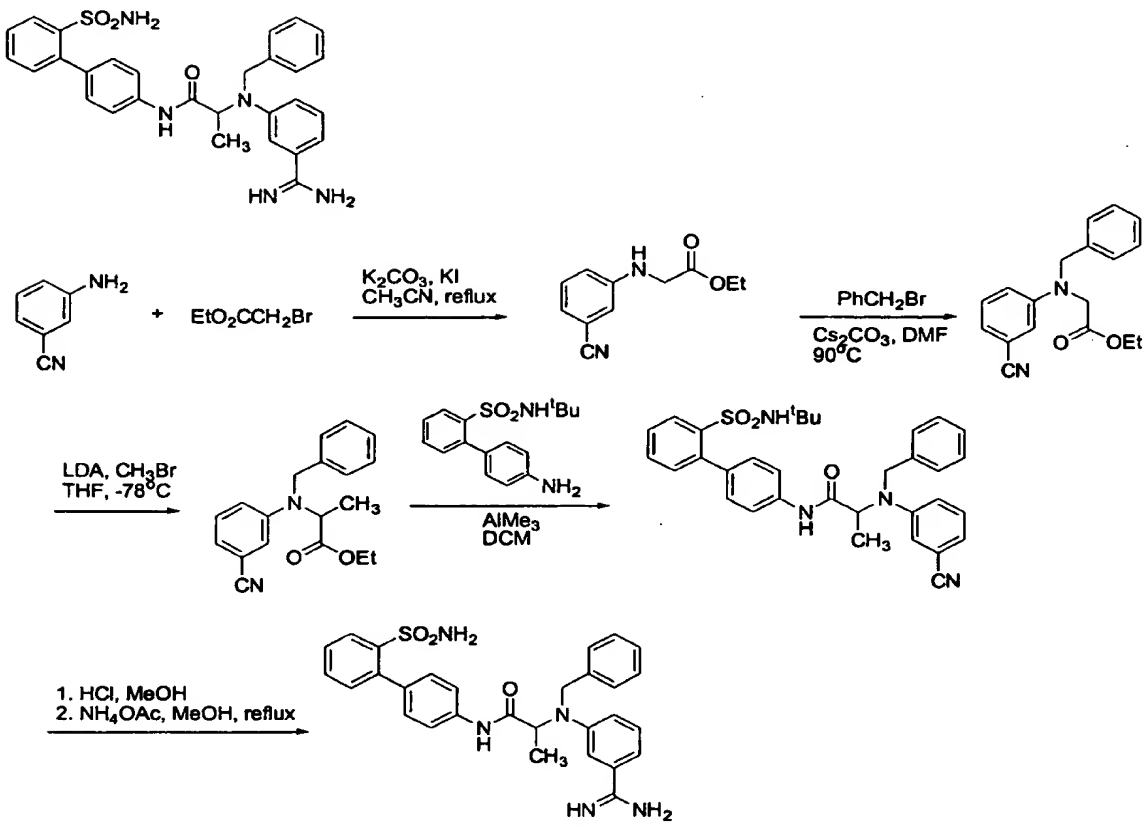
Scheme 8



Non-limiting exemplary synthesis schemes are outlined directly below, and specific steps are described in the Examples. The reaction products are isolated and purified by conventional methods, typically by solvent extraction into a compatible solvent. The products may be further purified by column chromatography or other appropriate methods.

Compositions and Formulations

The compounds of this invention may be isolated as the free acid or base or converted to salts of various inorganic and organic acids and bases. Such salts are within the scope of this invention. Non-toxic and physiologically compatible salts are particularly useful although other less desirable salts may have use in the processes of isolation and purification.

A number of methods are useful for the preparation of the salts described above and are known to those skilled in the art. For example, reaction of the free acid or free base form of a compound of the structures recited above with one or more molar equivalents of the desired acid or base in a solvent or solvent mixture in which the salt is insoluble, or in a solvent like water after which the solvent is removed by evaporation, distillation or freeze drying. Alternatively, the free acid or base form of the product may be passed over an ion exchange resin to form the desired salt or one salt form of the product may be converted to another using the same general process.

Diagnostic applications of the compounds of this invention will typically utilize formulations such as solution or suspension. In the management of thrombotic disorders the compounds of this invention may be utilized in compositions such as tablets, capsules or elixirs for oral administration, suppositories, sterile solutions or suspensions or injectable administration, and the like, or incorporated into shaped articles. Subjects in need of treatment (typically mammalian) using the compounds of this invention can be administered dosages that will provide optimal efficacy. The dose and method of administration will vary from subject to subject and be dependent upon such factors as the type of mammal being treated, its sex, weight, diet, concurrent medication, overall clinical condition,

the particular compounds employed, the specific use for which these compounds are employed, and other factors which those skilled in the medical arts will recognize.

Formulations of the compounds of this invention are prepared for storage or administration by mixing the compound having a desired degree of purity with physiologically acceptable carriers, excipients, stabilizers etc., and may be provided in sustained release or timed release formulations. Acceptable carriers or diluents for therapeutic use are well known in the pharmaceutical field, and are described, for example, in *Remington's Pharmaceutical Sciences*, Mack Publishing Co., (A.R. Gennaro edit. 1985). Such materials are nontoxic to the recipients at the dosages and concentrations employed, and include buffers such as phosphate, citrate, acetate and other organic acid salts, antioxidants such as ascorbic acid, low molecular weight (less than about ten residues) peptides such as polyarginine, proteins, such as serum albumin, gelatin, or immunoglobulins, hydrophilic polymers such as polyvinalpyrrolidinone, amino acids such as glycine, glutamic acid, aspartic acid, or arginine, monosaccharides, disaccharides, and other carbohydrates including cellulose or its derivatives, glucose, mannose or dextrins, chelating agents such as EDTA, sugar alcohols such as mannitol or sorbitol, counterions such as sodium and/or nonionic surfactants such as Tween, Pluronics or polyethyleneglycol.

Dosage formulations of the compounds of this invention to be used for therapeutic administration must be sterile. Sterility is readily accomplished by filtration through sterile membranes such as 0.2 micron membranes, or by other conventional methods. Formulations typically will be stored in lyophilized form or as an aqueous solution. The pH of the preparations of this invention typically will be between 3 and 11, more preferably from 5 to 9 and most preferably from 7 to 8. It will be understood that use of certain of the foregoing excipients, carriers, or stabilizers will result in the formation of cyclic polypeptide salts. While the

preferred route of administration is by injection, other methods of administration are also anticipated such as intravenously (bolus and/or infusion), subcutaneously, intramuscularly, colonically, rectally, nasally or intraperitoneally, employing a variety of dosage forms such as suppositories, implanted pellets or small cylinders, aerosols, oral dosage formulations and topical formulations such as ointments, drops and dermal patches. The compounds of this invention are desirably incorporated into shaped articles such as implants which may employ inert materials such as biodegradable polymers or synthetic silicones, for example, Silastic, silicone rubber or other polymers commercially available.

The compounds of this invention may also be administered in the form of liposome delivery systems, such as small unilamellar vesicles, large unilamellar vesicles and multilamellar vesicles. Liposomes can be formed from a variety of lipids, such as cholesterol, stearylamine or phosphatidylcholines.

The compounds of this invention may also be delivered by the use of antibodies, antibody fragments, growth factors, hormones, or other targeting moieties, to which the compound molecules are coupled. The compounds of this invention may also be coupled with suitable polymers as targetable drug carriers. Such polymers can include polyvinylpyrrolidone, pyran copolymer, polyhydroxy-propyl-methacrylamide-phenol, polyhydroxyethyl-aspartamide-phenol, or polyethyleneoxide-polylysine substituted with palmitoyl residues. Furthermore, the factor Xa inhibitors of this invention may be coupled to a class of biodegradable polymers useful in achieving controlled release of a drug, for example polylactic acid, polyglycolic acid, copolymers of polylactic and polyglycolic acid, polyepsilon caprolactone, polyhydroxy butyric acid, polyorthoesters, polyacetals, polydihydropyrans, polycyanoacrylates and cross linked or amphipathic block

copolymers of hydrogels. Polymers and semipermeable polymer matrices may be formed into shaped articles, such as valves, stents, tubing, prostheses and the like.

Therapeutic compound liquid formulations generally are placed into a container having a sterile access port, for example, an intravenous solution bag or vial having a stopper pierceable by hypodermic injection needle.

Therapeutically effective dosages may be determined by either *in vitro* or *in vivo* methods. For each particular compound of the present invention, individual determinations may be made to determine the optimal dosage required. The range of therapeutically effective dosages will naturally be influenced by the route of administration, the therapeutic objectives, and the condition of the patient. For injection by hypodermic needle, it may be assumed the dosage is delivered into the body's fluids. For other routes of administration, the absorption efficiency must be individually determined for each inhibitor by methods well known in pharmacology. Accordingly, it may be necessary for the therapist to titer the dosage and modify the route of administration as required to obtain the optimal therapeutic effect. The determination of effective dosage levels, that is, the dosage levels necessary to achieve the desired result, will be within the ambit of one skilled in the art. Typically, applications of compound are commenced at lower dosage levels, with dosage levels being increased until the desired effect is achieved.

A typical dosage might range from about 0.001 mg/kg to about 1000 mg/kg, preferably from about 0.01 mg/kg to about 100 mg/kg, and more preferably from about 0.10 mg/kg to about 20 mg/kg. Advantageously, the compounds of this invention may be administered several times daily, and other dosage regimens may also be useful.

Typically, about 0.5 to 500 mg of a compound or mixture of compounds of this invention, as the free acid or base form or as a pharmaceutically acceptable salt, is compounded with a physiologically acceptable vehicle, carrier, excipient, binder, preservative, stabilizer, dye, flavor etc., as called for by accepted pharmaceutical practice. The amount of active ingredient in these compositions is such that a suitable dosage in the range indicated is obtained.

Typical adjuvants which may be incorporated into tablets, capsules and the like are a binder such as acacia, corn starch or gelatin, and excipient such as microcrystalline cellulose, a disintegrating agent like corn starch or alginic acid, a lubricant such as magnesium stearate, a sweetening agent such as sucrose or lactose, or a flavoring agent. When a dosage form is a capsule, in addition to the above materials it may also contain a liquid carrier such as water, saline, a fatty oil. Other materials of various types may be used as coatings or as modifiers of the physical form of the dosage unit. Sterile compositions for injection can be formulated according to conventional pharmaceutical practice. For example, dissolution or suspension of the active compound in a vehicle such as an oil or a synthetic fatty vehicle like ethyl oleate, or into a liposome may be desired. Buffers, preservatives, antioxidants and the like can be incorporated according to accepted pharmaceutical practice.

In practicing the methods of this invention, the compounds of this invention may be used alone or in combination, or in combination with other therapeutic or diagnostic agents. In certain preferred embodiments, the compounds of this inventions may be coadministered along with other compounds typically prescribed for these conditions according to generally accepted medical practice, such as anticoagulant agents, thrombolytic agents, or other antithrombotics, including platelet aggregation inhibitors, tissue plasminogen activators, urokinase,

prourokinase, streptokinase, heparin, aspirin, or warfarin. The compounds of this invention can be utilized in vivo, ordinarily in mammals such as primates, such as humans, sheep, horses, cattle, pigs, dogs, cats, rats and mice, or *in vitro*.

The preferred compounds of the present invention are characterized by their ability to inhibit thrombus formation with acceptable effects on classical measures of coagulation parameters, platelets and platelet function, and acceptable levels of bleeding complications associated with their use. Conditions characterized by undesired thrombosis would include those involving the arterial and venous vasculature.

With respect to the coronary arterial vasculature, abnormal thrombus formation characterizes the rupture of an established atherosclerotic plaque which is the major cause of acute myocardial infarction and unstable angina, as well as also characterizing the occlusive coronary thrombus formation resulting from either thrombolytic therapy or percutaneous transluminal coronary angioplasty (PTCA).

With respect to the venous vasculature, abnormal thrombus formation characterizes the condition observed in patients undergoing major surgery in the lower extremities or the abdominal area who often suffer from thrombus formation in the venous vasculature resulting in reduced blood flow to the affected extremity and a predisposition to pulmonary embolism. Abnormal thrombus formation further characterizes disseminated intravascular coagulopathy commonly occurs within both vascular systems during septic shock, certain viral infections and cancer, a condition wherein there is rapid consumption of coagulation factors and systemic coagulation which results in the formation of life-threatening thrombi occurring throughout the microvasculature leading to widespread organ failure.

The compounds of this present invention, selected and used as disclosed herein, are believed to be useful for preventing or treating a condition characterized by undesired thrombosis, such as (a) the treatment or prevention of any thrombotically mediated acute coronary syndrome including myocardial infarction, unstable angina, refractory angina, occlusive coronary thrombus occurring post-thrombolytic therapy or post-coronary angioplasty, (b) the treatment or prevention of any thrombotically mediated cerebrovascular syndrome including embolic stroke, thrombotic stroke or transient ischemic attacks, (c) the treatment or prevention of any thrombotic syndrome occurring in the venous system including deep venous thrombosis or pulmonary embolus occurring either spontaneously or in the setting of malignancy, surgery or trauma, (d) the treatment or prevention of any coagulopathy including disseminated intravascular coagulation (including the setting of septic shock or other infection, surgery, pregnancy, trauma or malignancy and whether associated with multi-organ failure or not), thrombotic thrombocytopenic purpura, thromboangiitis obliterans, or thrombotic disease associated with heparin induced thrombocytopenia, (e) the treatment or prevention of thrombotic complications associated with extracorporeal circulation (e.g. renal dialysis, cardiopulmonary bypass or other oxygenation procedure, plasmapheresis), (f) the treatment or prevention of thrombotic complications associated with instrumentation (e.g. cardiac or other intravascular catheterization, intra-aortic balloon pump, coronary stent or cardiac valve), and (g) those involved with the fitting of prosthetic devices.

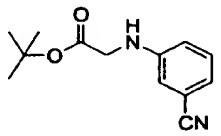
Anticoagulant therapy is also useful to prevent coagulation of stored whole blood and to prevent coagulation in other biological samples for testing or storage. Thus the compounds of this invention can be added to or contacted with any medium containing or suspected to contain factor Xa and in which it is desired that blood coagulation be inhibited, e.g., when contacting the mammal's blood with material

such as vascular grafts, stents, orthopedic prostheses, cardiac stents, valves and prostheses, extra corporeal circulation systems and the like.

Without further description, it is believed that one of ordinary skill in the art can, using the preceding description and the following illustrative examples, make and utilize the compounds of the present invention and practice the claimed methods. The following working examples therefore, specifically point out preferred embodiments of the present invention, and are not to be construed as limiting in any way the remainder of the disclosure.

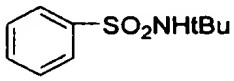
EXAMPLES

Example 1



To a solution of t-butyl bromoacetate (6.1 mL, 37.5 mmol), 3-aminobenzonitrile (2.95 g, 25 mmol), potassium carbonate (10.4 g, 75 mmol) in CH_3CN (50 mL), was added KI (0.83 g, 5 mmol). The mixture was heated to reflux for 2 hrs. The mixture was cooled to room temperature and solvent was removed in vacuo. Ether and water were added to the mixture and organic layer was washed with 2N NaOH, brine, dried over Na_2SO_4, filtered and the filtrated were concentrated *in vacuo* to give the title compound (5.7 g, 98.3%). ES-MS (M+H)+ = 233.1.

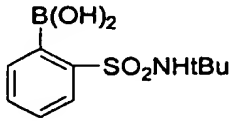
Example 2



To a solution of tert-Butylamine (41.4g, 566 mmol) and triethylamine (118 mL, 849 mmol) in DCM (1000 mL) in an ice bath, was added benzenesulfonyl chloride (100 g, 566 mmol) dropwise. The mixture was stirred at room temperature overnight.

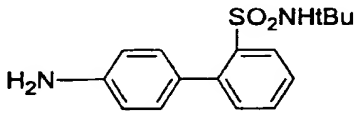
Water was added to the mixture and organic layer was washed with water, brine, dried over Na_2SO_4, filtered and filtrated evaporated *in vacuo* to give the title compound as light yellowish solid (117.63 g, 97.6%). ES-MS (M+H)+ = 214.1.

Example 3



To a solution of compound of example 6 (53.25 g, 250 mmol) in THF (600 mL) in an ice bath, was added n-butyllithium in hexane (200 mL, 500 mmol) dropwise. A thick precipitate was formed when the reaction mixture was warmed up to 10°C. Triisopropylborate was added keeping the temperature below 35°C. After 1 hr., the mixture was cooled in an ice bath, 1N HCl (405 mL) was added, and the mixture was stirred overnight. The mixture was extracted with ether (100 mL) three times. The combined organic extracts were extracted with 1N NaOH (130 mL) three times. The aqueous extracts were acidified to pH 1 with 12 N HCl, and then extracted with ether three times (140 ML). The combined ether extracts were dried over $MgSO_4$, and solvents evaporated *in vacuo*. Hexane and ether were added and a white precipitate was formed. The solid was collected and washed with 10% ether/hexane to give the title compound. ES-MS (M+H)+ = 258.1.

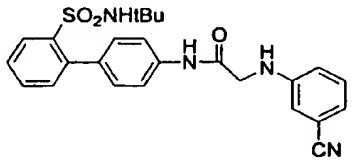
Example 4



To a solution of compound of example 7 (6.4 g, 25 mmol) in toluene (120 mL) was added water (15 mL), 5N NaOH solution (38.5 mL), isopropanol (60 mL), 4-bromoaniline and tetrakis(triphenylphosphine)palladium(0). The mixture was refluxed for six hours, cooled to room temperature, diluted with EtOAc. The organic layer was washed with water, dried with $MgSO_4$, filtered and concentrated. This was purified by silica gel column chromatography using solvent system 30%

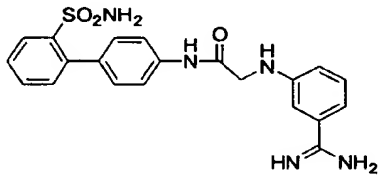
EtOAc in hexane as eluent to give the title compound (5g, 66%). ES-MS M+H = 305.1.

Example 5



The compound of example 5 (0.75 mmol) was treated with 50% TFA in DCM (4 mL). The mixture was stirred at room temperature for 30 minutes and solvent evaporated to give a white solid. This was dissolved in DMF (2 mL) and cooled to 0°C. The solution was neutralized with DIEA (0.26 mL, 1.5 mmol) followed by the addition of compound of example 8 (108 mg, 0.35 mmol) and coupling reagent HATU (285 mg, 0.75 mmol). The solution was stirred at room temperature for 15 hours. The reaction mixture was diluted in a mixture of EtOAc/H_2O (10 mL:5mL). The organic layer was washed with sat. $NaHCO_3$ (2 X 10 mL), sat. NaCl (2 X 10 mL), dried over $MgSO_4$, filtered and solvent evaporated to give the crude product. This was purified by silica gel column chromatography using solvent system 50% EtOAc in hexane as eluent to give the title compound (60 mg, 37.2%). ES-MS (M+Na)+= 485.1.

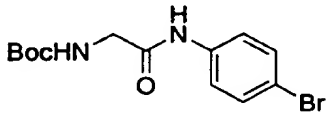
Example 6



A solution of the compound of example 9 (60 mg, 0.13 mmol), hydroxylamine hydrochloride (18.1mg, 0.26 mmol), TEA (54.3 µL, 0.39 mmol) in absolute ethanol (4 mL) was heated up to 60°C and stirred for 15 hrs. The solution was cooled and solvent evaporated. The residue was dissolved in AcOH (2 mL). Ac_2O (49 µL, 0.52 mmol) was added. The mixture was stirred at room temperature for 50 min. and the solvent evaporated. The residue was dissolved in MeOH (2-3 mL) and 10% Pd/C

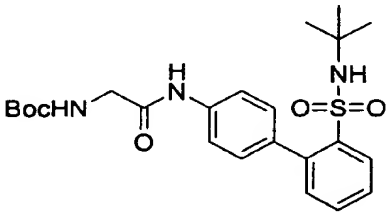
(catalytic amount) was added. The mixture was hydrogenated under balloon overnight, filtered through Celite to remove the catalyst and the filtrate was evaporated. TFA (2-3 mL) was added to the residue and the mixture was stirred at room temperature for 2-3 hrs. TFA was removed under reduced pressure to give the crude product. The obtained residue was purified by RP-HPLC to give the title compound as a white powder. ES-MS (M+H)+ = 424.1.

Example 7



Boc-Gly-OH (1.75 g, 10 mmol) and 4-bromoaniline (1.89 g, 11 mmol) were dissolved in DMF (25 mL). DIEA (3.48 mL, 20 mmol) was added followed by the addition of the coupling reagent BOP (4.87 g, 11 mmol). The solution was stirred at room temperature for 12 hours. The reaction mixture was diluted in a mixture of EtOAc/H_2O (100 mL:40 mL). The organic layer was washed with water, saturated Na_2CO_3, water, 1M $KHSO_4$, brine, dried over $MgSO_4$, filtered and solvent evaporated to give the title compound (1.123g, 34%). ES-MS (M+Na)+ = 353.

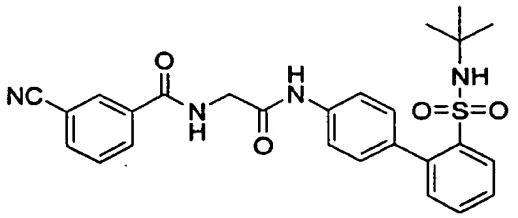
Example 8



A mixture of compound of example 11 (328 mg, 1 mmol) and 2-(t-butylamino)sulfonyl-phenylboronic acid (262 mg, 1.02 mmol), tetrakis(triphenylphosphine)palladium(0) (58 mg, 0.05mmol), tetrabutylammonium bromide (16 mg, 0.05 mmol) , and potassium carbonate (147 mg, 2.13 mmol in 0.64 mL water) were refluxed with toluene (6 mL) under N_2 for 6h. The toluene was removed *in vacuo* and the residue was dissolved in methylene chloride and water.

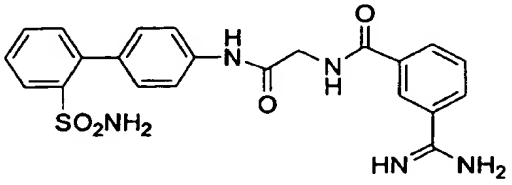
The two phases were separated and organic phase was washed with water and brine, dried over $MgSO_4$ and concentrated to give the title compound (445 mg, 96.5%). (M+H)+ = 462.1.

Example 9



The compound of example 12 (386 mg, 0.84 mmol) was treated with 50% TFA in DCM (2 mL). The mixture was stirred at room temperature for 30 minutes then solvent evaporated to give a white solid. This was dissolved in DMF (5 mL) and cooled to 0°C. The solution was neutralized with DIEA (0.44 mL, 2.52 mmol) followed by addition of 3-cyano-benzoic acid (147 mg, 1 mmol) and coupling reagent BOP (442.5 mg, 1 mmol). The solution was stirred at room temperature for 15 hours. The reaction mixture was diluted in a mixture of EtOAc/H_2O (10 mL:5mL). The organic layer was washed with sat. $NaHCO_3$ (2 X 20 mL), sat. NaCl (2 X 20 mL), dried over $MgSO_4$, filtered and solvent evaporated to give the crude product. This was purified by silica gel column chromatography using solvent system 50% EtOAc in hexane as eluent to give the title compound (148 mg, 30%). ES-MS (M+Na)+ = 513.

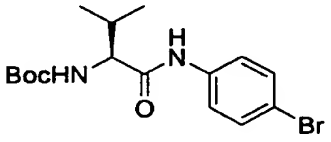
Example 10



A solution of the compound of example 13 (61.3 mg, 0.125 mmol), hydroxylamine hydrochloride (17.4mg, 0.25 mmol), TEA (52 µL, 0.375 mmol) in absolute ethanol (4 mL) was heated up to 60°C and stirred for 15 hrs. The solution was cooled and solvent evaporated. The residue was dissolved in AcOH (2 mL). Ac_2O (47 µL, 0.5

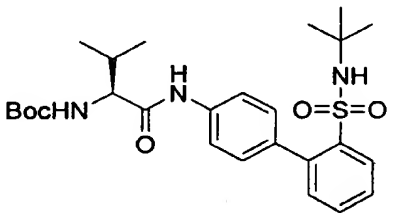
mmol) was added. The mixture was stirred at room temperature for 50 min. and the solvent evaporated. The residue was dissolved in MeOH (2-3 mL) and 10% Pd/C (catalytic amount) was added. The mixture was hydrogenated under balloon overnight, filtered through Celite to remove the catalyst and the filtrate was evaporated. TFA (2-3 mL) was added to the residue and the mixture was stirred at room temperature for 2-3 hrs. TFA was removed under reduced pressure to give the crude product. The obtained residue was purified by RP-HPLC to give the title compound as a white powder. ES-MS (M+H)+ = 452.1.

Example 11



Boc-Val-OH (2.17 g, 10 mmol) and 4-bromoaniline (1.89 g, 11 mmol) were dissolved in DMF (25 mL). DIEA (3.48 mL, 20 mmol) was added followed by the addition of the coupling reagent BOP (4.87 g, 11 mmol). The solution was stirred at room temperature for 12 hours. The reaction mixture was diluted in a mixture of EtOAc/H_2O (100 mL:40 mL). The organic layer was washed with water, sat. Na_2CO_3, water, 1M $KHSO_4$, brine, dried over $MgSO_4$, filtered and solvent evaporated to give the title compound (3.53g, 95.4%). ES-MS (M+H)+ = 371.

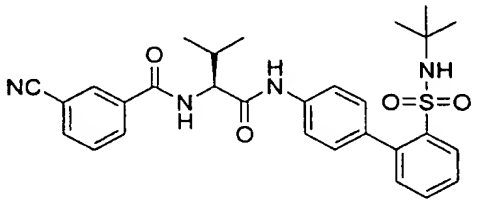
Example 12



A mixture of compound of example 15 (740 mg, 2 mmol) and 2-(t-butylamino)-sulfonyl-phenylboronic acid (616 mg, 2.4 mmol), tetrakis(triphenylphosphine)palladium(0) (115.5 mg, 0.1mmol), tetrabutylammonium bromide (32.2 mg, 0.1 mmol) , and potassium carbonate (691 mg, 5 mmol in 1.5 mL water) were refluxed with toluene (12 mL) under N_2 for 6h.

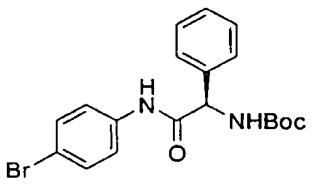
The toluene was removed *in vacuo* and the residue was dissolved in methylene chloride and water. The two phases were separated and organic phase was washed with water and brine, dried over $MgSO_4$ and concentrated to give the title compound (1.156g, 100%). (M+H)+ = 504.1.

Example 13



The compound of example 16 (0.7 mmol) was treated with 50% TFA in DCM (2 mL). The mixture was stirred at room temperature for 30 minutes then solvent evaporated to give a white solid. This was dissolved in DMF (2 mL) and cooled to 0°C. The solution was neutralized with DIEA (0.37 mL, 2.1 mmol) followed by addition of 3-cyano-benzoic acid (124 mg, 0.84 mmol) and coupling reagent BOP (371.7 mg, 0.84 mmol). The solution was stirred at room temperature for 15 hours. The reaction mixture was diluted in a mixture of EtOAc/H_2O (10 mL:5mL). The organic layer was washed with sat. $NaHCO_3$ (2 X 20 mL), sat. NaCl (2 X 20 mL), dried over $MgSO_4$, filtered and solvent evaporated to give the crude product. This was purified by silica gel column chromatography using solvent system 40% EtOAc in hexane as eluent to give the title compound (204 mg, 55%). ES-MS (M+Na)+ = 555.2.

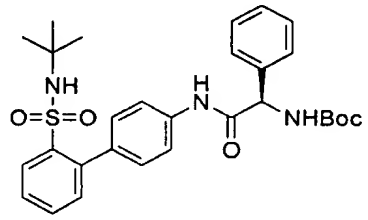
Example 14



Boc-Phg-OH (1.19 g, 4.73 mmol) and 4-bromoaniline (0.895 g, 5.2 mmol) were dissolved in DMF (25 mL). DIEA (1.65 mL, 9.46 mmol) was added followed by the

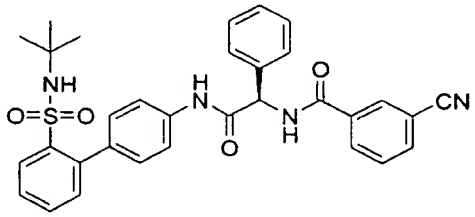
addition of the coupling reagent BOP (2.3 g, 5.2 mmol). The solution was stirred at room temperature for 12 hours. The reaction mixture was diluted in a mixture of EtOAc/H_2O (100 mL:40 mL). The organic layer was washed with water, sat. Na_2CO_3, water, 1M $KHSO_4$, brine, dried over $MgSO_4$, filtered and solvent evaporated to give the title compound. ES-MS (M+H)+ = 405.

Example 15



A mixture of compound of example 23 (472 mg, 1.17 mmol) and 2-(t-butylamino)sulfonyl-phenylboronic acid (359.5 mg, 1.4 mmol), tetrakis(triphenylphosphine)palladium(0) (67.6 mg, 0.0585mmol), tetrabutylammonium bromide (18.9 mg, 0.0585 mmol) , and potassium carbonate (404 mg, 2.93 mmol in 0.88 mL water) were refluxed with toluene (6 mL) under N_2 for 6 hrs. The toluene was removed *in vacuo* and the residue was dissolved in methylene chloride and water. The two phases were separated and organic phase was washed with water and brine, dried over $MgSO_4$ and concentrated to give the title compound(435mg, 69.2%). (M+H)+ = 538.1.

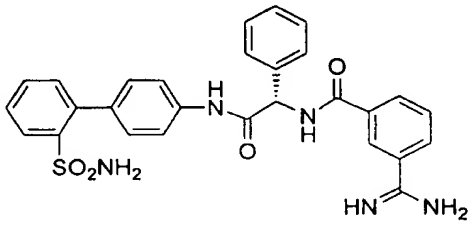
Example 16



The compound of example 24 (362 mg, 0.67 mmol) was treated with 50% TFA in DCM (2 mL). The mixture was stirred at room temperature for 30 minutes then solvent evaporated to give a white solid. This was dissolved in DMF (5 mL) and cooled to 0°C. The solution was neutralized with DIEA (0.35 mL, 2.02 mmol)

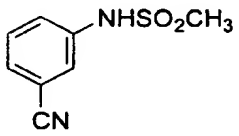
followed by addition of 3-CN-benzoic acid (119 mg, 0.81 mmol) and coupling reagent BOP (358 mg, 0.81 mmol). The solution was stirred at room temperature for 15 hours. The reaction mixture was diluted in a mixture of EtOAc/H_2O (10 mL:5mL). The organic layer was washed with sat. $NaHCO_3$ (2 X 20 mL), sat. NaCl (2 X 20 mL), dried over $MgSO_4$, filtered and solvent evaporated to give the crude product. This was purified by silica gel column chromatography using solvent system 40% EtOAc in hexane as eluent to give the title compound (137 mg, 36%). ES-MS (M+Na)+ = 589.

Example 17



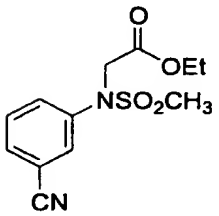
A solution of the compound of example 25 (36.9 mg, 0.065 mmol), hydroxylamine hydrochloride (9 mg, 0.13 mmol), TEA (27 µL, 0.20 mmol) in absolute ethanol (4 mL) was heated up to 60°C and stirred at room temperature for 15 hrs. The solution was cooled and solvent evaporated. The residue was dissolved in AcOH (2 mL). Ac_2O (24.5 µL, 0.26 mmol) was added. The mixture was stirred at room temperature for 50 min. and the solvent evaporated. The residue was dissolved in MeOH (2-3 mL) and 10% Pd/C (catalytic amount) was added. The mixture was hydrogenated under balloon overnight, filtered through Celite to remove the catalyst and the filtrate was evaporated. TFA (2-3 mL) was added to the residue and the mixture was stirred at room temperature for 2-3 hrs. TFA was removed under reduced pressure to give the crude product. The obtained residue was purified by RP-HPLC to give the title compound as a white powder. ES-MS (M+H)+ = 528.1.

Example 18



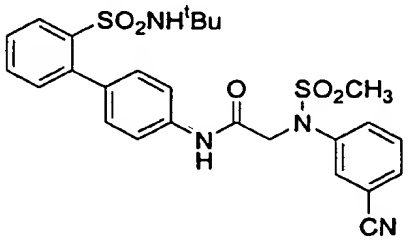
To a solution of 3-aminobenzonitrile (1.18g, 10mmol) in pyridine (2ml) was added methanesulfonyl chloride (1.15g, 10mmol) dropwise at 0 °C. The mixture was stirred at 0 °C for 1 hr, then warmed up to room temperature and stirred for 1 hr. After the concentration *in vacuo*, the residue was diluted with ethyl acetate and washed with 0.1N hydrochloride. The organic layer was dried over magnesium sulfate and concentrated *in vacuo to* give yellow solid (1.83g, 93%). ES-MS (M+H)+=197.

Example 19



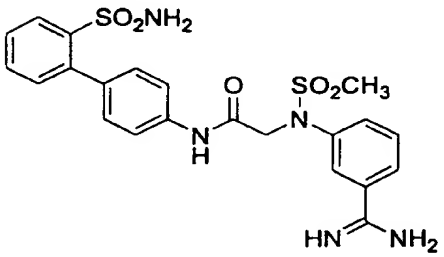
To a solution of the compound of example 18 (196mg, 1mmol) in dimethylformamide (2ml) was added ethyl bromoacetate (200mg, 1.2mmol) and potassium carbonate (332mg, 2.4mmol). The mixture was stirred at room temperature for 1 hr. After the filtration of the precipitate, the residue was concentrated *in vacuo* to give the title compound (300mg, 100%). ES-MS (M+H)+=283.

Example 20



To a solution of [2-(4-aminophenyl)phenylsulfonyl](t-butyl)amine (304mg, 1mmol) in dichloromethane (2ml) was added 2M trimethylaluminum in hexane (1.5ml, 3mmol). The mixture was stirred at room temperature for 30minutes, methane gas evolved. A solution of the compound of example 19 (282mg, 1mmol) in dichloromethane (2ml) was added. The mixture was stirred at room temperature overnight. 1N hydrochloride was added to acidify the solution to PH=2. After the addition of water and dichloromethane, the organic layer was separated, and the aqueous layer was extracted with dichloromethane. The combined organic extracts were dried over magnesium sulfate, and concentrated *in vacuo*. The crude residue was purified by silica gel column chromatography using solvent system 25% ethyl acetate in hexane as eluent to give the title compound as a solid (240mg, 56%). ES-MS (M+H)+=541.

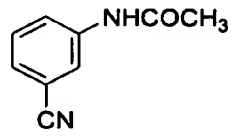
Example 21



A solution of the compound of example 20 (150mg, 0.28mmol), hydroxylamine hydrochloride (48mg, 0.69mmol), tirethylamine (97ul, 0.69mmol) in absolute ethanol (2ml) was stirred at 40 °C for 15 hrs. After the evaporation of the solvent *in vacuo*, the residue was dissolved in acetic acid (5ml), and acetic anhydride (53ul, 0.56mmol) was added. The mixture was stirred at room temperature for 3 hrs. It was diluted with absolute methanol (3ml), and 10% Pd/C (catalytic amount) was added.

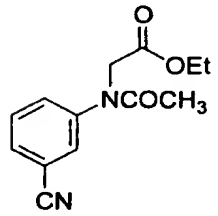
The mixture was applied with 1 atm hydrogen for 6 hrs. After the filtration through Celite to remove the catalyst, the filtrate was concentrated *in vacuo*. The residue was dissolved in trifluoroacetic acid (5ml). The mixture was refluxed for 1.5 hrs. After the evaporation of the solvent *in vacuo*, the crude residue was purified by RP-HPLC to give the title compound as a white powder (60mg, 63%). ES-MS (M+H)+=502.

Example 22



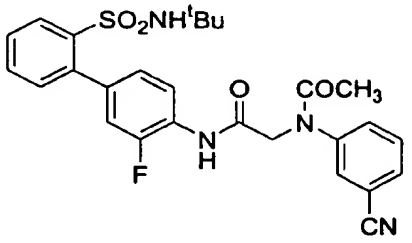
To a solution of 3-aminobenzonitrile (3.54g, 30mmol) in pyridine (10ml) was added acetyl chloride (2.36g, 30mmol) dropwise at 0 °C . The mixture was stirred at 0 °C for 1 hr, then warmed up to room temperature and stirred for 1 hr. After the concentration *in vacuo*, the residue was diluted with ethyl acetate and washed with 0.1N hydrochloride. The organic layer was dried over magnesium sulfate and concentrated to give the title compound (4.84g, 100%). ES-MS (M+H)+=161.

Example 23



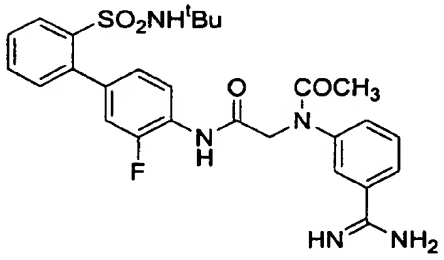
To a solution of the compound of example 22 (320mg, 2mmol) in dimethylformamide (4ml) was added ethyl bromoacetate (401mg, 2.4mmol) and potassium carbonate (663mg, 4.8mmol). The mixture was stirred at room temperature for 1 hr. After the filtration of the precipitate, the residue was concentrated *in vacuo* to give oil (440mg, 89%). ES-MS (M+H)+=247.

Example 24



To a solution of [2-(4-amino-3-fluorophenyl)phenylsulfonyl](t-butyl)amine (131mg, 0.41mmol) in dichloromethane (5ml) was added 2M trimethylaluminum in hexane(0.61ml, 1.22mmol). The mixture was stirred at room temperature for 30min, methane gas evolved. A solution of example 23 (100mg, 0.41mmol) in dichloromethane (2ml) was added. The mixture was stirred at room temperature overnight. 1N hydrochloride was added to acidify the solution to PH=2. After the addition of water and dichloromethane, the organic layer was separated, and the aqueous layer was extracted with dichloromethane. The combined organic extracts were dried over magnesium sulfate, and concentrated *in vacuo*. The crude residue was purified by silica gel column chromatography using solvent system 50% ethyl acetate in hexane as eluent to give the title compound as a solid (110mg, 68%). ES-MS (M+H)+=523.

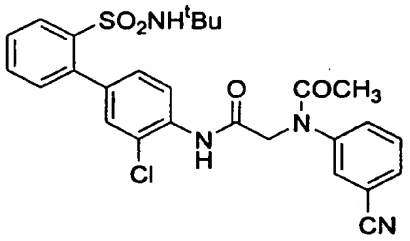
Example 25



To a solution of the compound of example 24 (110mg, 0.21mmol) in absolute methanol (3ml) in an ice bath was saturated with hydrochloride gas for 10 minutes. The mixture was stirred at room temperature for 3 hrs. After the evaporation of the solvent *in vacuo*, the residue was dissolved in absolute methanol (3ml), and

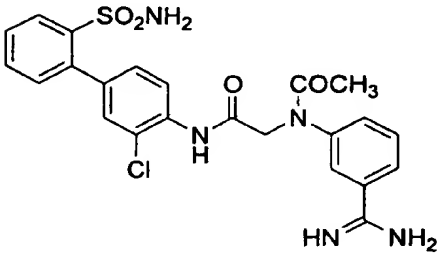
ammonia acetate (97mg, 1.26mmol) was added. The mixture was refluxed for 3 hrs. The solvent was evaporated *in vacuo*. The crude residue was purified by RP-HPLC to give the title compound as a white powder (21mg, 21%). ES-MS (M+H)+=484.

Example 26



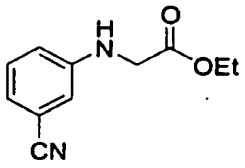
To a solution of [2-(4-amino-3-chlorophenyl)phenylsulfonyl](t-butyl)amine (138mg, 0.41mmol) in dichloromethane (5ml) was added 2M trimethylaluminum in hexane (0.61ml, 1.22mmol). The mixture was stirred at room temperature for 30 minutes, methane gas evolved. A solution of example 23 (100mg, 0.41mmol) in dichloromethane (2ml) was added. The mixture was stirred at room temperature overnight. 1N hydrochloride was added to acidify the solution to PH=2. After the addition of water and dichloromethane, the organic layer was separated, and the aqueous layer was extracted with dichloromethane. The combined organic extracts were dried over magnesium sulfate, and concentrated *in vacuo*. The crude residue was purified by silica gel column chromatography using solvent system 50% ethyl acetate in hexane as eluent to give the title compound as a solid (110mg, 64%). ES-MS (M+H)+=539.5.

Example 27



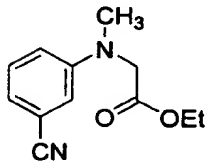
To a solution of the compound of example 26 (110mg, 0.20mmol) in absolute methanol (3ml) in an ice bath was saturated with hydrochloride gas for 10 minutes. The mixture was stirred at room temperature for 3 hrs. After the evaporation of the solvent *in vacuo*, the residue was dissolved in absolute methanol (3ml), and ammonia acetate (95mg, 1.23mmol) was added. The mixture was refluxed for 3 hrs. The solvent was evaporated *in vacuo*. The crude residue was purified by RP-HPLC to give the title compound as a white powder (18mg, 18%). ES-MS (M+H)+=500.

Example 28



To a solution of ethyl bromoacetate (10.6g, 60mmol), 3-aminobenzonitrile (5g, 40mmol), and potassium carbonate (17.5g, 120mmol) in acetonitrile (30ml) was added potassium iodide (1.4g, 8mmol). The mixture was heated to reflux for 6 hrs. The mixture was cooled to room temperature, and solvent was removed *in vacuo*. Ether and water were added to the mixture. Organic layer was washed with 1N hydrochloride and brine, and dried over magnesium sulfate. After the concentration *in vacuo*, the crude residue was purified by silica gel column chromatography using solvent system 15% ethyl acetate in hexane as eluent to give the title compound as light yellowish solid (7.94g, 97%). ES-MS (M+H)+ = 205.

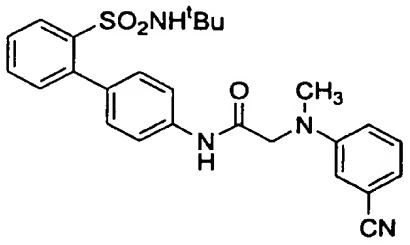
Example 29



To a solution of the compound of example 28 (200mg, 1mmol) and cesium carbonate (650mg, 2mmol) in dimethylformamide (5ml) was added iodomethane (75ul, 1.2mmol). The mixture was stirred at 90 °C for 2 hrs. After the filtration of the solid, the filtrate was concentrated *in vacuo*, and the residue was purified by silica

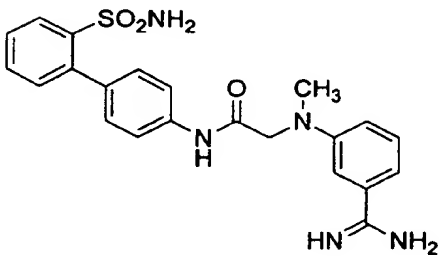
gel column chromatography using solvent system 15% ethyl acetate in hexane as eluent to give the title compound as an oil (270mg, 100%). ES-MS (M+H)+ = 219.

Example 30



To a solution of the compound of [2-(4-aminophenyl)phenylsulfonyl](t-butyl)amine (126mg, 0.41mmol) in dichloromethane (5ml) was added 2.0M trimethylaluminum in hexane (0.62ml, 1.24mmol). The mixture was stirred at room temperature for 30 minutes, methane gas evolved. A solution of the compound of example 29 (90mg, 0.41mmol) in dichlodomethane (1ml) was added. The mixture was stirred at room temperature overnight. 1N hydrochloride was added to acidify the solution to pH=2. After the addition of water and dichloromethane, the organic layer was separated and the aqueous layer was extracted with dichloromethane. The combined organic extracts were dried over magnesium sulfate, and concentrated *in vacuo*. The crude residue was purified by silica gel column chromatography using solvent system 30% ethyl acetate in hexane as eluent to give the title compound as a solid (70mg, 36%). ES-MS (M+H)+ = 477.

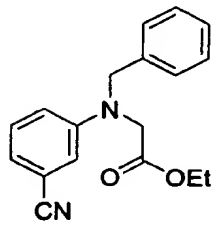
Example 31



A solution of the compound of example 30 (150mg, 0.32mmol), hydroxylamine hydrochloride (55mg, 0.79mmol) and triethylamine (110ul, 0.79mmol) in absolute

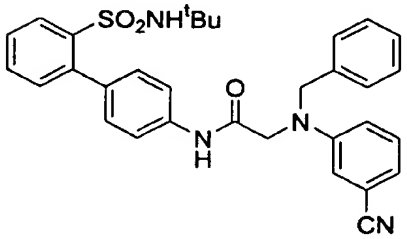
ethanol (5ml) was stirred at 40 °C for 15 hrs. After the evaporation of the solvent *in vacuo*, the residue was dissolved in acetic acid (3ml), and acetic anhydride (60ul, 0.64mmol) was added. The mixture was stirred at room temperature for 3 hrs. It was diluted with absolute methanol (5ml), and 10% Pd/C (catalytic amount) was added. The mixture was applied with 50psi hydrogen for 6 hrs. After the filtration through Celite to remove the catalyst, the filtrate was concentrated *in vacuo*. The crude residue was dissolved in trifluoroacetic acid (5ml). The mixture was stirred at room temperature for 4 hrs. After the evaporation of the solvent *in vacuo*, the crude residue was purified by RP-HPLC to give the title compound as a white powder (45mg, 79%). ES-MS (M+H)+ = 438.

Example 32



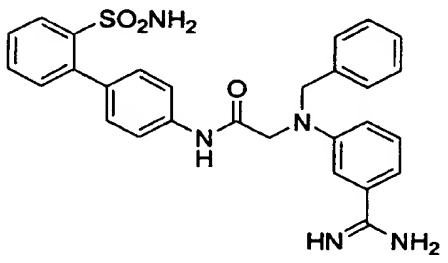
To a solution of the compound of example 28 (200mg, 1mmol) and cesium carbonate (650mg, 2mmol) in dimethylformamide (5ml) was added benzyl bromide (180ul, 1.5mmol). The mixture was stirred at 90 °C for 2 hr. After the filtration of the solid, the filtrate was concentrated *in vacuo* and the residue was purified by silica gel column chromatography using solvent system 10% ethyl acetate in hexane as eluent to give the title compound as an oil (210mg, 71%). ES-MS (M+H)+ = 295.

Example 33



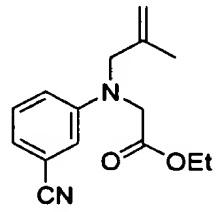
To a solution of the compound of [2-(4-aminophenyl)phenylsulfonyl](t-butyl)amine (126mg, 0.41mmol) in dichloromethane (5ml) was added 2.0M trimethylaluminum in hexane (0.62ml, 1.24mmol). The mixture was stirred at room temperature for 30 minutes, methane gas evolved. A solution of the compound of example 32 (120mg, 0.41mmol) in dichlodomethane (1ml) was added. The mixture was stirred at room temperature overnight. 1N hydrochloride was added to acidify the solution to pH=2. After the addition of water and dichloromethane, the organic layer was separated, and the aqueous layer was extracted with dichloromethane. The combined organic extracts were dried over magnesium sulfate, and concentrated *in vacuo*. The crude residue was purified by silica gel column chromatography using solvent system 20% ethyl acetate in hexane as eluent to give the title compound as a solid (172mg, 76%). ES-MS (M+H)+ = 553.

Example 34



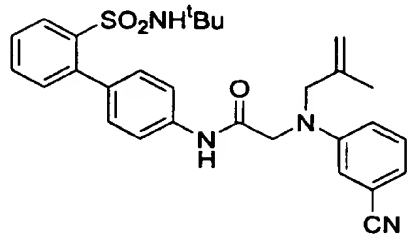
To a solution of the compound of example 33 (100mg, 0.18mmol) and absolute methanol (73ul, 1.8mmol) in ethyl acetate (3ml) in an ice bath was saturated with hydrochloride gas for 10 minutes. The mixture was stirred at room temperature for 3 hrs. After the evaporation of the solvent *in vacuo*, the residue was dissolved in absolute methanol (3ml), and ammonia acetate (83mg, 1.08mmol) was added. The mixture was refluxed for 3 hrs. The solvent was evaporated *in vacuo*. The crude residue was purified by RP-HPLC to give the title compound as white powder (25mg, 27%). ES-MS (M+H)+ = 514.

Example 35



To a solution of the compound of example 28 (200mg, 1mmol) and cesium carbonate (650mg, 2mmol) in dimethylformamide (5ml) was added 3-bromo-2-methyl-propene (121ul, 1.2mmol). The mixture was stirred at 90 °C for 2 hrs. After the filtration of the solid, the filtrate was concentrated *in vacuo*, and the residue was purified by silica gel column chromatography using solvent system 10% ethyl acetate in hexane as eluent to give the title compound (140mg, 54%). ES-MS (M+H)+ = 259.

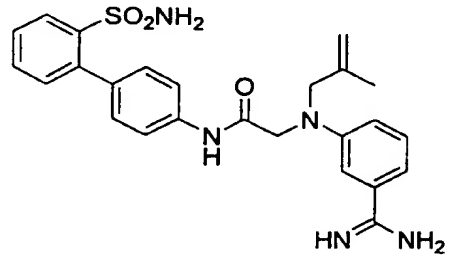
Example 36



To a solution of the compound of [2-(4-aminophenyl)phenylsulfonyl](t-butyl)amine (165mg, 0.54mmol) in dichloromethane (5ml) was added 2.0M trimethylaluminum in hexane (0.81ml, 1.63mmol). The mixture was stirred at room temperature for 30 minutes, methane gas evolved. A solution of the compound of example 35 (140mg, 0.54mmol) in dichlodomethane (1ml) was added. The mixture was stirred at room temperature overnight. 1N hydrochloride was added to acidify the solution to pH=2. After the addition of water and dichloromethane, the organic layer was separated and the aqueous layer was extracted with dichloromethane. The combined organic extracts were dried over magnesium sulfate, and concentrated *in vacuo*. The crude residue was purified by silica gel column chromatography using solvent system 30%

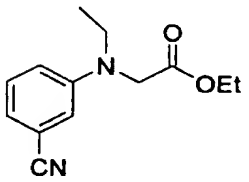
ethyl acetate in hexane as eluent to give the title compound as a solid (210mg, 75%). ES-MS (M+H)+ = 517.

Example 37



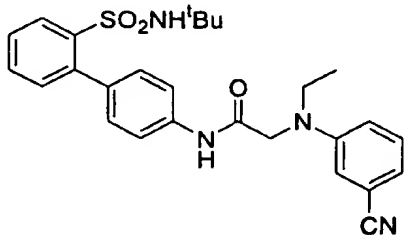
To a solution of the compound of example 36 (210mg, 0.41mmol) and absolute methanol (330ul, 8.1mmol) in ethyl acetate (3ml) in an ice bath was saturated with hydrochloride gas for 10 minutes. The mixture was stirred at room temperature for 3 hrs. After the evaporation of the solvent *in vacuo*, the residue was dissolved in absolute methanol (3ml), and ammonia acetate (190mg, 2.46mmol) was added. The mixture was refluxed for 3 hrs. The solvent was evaporated *in vacuo*. The crude residue was purified by RP-HPLC to give the title compound as white powder (12mg, 6%). ES-MS (M+H)+ = 478.

Example 38



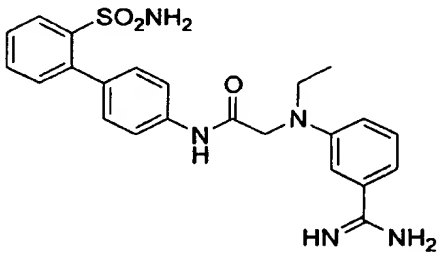
To a solution of the compound of example 28 (200mg, 1mmol) and cesium carbonate (650mg, 2mmol) in dimethylformamide (5ml) was added iodoethane (96ul, 1.2mmol). The mixture was stirred at 90 °C for 2 hrs. After the filtration of the solid, the filtrate was concentrated *in vacuo*, and the residue was purified by silica gel column chromatography using solvent system 20% ethyl acetate in hexane as eluent to give the title compound (80mg, 34%). ES-MS (M+H)+ = 233.

Example 39



To a solution of the compound of [2-(4-aminophenyl)phenylsulfonyl](t-butyl)amine (288mg, 0.95mmol) in dichloromethane (3ml) was added 2.0M trimethylaluminum in hexane (1.42ml, 2.84mmol). The mixture was stirred at room temperature for 30 minutes, methane gas evolved. A solution of the compound of example 38 (20mg, 0.95mmol) in dichlodomethane (1ml) was added. The mixture was stirred at room temperature overnight. 1N hydrochloride was added to acidify the solution to pH=2. After the addition of water and dichloromethane, the organic layer was separated and the aqueous layer was extracted with dichloromethane. The combined organic extracts were dried over magnesium sulfate, and concentrated *in vacuo*. The crude residue was purified by silica gel column chromatography using solvent system 30% ethyl acetate in hexane as eluent to give the title compound as a solid (243mg, 52%). ES-MS (M+H)+ = 491.

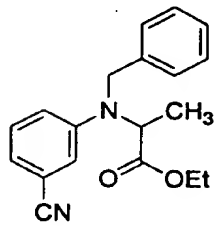
Example 40



A solution of the compound of example 39 (243mg, 0.50mmol), hydroxylamine hydrochloride (86mg, 1.24mmol) and triethylamine (173ul, 1.24mmol) in absolute ethanol (3ml) was stirred at 40 °C for 15 hrs. After the evaporation of the solvent *in vacuo*, the residue was dissolved in acetic acid (3ml), and acetic anhydride (94ul, 1mmol) was added. The mixture was stirred at room temperature for 3 hrs. It was

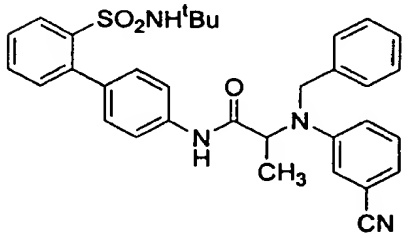
diluted with absolute methanol (5ml), and 10% Pd/C (catalytic amount) was added. The mixture was applied with 50psi hydrogen for 6 hrs. After the filtration through Celite to remove the catalyst, the filtrate was concentrated *in vacuo*. The crude residue was treated with trifluoroacetic acid (5ml) at room temperature for 2 hrs. After concentration *in vacuo*, the residue was purified by RP-HPLC to give the title compound as a white powder (104mg, 70%). ES-MS (M+H)+ = 452.

Example 41



To a solution of the compound of example 32 (00mg, 0.68mmol) in THF (3ml) was added LDA dropwise at –78 °C. The mixture was stirred at –78 °C for 30 minutes. Iodomethane was added at –78 °C, then warmed up to room temperature and stirred overnight. Saturated ammonium chloride was added to quench the reaction. The mixture was extracted with ethyl acetate and dried over magnesium sulfate. After concentration *in vacuo*, the crude residue was purified by silica gel column chromatography using solvent system 10% ethyl acetate in hexane as eluent to give title compound (70mg, 33%). ES-MS (M+H)+=309.

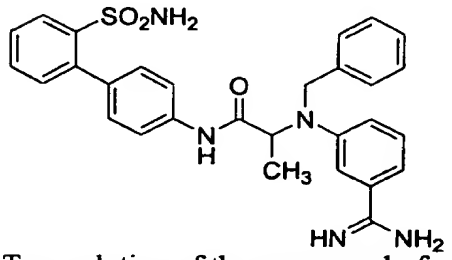
Example 42



To a solution of the compound of [2-(4-aminophenyl)phenylsulfonyl](t-butyl)amine (99mg, 0.32mmol) in dichloromethane (5ml) was added 2.0M trimethylaluminum in hexane (0.49ml, 0.97mmol). The mixture was stirred at room temperature for 30

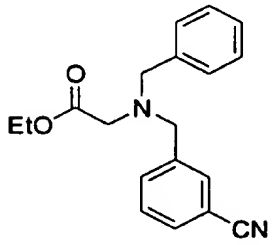
minutes, methane gas evolved. A solution of the compound of example 41 (100mg, 0.32mmol) in dichlodomethane (1ml) was added. The mixture was stirred at room temperature overnight. 1N hydrochloride was added to acidify the solution to pH=2. After the addition of water and dichloromethane, the organic layer was separated, and the aqueous layer was extracted with dichloromethane. The combined organic extracts were dried over magnesium sulfate, and concentrated *in vacuo*. The crude residue was purified by silica gel column chromatography using solvent system 30% ethyl acetate in hexane as eluent to give the title compound as a solid (100mg, 61%). ES-MS (M+H)+ = 567.

Example 43



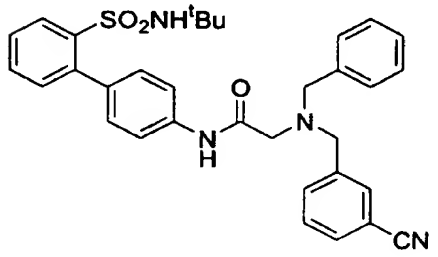
To a solution of the compound of example 42 (110mg, 0.19mmol) and absolute methanol (157ul, 3.89mmol) in ethyl acetate (3ml) in an ice bath was saturated with hydrochloride gas for 10 minutes. The mixture was stirred at room temperature for 3 hrs. After the evaporation of the solvent *in vacuo*, the residue was dissolved in absolute methanol (3ml), and ammonia acetate (88mg, 1.14mmol) was added. The mixture was refluxed for 3 hrs. The solvent was evaporated *in vacuo*. The crude residue was purified by RP-HPLC to give the title compound (68mg, 70%). ES-MS (M+H)+ = 528.

Example 44



To a solution of N-benzylglycine ethyl ester (1g, 5mmol) and cesium carbonate (4.05g, 2.4mmol) in dimethylformamide (5ml) was added 2-bromo-m-tolunitrile (1.22, 6.2mmol). The mixture was stirred overnight. After the filtration of the solid, the filtrate was concentrated *in vacuo*, and the residue was purified by silica gel column chromatography using solvent system 20% ethyl acetate in hexane as eluent to give the title compound (1.24g, 81%). ES-MS (M+H)+ = 309.

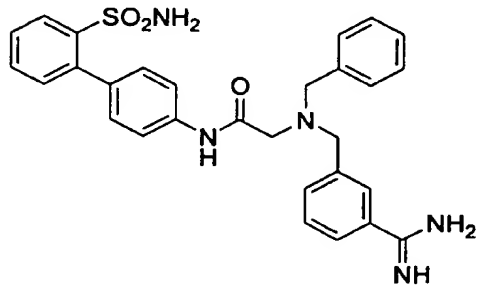
Example 45



To a solution of the compound of [2-(4-aminophenyl)phenylsulfonyl](t-butyl)amine (247mg, 0.81mmol) in dichloromethane (3ml) was added 2.0M trimethylaluminum in hexane (1.22ml, 2.44mmol). The mixture was stirred at room temperature for 30 minutes, methane gas evolved. A solution of the compound of example 44 (250mg, 0.81mmol) in dichlodomethane (1ml) was added. The mixture was stirred at room temperature overnight. 1N hydrochloride was added to acidify the solution to pH=2. After the addition of water and dichloromethane, the organic layer was separated and the aqueous layer was extracted with dichloromethane. The combined organic extracts were dried over magnesium sulfate, and concentrated *in vacuo*. The crude residue was purified by silica gel column chromatography using solvent system 30%

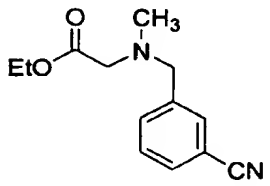
ethyl acetate in hexane as eluent to give the title compound as a solid (270mg, 59%). ES-MS (M+H)+ = 567.

Example 46



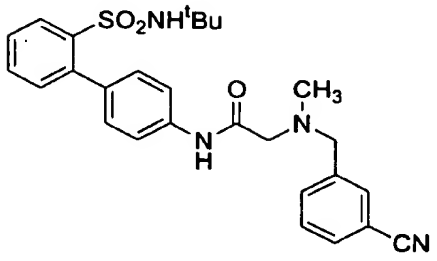
To a solution of the compound of example 45 (270mg, 0.48mmol) and absolute methanol (387ul, 9.54mmol) in ethyl acetate (3ml) in an ice bath was saturated with hydrochloride gas for 10 minutes. The mixture was stirred at room temperature for 3 hrs. After the evaporation of the solvent *in vacuo*, the residue was dissolved in absolute methanol (3ml), and ammonia acetate (222mg, 2.88mmol) was added. The mixture was refluxed for 3 hrs. The solvent was evaporated *in vacuo*. The crude residue was purified by RP-HPLC to give the title compound as white powder (200mg, 91%). ES-MS (M+H)+ = 528.

Example 47



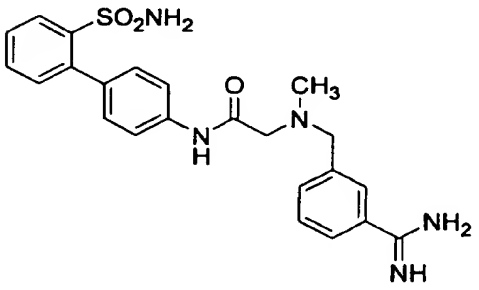
To a solution of sarcosine ethyl ester (1.54g, 10mmol) and cesium carbonate (7.82g, 24mmol) in dimethylformamide (10ml) was added 2-bromo-m-tolunitrile (2.35g, 12mmol). The mixture was stirred overnight. After the filtration of the solid, the filtrate was concentrated *in vacuo*, and the residue was purified by silica gel column chromatography using solvent system 20% ethyl acetate in hexane as eluent to give the title compound (1.77g, 76%). ES-MS (M+H)+ = 233.

Example 48



To a solution of the compound of [2-(4-aminophenyl)phenylsulfonyl](t-butyl)amine (262mg, 0.86mmol) in dichloromethane (3ml) was added 2.0M trimethylaluminum in hexane (1.29ml, 2.59mmol). The mixture was stirred at room temperature for 30 minutes, methane gas evolved. A solution of the compound of example 47 (200mg, 0.86mmol) in dichlodomethane (1ml) was added. The mixture was stirred at room temperature overnight. 1N hydrochloride was added to acidify the solution to pH=2. After the addition of water and dichloromethane, the organic layer was separated and the aqueous layer was extracted with dichloromethane. The combined organic extracts were dried over magnesium sulfate, and concentrated *in vacuo*. The crude residue was purified by silica gel column chromatography using solvent system 40% ethyl acetate in hexane as eluent to give the title compound (340mg, 81%). ES-MS (M+H)+ = 491.

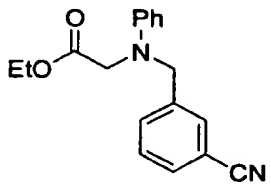
Example 49



To a solution of the compound of example 48 (260mg, 0.53mmol) and absolute methanol (679ul, 16.8mmol) in ethyl acetate (3ml) in an ice bath was saturated with

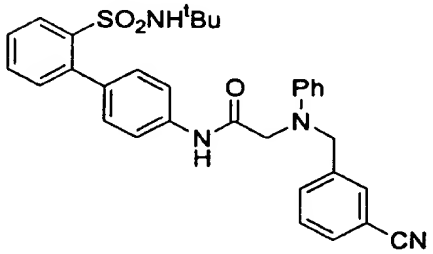
hydrochloride gas for 10 minutes. The mixture was stirred at room temperature for 3 hrs. After the evaporation of the solvent *in vacuo*, the residue was dissolved in absolute methanol (3ml), and ammonia acetate (245mg, 3.18mmol) was added. The mixture was refluxed for 3 hrs. The solvent was evaporated *in vacuo*. The crude residue was purified by RP-HPLC to give the title compound as white powder (193mg, 81%). ES-MS (M+H)+ = 452.

Example 50



To a solution of N-phenylglycine ethyl ester (1g, 5.58mmol) and cesium carbonate (4.37g, 13.4mmol) in dimethylformamide (10ml) was added 2-bromo-m-tolunitrile (1.31g, 6.7mmol). The mixture was stirred overnight. After the filtration of the solid, the filtrate was concentrated *in vacuo*, and the residue was purified by silica gel column chromatography using solvent system 20% ethyl acetate in hexane as eluent to give the title compound (1.2g, 73%). ES-MS (M+H)+ = 295.

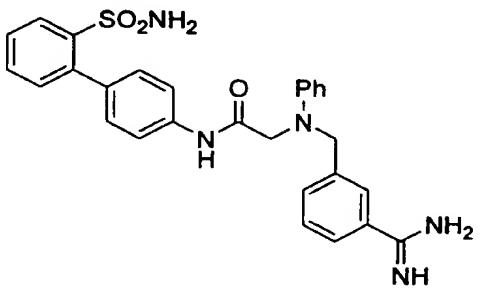
Example 51



To a solution of the compound of [2-(4-aminophenyl)phenylsulfonyl](t-butyl)amine (259mg, 0.85mmol) in dichloromethane (3ml) was added 2.0M trimethylaluminum in hexane (1.28ml, 2.55mmol). The mixture was stirred at room temperature for 30 minutes, methane gas evolved. A solution of the compound of example 50 (259mg, 0.85mmol) in dichlodomethane (1ml) was added. The mixture was stirred at room temperature overnight. 1N hydrochloride was added to acidify the solution to pH=2.

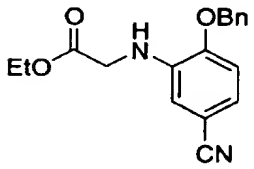
After the addition of water and dichloromethane, the organic layer was separated and the aqueous layer was extracted with dichloromethane. The combined organic extracts were dried over magnesium sulfate, and concentrated *in vacuo*. The crude residue was purified by silica gel column chromatography using solvent system 30% ethyl acetate in hexane as eluent to give the title compound as a solid (280mg, 60%). ES-MS (M+H)+ = 553.

Example 52



To a solution of the compound of example 51 (280mg, 0.51mmol) and absolute methanol (411ul, 10.1mmol) in ethyl acetate (3ml) in an ice bath was saturated with hydrochloride gas for 10 minutes. The mixture was stirred at room temperature for 3 hrs. After the evaporation of the solvent *in vacuo*, the residue was dissolved in absolute methanol (3ml), and ammonia acetate (236mg, 3.06mmol) was added. The mixture was refluxed for 3 hrs. The solvent was evaporated *in vacuo*. The crude residue was purified by RP-HPLC to give the title compound as white powder (150mg, 57%). ES-MS (M+H)+ = 514.

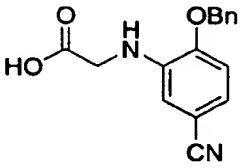
Example 53



To a solution of 3-amino-4-(phenylmethoxyl)benzonitrile (1.12g, 5mmol) and cesium carbonate (3.26g, 10mmol) in dimethylformamide (10ml) was added ethyl bromoactate (830ul, 7.5mmol). The mixture was stirred at room temperature for 2

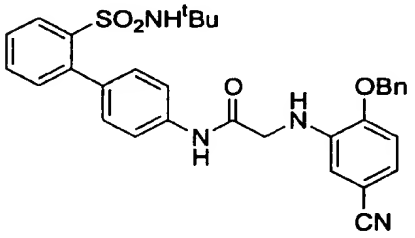
hrs. After the filtration of the solid, the filtrate was concentrated *in vacuo*, and the residue was purified by silica gel column chromatography using solvent system 15% ethyl acetate in hexane as eluent to give the title compound as an oil (1.33g, 85%). ES-MS (M+H)+ = 311.

Example 54



To a solution of the compound of example 53 (200mg, 0.64mmol) in methanol (2ml) was added 1N lithium hydroxide (1.28ml, 1.28mmol). The mixture was stirred at room temperature for 2 hr. After concentrated *in vacuo*, the residue was acidified by 1N hydrochloride to PH=2, and extracted with ethyl acetate. The organic layer was dried over magnesium sulfate and concentrated to give title compound (90mg, 50%). ES-MS (M+H)+=283.

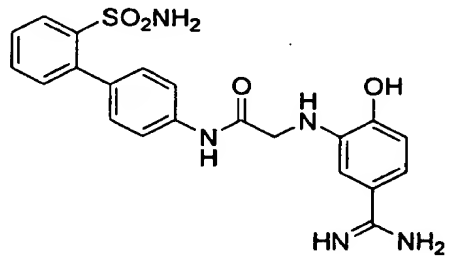
Example 55



To a solution of the compound of [2-(4-aminophenyl)phenylsulfonyl](t-butyl)amine (377mg, 1.24mmol) in DMF (5ml) was added the compound of example 55(350mg, 1.24mmol), BOP reagent (659mg, 1.49mmol) and triethylamine (346ul, 2.48mmol). The mixture was stirred at room temperature overnight. After the addition of water and dichloromethane, the organic layer was separated and the aqueous layer was extracted with dichloromethane. The combined organic extracts were dried over magnesium sulfate, and concentrated *in vacuo*. The crude residue was purified by

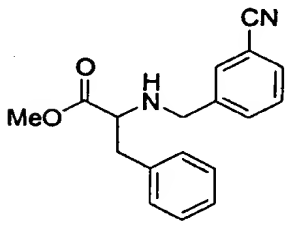
silica gel column chromatography using solvent system 30% ethyl acetate in hexane as eluent to give the title compound (470mg, 53%). ES-MS (M+H)+ = 569.

Example 56



To a solution of the compound of example 55 (200mg, 0.35mmol) and absolute methanol (285ul, 7.04mmol) in ethyl acetate (3ml) in an ice bath was saturated with hydrochloride gas for 10 minutes. The mixture was stirred at room temperature for 3 hrs. After the evaporation of the solvent *in vacuo*, the residue was dissolved in absolute methanol (3ml), and ammonia acetate (162mg, 2.1mmol) was added. The mixture was refluxed for 3 hrs. The solvent was evaporated *in vacuo*. The crude residue was dissolved in methanol. The mixture was applied with hydrogen balloon overnight. After concentrated in vacuo, the residue was purified by RP-HPLC to give the title compound as white powder (105mg, 61%). ES-MS (M+H)+ = 440.

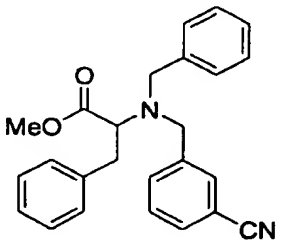
Example 57



To a solution of DL-phenylalanine methyl ester hydrochloride (1g, 4.6mmol) in dichloromethane (10ml) was added a solution of 3-cyanobenzaldehyde (0.61g, 4.6mmol) in dichloromethane (5ml) followed by acetic acid (2.65ml, 46mmol) and sodium triacetoxy-borohydrate (1.18g, 56mmol). The mixture was stirred at room temperature overnight. The mixture was washed with saturated sodium bicarbonate and sodium chloride solution. The orgaince layer was separated and dried over

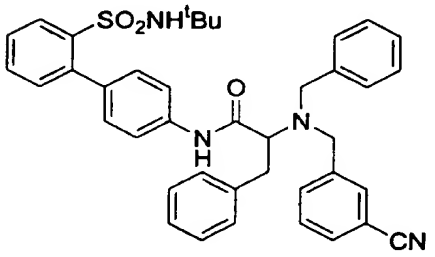
magnesium sulfate, concentrated *in vacuo* and purified by silica gel column using solvent system 30% ethyl acetate in hexane as eluent to give the title compound (1.46g, 54%). ES-MS (M+H)+=295.

Example 58



To a solution of the compound 57 (294mg, 1mmol) and cesium carbonate (650mg, 2mmol) in dimethylformamide (3ml) was added benzyl bromide (179ul, 1.5mmol). The mixture was stirred overnight. After the filtration of the solid, the filtrate was concentrated *in vacuo*, and the residue was purified by silica gel column chromatography using solvent system 20% ethyl acetate in hexane as eluent to give the title compound (290mg, 77%). ES-MS (M+H)+ = 385.

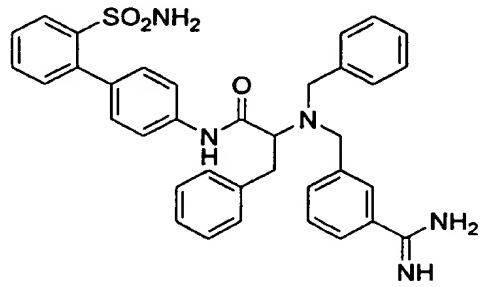
Example 59



To a solution of the compound of [2-(4-aminophenyl)phenylsulfonyl](t-butyl)amine (95mg, 0.31mmol) in dichloromethane (3ml) was added 2.0M trimethylaluminum in hexane (0.47ml, 0.94mmol). The mixture was stirred at room temperature for 30 minutes, methane gas evolved. A solution of the compound of example 58 (120mg, 0.31mmol) in dichlodomethane (1ml) was added. The mixture was stirred at room temperature overnight. 1N hydrochloride was added to acidify the solution to pH=2. After the addition of water and dichloromethane, the organic layer was separated and

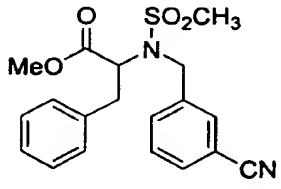
the aqueous layer was extracted with dichloromethane. The combined organic extracts were dried over magnesium sulfate, and concentrated *in vacuo*. The crude residue was purified by silica gel column chromatography using solvent system 30% ethyl acetate in hexane as eluent to give the title compound as a solid (120mg, 59%). ES-MS (M+H)+ = 657.

Example 60



To a solution of the compound of example 59 (190mg, 0.29mmol) and absolute methanol (235ul, 5.8mmol) in ethyl acetate (3ml) in an ice bath was saturated with hydrochloride gas for 10 minutes. The mixture was stirred at room temperature for 3 hrs. After the evaporation of the solvent *in vacuo*, the residue was dissolved in absolute methanol (3ml), and ammonia acetate (134mg, 1.74mmol) was added. The mixture was refluxed for 3 hrs. The solvent was evaporated *in vacuo*. The crude residue was purified by RP-HPLC to give the title compound as white powder (120mg, 67%). ES-MS (M+H)+ = 618.

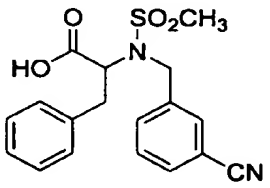
Example 61



To a solution of the compound of example 57 (200mg, 0.68mmol) in pyridine (2ml) was added methanesulfonyl chloride (63ul, 0.82mmol) dropwise at 0 °C. The mixture was stirred at room temperature overnight. After the concentration *in vacuo*, the residue was dissolved in ethyl acetate and washed with 0.1N hydrochloride. The

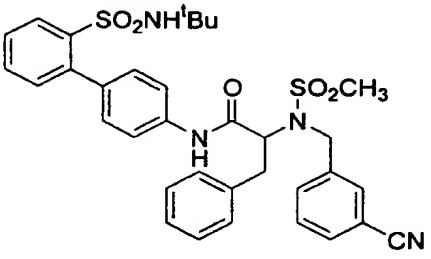
organic layer was dried over magnesium sulfate, concentrated *in vacuo* and purified by silica gel column chromatography using solvent system 30% ethyl acetate in hexane as eluent to give title compound (190mg, 75%). ES-MS (M+H)+=373.

Example 62



To a solution of the compound of example 61 (190mg, 0.51mmol) in methanol (2ml) was added 1N lithium hydroxide (1.02ml, 1.02mmol). The mixture was stirred at room temperature for 2 hr. After concentrated *in vacuo*, the residue was acidified by 1N hydrochloride to PH=2, and extracted with ethyl acetate. The organic layer was dried over magnesium sulfate and concentrated to give title compound (220mg, 100%). ES-MS (M+H)+=359.

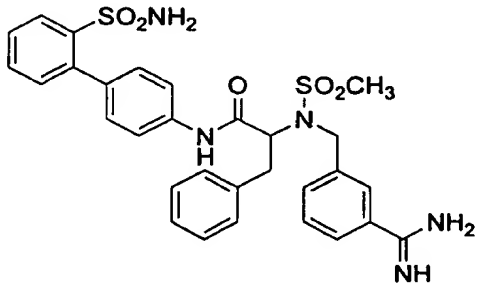
Example 63



To a solution of the compound of [2-(4-aminophenyl)phenylsulfonyl](t-butyl)amine (187mg, 0.61mmol) in DMF (5ml) was added the compound of example 62 (220mg, 0.61mmol), BOP reagent (326mg, 0.74mmol) and triethylamine (171ul, 1.23mmol). The mixture was stirred at room temperature overnight. After the addition of water and dichloromethane, the organic layer was separated and the aqueous layer was extracted with dichloromethane. The combined organic extracts were dried over magnesium sulfate, and concentrated *in vacuo*. The crude residue was purified by

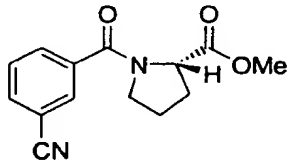
silica gel column chromatography using solvent system 30% ethyl acetate in hexane as eluent to give the title compound (230mg, 59%). ES-MS (M+H)+ = 645.

Example 64



To a solution of the compound of example 63 (230mg, 0.36mmol) and absolute methanol (290ul, 7.14mmol) in ethyl acetate (3ml) in an ice bath was saturated with hydrochloride gas for 10 minutes. The mixture was stirred at room temperature for 3 hrs. After the evaporation of the solvent *in vacuo*, the residue was dissolved in absolute methanol (3ml), and ammonia acetate (167mg, 2.16mmol) was added. The mixture was refluxed for 3 hrs. The solvent was evaporated *in vacuo*. The crude residue was purified by RP-HPLC to give the title compound as white powder (78mg, 36%). ES-MS (M+H)+ = 606.

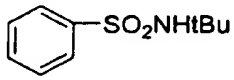
Example 65



H-Pro-OMe (3.38 g, 20.4 mmol) and 3-cyano-benzoic acid (3 g, 20.4 mmol) were dissolved in DMF (100 mL). DIEA (7.28 mL, 40.8 mmol) was added followed by the addition of the coupling reagent BOP (9.03 g, 20.4 mmol). The solution was stirred at room temperature for 12 hours. The reaction mixture was diluted in a mixture of EtOAc/H_2O (100 mL:40 mL). The organic layer was washed with water, sat. $NaHCO_3$, water, brine, dried over $MgSO_4$, filtered and solvent evaporated. The

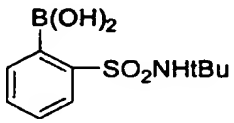
residue was purified by silica gel column chromatography using solvent system 20% hexane in EtOAc as eluant to give the title compound. ES-MS (M+H)+ = 259.0.

Example 66



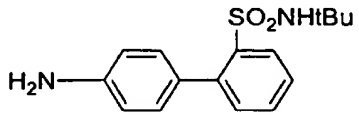
To a solution of tert-Butylamine (41.4g, 566 mmol) and triethylamine (118 mL, 849 mmol) in DCM (1000 mL) in an ice bath, was added benzenesulfonyl chloride (100 g, 566 mmol) dropwise. The mixture was stirred at room temperature overnight. Water was added to the mixture and organic layer was washed with water, brine, dried over Na_2SO_4, filtered and filtrated evaporated in vacuo to give the title compound as light yellowish solid (117.63 g, 97.6%). (M+H)+ = 214.1.

Example 67



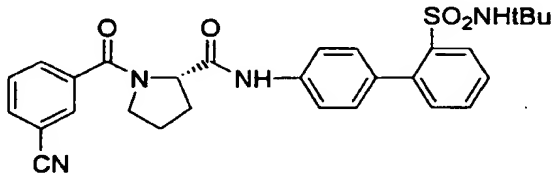
To a solution of compound of example 66 (53.25 g, 250 mmol) in THF (600 mL) in an ice bath, was added n-butyllithium in hexane (200 mL, 500 mmol) dropwise. A thick precipitate was formed when the reaction mixture was warmed up to 10°C. Triisopropylborate was added keeping the temperature below 35°C. After 1 hr., the mixture was cooled in an ice bath, 1N HCl (405 mL) was added, and the mixture was stirred overnight. The mixture was extracted with ether (100 mL) three times. The combined organic extracts were extracted with 1N NaOH (130 mL) three times. The aqueous extracts were acidified to pH 1 with 12 N HCl, and then extracted with ether three times (140 ML). The combined ether extracts were dried over $MgSO_4$, and solvents evaporated *in vacuo*. Hexane and ether were added and a white precipitate formed. The solid was collected and washed with 10% ether/hexane to give the title compound. (M+H)+ = 257.1.

Example 68



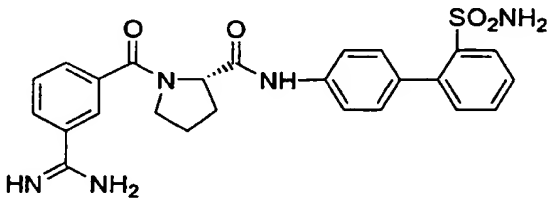
To a solution of compound of example 67 (6.4 g, 25 mmol) in toluene (120 mL) was added water (15 mL), 5N NaOH solution (38.5 mL), isopropanol (60 mL), 4-bromoaniline and tetrakis(triphenylphosphine) palladium(0). The mixture was refluxed for six hours, cooled to room temperature, diluted with EtOAc. The organic layer was washed with water, dried with $MgSO_4$, filtered and concentrated. This was purified by silica gel column chromatography using solvent system 30% EtOAc in hexane as eluant to give the title compound (5g, 66%). ES-MS (M+H)+ = 305.1.

Example 69



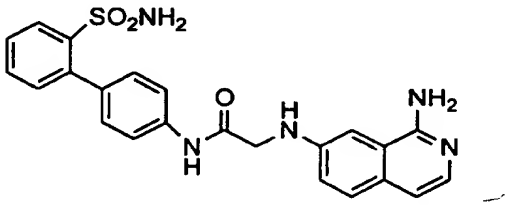
To a solution of compound of example 68 (278 mg, 0.92 mmol) in DCM (5 mL) was added trimethylaluminum (1.37 mL, 2 M in hexane) dropwise. The reaction mixture was stirred at room temperature for 30 min. Compound of example 17 (236 mg, 0.92 mmol) in DCM (3 mL) was added dropwise. The mixture was stirred at room temperature overnight. 2N HCl was added to PH 2 to neutralize excess $AlMe_3$. Water and DCM were added. The organic layer was dried over $MgSO_4$ and concentrated in vacuo. The obtained resudue was purified by silica gel column chromatography using solvent system 50% EtOAc in hexane as eluant to give the title compound. ES-MS (M+Na)+ =553.2.

Example 70

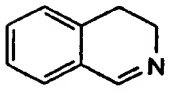


A solution of the compound of example 69 (96 mg, 0.18 mmol) in MeOH (3 mL) was treated with a stream of HCl gas for 10 min. at 0°C. The resulting solution was capped, stirred at room temperature overnight and evaporated *in vacuo*. The residue was reconstituted in MeOH (3 mL) and the mixture was treated with NH_4OAc (69 mg, 0.9 mmol). The reaction mixture was refluxed for 1.5 hrs. and concentrated *in vacuo*. The obtained residue was purified by RP-HPLC to give the title compound as a white powder. ES-MS (M+H)+ = 492.0

Example 71



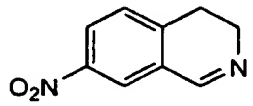
Step 1: Synthesis of:



To the solution of 1,2,3,4-tetrahydroisoquinoline (10 g, 0.075 mol) in dichloromethane (10 ml), was added N-bromosuccinamide (20 g, 0.1126 mol) portion wise. Gas was generated accomplished with heat, the color changed to dark reddish. Stirred at room temperature under argon for 30 min. Reaction was complete. To the reaction mixture was carefully added 30% sodium hydroxide aquous solution (50 ml). The mixture was stirred at room temperature overnight. More water was added. The organic layer was seperated and treated with 4N HCl aquous solution (200 ml). The aquous layer was seperated and basified with ammonium hydroxide (28%, 100 ml). The suspension was extracted with

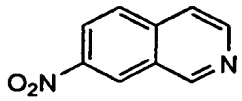
dichloromethane (2x200 ml). The organics were dried over anhydrous MgSO4, filtered and concentrated. The brown oil crude product (8.5 g) was distilled at 0.07 mmHg at 41~49 0C to give colorless oil as the title compound in a yield of 86.7%.

Step 2. Synthesis of:



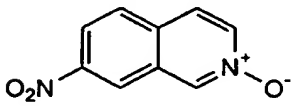
To the colorless oil of compound of step 1 (5.09 g, 0.039 mol), was added sulfuric acid (20 ml). The mixture was added to the solution of potassium nitrate (4.15 g, 0.041 mol) in sulfuric acid (20 ml) at 0 0C. The yellow green solution was stirred at 0 0C for 1 hr, at room temperature for another 2 hr.. Heated to 70 0C for over night. To the reaction mixture was added ice (100 g), ammonium hydroxide solution (28%) cautiously. Brown solid precipated when PH>9. Filtered. The filtercake was vacuum dry to give 1.3 g of desired title compound. The filtrate was extracted with ethyl acetate (2x200 ml). The organics were combined and dried over anhydrous MgSO4, filtered, concentrated to give 5.5 g of the title compound as reddish solid. The total yield was about 100% (6.8 g).

Step 3. Synthesis of:



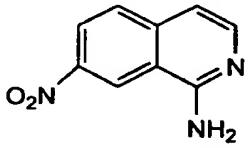
The reaction mixture of compound of step 2 (5.5 g, 0.03 mol) and palladium black (1.4 g, 0.013 mol) in decalin (100 ml) was heated to reflux under argon for 3.5 hr.. After sitting at the room temperature over night, the mixture was filtered through celite, washed with chloroform (200 ml). The filtrate was extrated with 2 N HCl solution (2x200 ml). The acidic aquous layers were basified by potassium hydroxide until PH>9. Extracted with dichloromethane (200 ml) and ethyl acetate (200 ml). The organics were combined and concentrated. Prep. HPLC purification afforded pure title compound in a yield of 59% (3.18 g)

Step 4. Synthesis of :



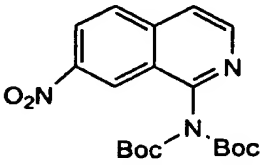
To the solution of 7-nitro isoquinoline (0.8 g, 0.0046 mol) in acetone (10 ml) was added MCPBA (0.952 g, 0.0055 mol) at room temperature. The reaction was complete after over night, concentrated to driness. Diluted with dichloromethane (100 ml), washed with sodium bicarbonate solution (100 ml) and sodium chloride solution (100 ml). The organic portion was dried over Mg2SO4, filtered, concentrated to give yellow solid of title compound in a yield of 64% (0.56 g).

Step 5. Synthesis of:



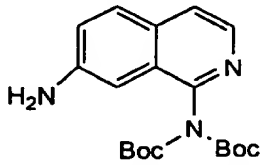
To the solution of compound of step 4 (0.56 g, 0.0027 mol) in pyridine (20 ml), was added tosyl chloride (0.618g, 0.00324 mol). The brown solution was stirred at room temperature for 2 hr. Concentrated to driness. To the reddish syrup was added ethanolamine (40 ml), the reaction was stirred at room temperature for 6 hr. Diluted with dichloromethane (200 ml), washed with water (100 ml), and sodium chloride solution (100 ml). The organic part was dried over anhydrous MgSO4, filtered, concentrated to give the title compound in a yield of 85% (0.457 g).

Step 6. Synthesis of:



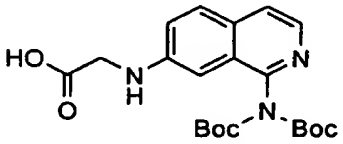
To the solution of the compound of step 5 (0.44 g, 0.0023 mol) in acetonitrile (20 ml) and chloroform (10 ml) was added 4-Dimethylaminopyridine (0.04 g, 0.0003 mol), Di-tert-butyldicarbonate (1.52 g, 0.007 mol) and triethylamine (1.28 ml, 0.0092 mol). The brown solution was stirred at room temperature for 4 hr. Not complete. Sitting in the refridgerator for 15 days, HPLC showed complete reaction. Prep HPLC purification afforded yellow solid of title compound in a yield of 58.3% (0.522 g).

Step 7. Synthesis of:



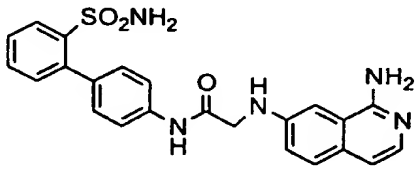
The mixture of the compound of step 6 (0.522 g, 0.00134 mol) and Pd/C (10%Wt, 100 mg) in methanol (10 ml) was stirred under hydrogen (1 atm) over night. Filtered through celite, washed with methanol, concentrated to give yellow solid of title compound in a yield of 86% (0.412 g).

Step 8. Synthesis of:



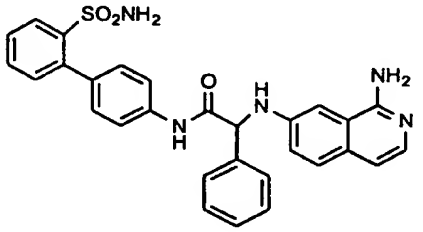
To the solution of the compound of step 7 (66.5 mg, 0.185 mmol) and glyoxylic acid (13.68 mg, 0.185 mmol) in dichloromethane (2 ml), was added acidic acid (0.1 ml, 1.85 mmol). The brown solution was stirred under argon for 10 min. NaBH(OAc)3 (59 mg, 0.278 mmol) was added at room temperature. Gas bubble was generated, the color changed to yellow. The reation completed after 3 hr. Purification via prep. HPLC gave title compound in a yield of 16% (12 mg).

Step 9. Synthesis of:



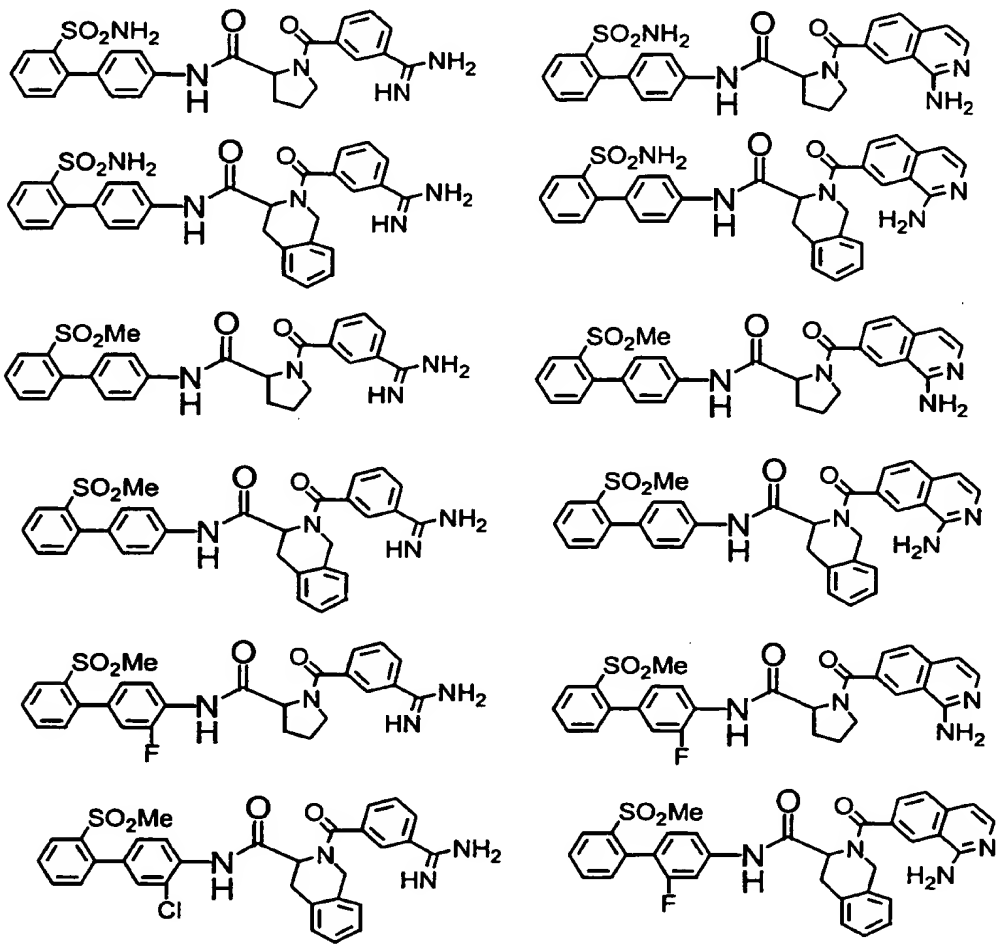
To the solution of the compound of step 8 (12 mg, 0.038 mmol) and 4-{2-{[(tert-butyl)amino]sulfonyl}- phenyl} benzoic acid (11.5 mg, 0.038 mmol) in DMF (2ml), was added diethylamine (0.02 ml, 0.114 mmol) followed by BOP (20 mg, 0.046 mmol). The yellow mixture was stirred at room temperature under argon for 1.5 hr. Diluted with ethyl acetate (20 ml), washed with sodium bicarbonate solution (2x10 ml). The organic part was dried over MgSO4, filtered and concentrated to give yellow syrup as crude product. It was subjected to TFA (10 ml) for over night. TFA was removed by reduced pressure evaporation. Prep HPLC purification afforded the title compound in a yield of 9% (1.5 mg). ES-MS (M+H)+ = 448

According to the procedures for the synthesis of the above compounds, the following compounds were also synthesized.



ES-MS $(M+H)^+ = 524.3$

By following the similar procedures, the following compounds were also synthesized.



BIOLOGICAL ACTIVITY EXAMPLES

Evaluation of the compounds of this invention is guided by in vitro protease activity assays (see below) and in vivo studies to evaluate antithrombotic efficacy, and effects on hemostasis and hematological parameters.

The compounds of the present invention are dissolved in buffer to give solutions containing concentrations such that assay concentrations range from 0 to 100 μM. In the assays for thrombin, prothrombinase and factor Xa, a synthetic

chromogenic substrate is added to a solution containing test compound and the enzyme of interest and the residual catalytic activity of that enzyme is determined spectrophotometrically. The IC_{50} of a compound is determined from the substrate turnover. The IC_{50} is the concentration of test compound giving 50% inhibition of the substrate turnover. The compounds of the present invention desirably have an IC_{50} of less than 500 nM in the factor Xa assay, preferably less than 200 nM, and more preferred compounds have an IC_{50} of about 100 nM or less in the factor Xa assay. The compounds of the present invention desirably have an IC_{50} of less than 4.0 μM in the prothrombinase assay, preferably less than 200 nM, and more preferred compounds have an IC_{50} of about 10 nM or less in the prothrombinase assay. The compounds of the present invention desirably have an IC_{50} of greater than 1.0 μM in the thrombin assay, preferably greater than 10.0 μM, and more preferred compounds have an IC_{50} of greater than 100.0 μM in the thrombin assay.

Amidolytic Assays for determining protease inhibition activity

The factor Xa and thrombin assays are performed at room temperature, in 0.02 M Tris·HCl buffer, pH 7.5, containing 0.15 M NaCl. The rates of hydrolysis of the para-nitroanilide substrate S-2765 (Chromogenix) for factor Xa, and the substrate Chromozym TH (Boehringer Mannheim) for thrombin following preincubation of the enzyme with inhibitor for 5 minutes at room temperature, and were determined using the Softmax 96-well plate reader (Molecular Devices), monitored at 405 nm to measure the time dependent appearance of p-nitroaniline.

The prothrombinase inhibition assay is performed in a plasma free system with modifications to the method described by Sinha, U. *et al.*, Thromb. Res., 75, 427-436 (1994). Specifically, the activity of the prothrombinase complex is determined by measuring the time course of thrombin generation using the p-

nitroanilide substrate Chromozym TH. The assay consists of preincubation (5 minutes) of selected compounds to be tested as inhibitors with the complex formed from factor Xa (0.5 nM), factor Va (2 nM), phosphatidyl serine:phosphatidyl choline (25:75, 20 μM) in 20 mM Tris·HCl buffer, pH 7.5, containing 0.15 M NaCl, 5 mM $CaCl_2$ and 0.1% bovine serum albumin. Aliquots from the complex-inhibitor mixture are added to prothrombin (1 nM) and Chromozym TH (0.1 mM). The rate of substrate cleavage is monitored at 405 nm for two minutes. Eight different concentrations of inhibitor are assayed in duplicate. A standard curve of thrombin generation by an equivalent amount of untreated complex are used for determination of percent inhibition.

Antithrombotic Efficacy in a Rabbit Model of Venous Thrombosis

A rabbit deep vein thrombosis model as described by Hollenbach, S. et al., Thromb. Haemost. 71, 357-362 (1994), is used to determine the in-vivo antithrombotic activity of the test compounds. Rabbits are anesthetized with I.M. injections of Ketamine, Xylazine, and Acepromazine cocktail. A standardized protocol consists of insertion of a thrombogenic cotton thread and copper wire apparatus into the abdominal vena cava of the anesthetized rabbit. A non-occlusive thrombus is allowed to develop in the central venous circulation and inhibition of thrombus growth is used as a measure of the antithrombotic activity of the studied compounds. Test agents or control saline are administered through a marginal ear vein catheter. A femoral vein catheter is used for blood sampling prior to and during steady state infusion of test compound. Initiation of thrombus formation begins immediately after advancement of the cotton thread apparatus into the central venous circulation. Test compounds are administered from time = 30 min to time = 150 min at which the experiment is terminated. The rabbits are euthanized and the thrombus excised by surgical dissection and characterized by weight and histology. Blood samples are analyzed for changes in hematological and coagulation parameters.

Effects of Compounds in Rabbit Venous Thrombosis model

Administration of compounds in the rabbit venous thrombosis model demonstrates antithrombotic efficacy at the higher doses evaluated. There are no significant effects of the compound on the aPTT and PT prolongation with the highest dose (100 μg/kg + 2.57 μg/kg/min). Compounds have no significant effects on hematological parameters as compared to saline controls. All measurements are an average of all samples after steady state administration of vehicle or (D)-Arg-Gly-Arg-thiazole. Values are expressed as mean ± SD.

Without further description, it is believed that one of ordinary skill in the art can, using the preceding description and the following illustrative examples, make and utilize the compounds of the present invention and practice the claimed methods.

WHAT IS CLAIMED IS:

1. A compound according to the formula I:

A-Y-D-E-G-J-Z-L

wherein:

A is selected from:

(a) C_1-C_6-alkyl;

(b) C_3-C_8-cycloalkyl;

(c) $-NR^2R^3$, $R^3C(=NR^2)-$, $R^2R^3N-C(=NR^2)-$, $R^2R^3N-C(=NR^2)-N(R^3)-$;

(d) phenyl, which is independently substituted with 0-2 R^1 substituents;

(e) naphthyl, which is independently substituted with 0-2 R^1 substituents;and

(f) a monocyclic or fused bicyclic heterocyclic ring system having from 5 to 10 ring atoms, wherein 1-4 ring atoms of the ring system are selected from N, O and S, and wherein the ring system may be subsituted from 0-2 R^1 substituents;

R^1 is selected from:

Halo, $R^2-C(=NR^3)-$, $R^2R^3N-C(=NR^2)-$, C_{1-4}alkyl, C_{2-6}alkenyl, C_{2-6}alkynyl, C_{3-8}cycloalkyl, C_{0-4}alkylC_{3-8}cycloalkyl,-CN, $-NO_2$, $(CH_2)_mNR^2R^3$, $SO_2NR^2R^3$, SO_2R^2, CF_3, OR^2, and a 5-6 membered aromatic heterocyclic system containing from 1-4 heteroatoms selected from N, O and S, wherein from 1-4 hydrogen atoms on the aromatic heterocyclic system may be independently replaced with a member selected from the group consisting of halo, C_1-C_4-alkyl, -CN C_{1-4}alkyl, C_{2-6}alkenyl, C_{2-6}alkynyl, C_{3-8}cycloalkyl, C_{0-4}alkylC_{3-8}cycloalkyl and $-NO_2$;

R^2 and R^3 are independently selected from the group consisting of:

H, C_{1-4}alkyl, C_{2-6}alkenyl, C_{2-6}alkynyl, C_{3-8}cycloalkyl, C_{0-4}alkylC_{3-8}cycloalkyl, C_{0-4}alkylphenyl and C_{0-4}alkylnaphthyl, wherein from 1-4 hydrogen atoms on the ring atoms of the phenyl and naphthyl moieties may be independently

replaced with a member selected from the group consisting of halo, C_{1-4}alkyl, C_{2-6}alkenyl, C_{2-6}alkynyl, C_{3-8}cycloalkyl, C_{0-4}alkylC_{3-8}cycloalkyl, -CN, and $-NO_2$;

m is an integer of 0-2;

Y is a member selected from the group consisting of:

a direct link, -CH2-, -C(=O)-, -N(R^4)-, -N(R^4)CH2-, -C=N(R4)-, -C(=O)-N(R^4)-, -N(R^4)-C(=O)-, $-SO_2$-, -O-, $-SO_2$-N(R^4)- and –N(R^4)-SO_2-;

R^4 is selected from:

H, C_{1-4}alkyl, C_{2-6}alkenyl, C_{2-6}alkynyl, C_{3-8}cycloalkyl, C_{0-4}alkylC_{3-8}cycloalkyl, C_{0-4}alkylphenyl and C_{0-4}alkylnaphthyl, wherein from 1-4 hydrogen atoms on the ring atoms of the phenyl and naphthyl moieties may be independently replaced with a member selected from the group consisting of halo, C_{1-4}alkyl, C_{2-6}alkenyl, C_{2-6}alkynyl, C_{3-8}cycloalkyl, C_{0-4}alkylC_{3-8}cycloalkyl, -CN, and $-NO_2$;.

D is a direct link or is a member selected from the group consisting of:

(a) phenyl, which is independently substituted with 0-2 R^{1a} substituents;

(b) naphthyl, which is independently substituted with 0-2 R^{1a} substituents; and

(c) a monocyclic or fused bicyclic heterocyclic ring system having from 5 to 10 ring atoms, wherein 1-4 ring atoms of the ring system are selected from N, O and S, and wherein the ring system may be subsituted from 0-2 R^{1a} substituents;

R^{1a} is selected from:

Halo, C_{1-4}alkyl, C_{2-6}alkenyl, C_{2-6}alkynyl, C_{3-8}cycloalkyl, C_{0-4}alkylC_{3-8}cycloalkyl, -CN, $-NO_2$, $(CH_2)_mNR^{2a}R^{3a}$, $SO_2NR^{2a}R^{3a}$, SO_2R^{2a}, CF_3, OR^{2a}, and a 5-6 membered aromatic heterocyclic system containing from 1-4 heteroatoms selected from N, O and S, wherein from 1-4 hydrogen atoms on the aromatic heterocyclic system may be independently replaced with a member selected from the group consisting of halo, C_{1-4}alkyl, C_{2-6}alkenyl, C_{2-6}alkynyl, C_{3-8}cycloalkyl, C_{0-4}alkylC_{3-8}cycloalkyl, -CN and $-NO_2$;

R^{2a} and R^{3a} are independently selected from the group consisting of:

H, C_{1-4}alkyl, C_{2-6}alkenyl, C_{2-6}alkynyl, C_{3-8}cycloalkyl, C_{0-4}alkylC_{3-8}cycloalkyl, C_{0-4}alkylphenyl and C_{0-4}alkylnaphthyl, wherein from 1-4 hydrogen atoms on the ring atoms of the phenyl and naphthyl moieties may be independently replaced with a member selected from the group consisting of halo, C_{1-4}alkyl, C_{2-6}alkenyl, C_{2-6}alkynyl, C_{3-8}cycloalkyl, C_{0-4}alkylC_{3-8}cycloalkyl, -CN and $-NO_2$;.

E is a member selected from the group consisting of:

$-N(R^5)-C(=O)-$, $-C(=O)-N(R^5)-$, $-N(R^5)-C(=O)-N(R^6)-$, $-SO_2-N(R^5)-$, $-N(R^5)-SO_2-N(R^6)-$ and $-N(R^5)-SO_2-N(R^6)-C(=O)-$;

R^5 and R^6 are independently selected from:

H, C_{1-4}alkyl, C_{2-6}alkenyl, C_{2-6}alkynyl, C_{3-8}cycloalkyl, C_{0-4}alkylC_{3-8}cycloalkyl, C_{0-4}alkylphenyl, C_{0-4}alkylnaphthyl, C_{0-4}alkylheteroaryl, C_{1-4}alkylCOOH and C_{1-4}alkylCOOC_{1-4}alkyl, wherein from 1-4 hydrogen atoms on the ring atoms of the phenyl, naphthyl and heteroaryl moieties may be independently replaced with a member selected from the group consisting of halo, C_{1-4}alkyl, C_{2-6}alkenyl, C_{2-6}alkynyl, C_{3-8}cycloalkyl, C_{0-4}alkylC_{3-8}cycloalkyl, -CN and $-NO_2$;

G is selected from:

$-CR^7R^8-$ and $-CR^{7a}R^{8a}-CR^{7b}R^{8b}-$

wherein R^7, R^8, R^{7a}, R^{8a}, R^{7b} and R^{8b} are independently a member selected from from the group consisting of:

hydrogen, C_{1-4}alkyl, C_{2-6}alkenyl, C_{2-6}alkynyl, C_{3-8}cycloalkyl, C_{0-4}alkyl-C_{3-8}cycloalkyl, C_{0-4}alkylphenyl, C_{0-4}alkylnaphthyl $-C_{0-4}$alkylCOOR^9, $-C_{0-4}$alkylC(=O)NR^9R^{10}, $-C_{0-4}$alkylC(=O)N$R^9-CH_2-CH_2-O-R^{10}$, $-C_{0-4}$alkylC(=O)N$R^9(-CH_2-CH_2-O-R^{10}-)_2$, $-N(R^9)COR^{10}$, $-N(R^9)C(=O)R^{10}$, $-N(R^9)SO_2R^{10}$, and a naturally occurring or synthetic amino acid side chain, wherein from 1-4 hydrogen atoms on the ring atoms of the phenyl and naphthyl moieties may be independently replaced with a member selected

from the group consisting of halo, C_{1-4}alkyl, C_{2-6}alkenyl, C_{2-6}alkynyl, C_{3-8}cycloalkyl, C_{0-4}alkyl-C_{3-8}cycloalkyl, -CN and -NO_2;

R^9 and R^{10} are independently selected from:

H, C_{1-4}alkyl, C_{0-4}alkylphenyl and C_{0-4}alkylnaphthyl, wherein from 1-4 hydrogen atoms on the ring atoms of the phenyl and naphthyl moieties may be independently replaced with a member selected from the group consisting of halo, C_{1-4}alkyl, C_{2-6}alkenyl, C_{2-6}alkynyl, C_{3-8}cycloalkyl, C_{0-4}alkyl-C_{3-8}cycloalkyl, -CN and -NO_2, and wherein R^9 and R^{10} taken together can form a 5-8 membered heterocylic ring;

J is a member selected from the group consisting of:

a direct link, -C(=O)-N(R^{11})-$(CH_2)_{0-2}$, -N(R^{11})-$(CH_2)_{0-2}$-C(=O)-, and -N(R^{11})-$(CH_2)_{0-2}$;

R^{11} is a member selected from the group consisting of:

hydrogen, C_{1-4}alkyl, C_{2-6}alkenyl, C_{2-6}alkynyl, C_{3-8}cycloalkyl, C_{0-4}alkyl-C_{3-8}cycloalkyl, C_{0-4}alkylphenyl, C_{0-4}alkylnaphthyl, C_{0-4}alkylheterocyclic ring having from 1 to 4 hetero ring atoms selected from the group consisting of N, O and S, CH_2COOC_{1-4}alkyl, CH_2COOC_{1-4}alkylphenyl and CH_2COOC_{1-4}alkylnaphthyl;

G and J can form a cyclic ring structure.

Z is a member selected from the group consisting of:

(a) phenyl, which is independently substituted with 0-2 R^{1b} substituents;

(b) naphthyl, which is independently substituted with 0-2 R^{1b} substituents; and

(c) a monocyclic or fused bicyclic heterocyclic ring system having from 5 to 10 ring atoms, wherein 1-4 ring atoms of the ring system are selected from N, O and S, and wherein the ring system may be subsituted from 0-2 R^{1b} substituents;

R^{1b} is selected from:

Halo, C_{1-4}alkyl, C_{2-6}alkenyl, C_{2-6}alkynyl, C_{3-8}cycloalkyl, C_{0-4}alkylC_{3-8}cycloalkyl, -CN, $-NO_2$, $NR^{2b}R^{3b}$, $SO_2NR^{2b}R^{3b}$, SO_2R^{2b}, CF_3, OR^{2b}, O-CH_2-OPh, O-CH_2-Ph, O-CH_2-CH_2-OR^{2b}, O-CH_2-$COOR^{2b}$, N(R^{2b})-CH_2-CH_2-OR^{2b}, N(-CH_2-CH_2-OR^{2b})$_2$, N(R^{2b})-C(=O)R^{3b}, N(R^{2b})-SO_2-R^{3b}, and a 5-6 membered aromatic heterocyclic system containing from 1-4 heteroatoms selected from N, O and S, wherein from 1-4 hydrogen atoms on the aromatic heterocyclic system may be independently replaced with a member selected from the group consisting of halo, C_{1-4}alkyl, C_{2-6}alkenyl, C_{2-6}alkynyl, C_{3-8}cycloalkyl, C_{0-4}alkylC_{3-8}cycloalkyl, -CN and $-NO_2$;

R^{2b} and R^{3b} are independently selected from the group consisting of:

H, C_{1-4}alkyl, C_{2-6}alkenyl, C_{2-6}alkynyl, C_{3-8}cycloalkyl, C_{0-4}alkylC_{3-8}cycloalkyl, C_{0-4}alkylphenyl and C_{0-4}alkylnaphthyl, wherein from 1-4 hydrogen atoms on the ring atoms of the phenyl and naphthyl moieties may be independently replaced with a member selected from the group consisting of halo, C_{1-4}alkyl, C_{2-6}alkenyl, C_{2-6}alkynyl, C_{3-8}cycloalkyl, C_{0-4}alkylC_{3-8}cycloalkyl, -CN and $-NO_2$;

L is selected from:

H, -CN, C(=O)$NR^{12}R^{13}$, $(CH_2)_nNR^{12}R^{13}$, C(=NR^{12})$NR^{12}R^{13}$, OR^{12}, $NR^{12}R^{13}$, -NR^{12}C(=NR^{12})$NR^{12}R^{13}$, and NR^{12}C(=NR^{12})-R^{13};

R^{12} and R^{13} are independently selected from:

hydrogen, -OR^{14}, -$NR^{14}R^{15}$, C_{1-4}alkyl, C_{0-4}alkylphenyl, C_{0-4}alkylnaphthyl, $COOC_{1-4}$alkyl, COO-C_{0-4}alkylphenyl and COO-C_{0-4}alkylnaphthyl, wherein from 1-4 hydrogen atoms on the ring atoms of the phenyl and naphthyl moieties may be independently replaced with a member selected from the group consisting of halo, C_{1-4}alkyl, C_{2-6}alkenyl, C_{2-6}alkynyl, C_{3-8}cycloalkyl, C_{0-4}alkylC_{3-8}cycloalkyl, -CN, and $-NO_2$;

R^{14} and R^{15} are independently selected from:

H, C_{1-4}alkyl, C_{2-6}alkenyl, C_{2-6}alkynyl, C_{3-8}cycloalkyl, C_{0-4}alkylC_{3-8}cycloalkyl, C_{0-4}alkylphenyl and C_{0-4}alkylnaphthyl, wherein from 1-4 hydrogen atoms on the ring atoms of the phenyl and naphthyl moieties may be independently replaced with a member selected from the group consisting of halo, C_{1-4}alkyl,

C_{2-6}alkenyl, C_{2-6}alkynyl, C_{3-8}cycloalkyl, C_{0-4}alkylC_{3-8}cycloalkyl, -CN, and $-NO_2$;

and all pharmaceutically acceptable isomers, salts, hydrates, solvates and prodrug derivatives thereof.

2. A compound of claim 1, wherein:

A is selected from:

(a) C_1-C_6-alkyl;

(b) C_3-C_8-cycloalkyl;

(c) $-NR_2R^3$, $R^3C(=NR^2)$-, R^2R^3N-$C(=NR^2)$-, R^2R^3N-$C(=NR^2)$-$N(R^3)$-

(d) phenyl, which is independently substituted with 0-2 R^1 substituents;

(e) naphthyl, which is independently substituted with 0-2 R^1 substituents; and

(f) a monocyclic or fused bicyclic heterocyclic ring system having from 5 to 10 ring atoms, wherein 1-4 ring atoms of the ring system are selected from N, O and S, and wherein the ring system may be subsituted from 0-2 R^1 substituents;

R^1 is selected from:

halo, C_{1-4}alkyl, R2-C(=NR3)-, R2R3N-C(=NR2)-, -CN, $(CH_2)_mNR^2R^3$, $SO_2NR^2R^3$, SO_2R^2, CF_3, OR^2, and a 5-6 membered aromatic heterocyclic system containing from 1-4 heteroatoms selected from N, O and S;

R^2 and R^3 are independently selected from the group consisting of:

H, C_{1-4}alkyl and C_{0-4}alkylaryl,

m is an integer of 0-2;

Y is a member selected from the group consisting of:

a direct link, -CH2-, -C(=O)-, $-N(R^4)$-, $-N(R^4)$CH2-, -C=N(R4)-, $-C(=O)-N(R^4)$-, $-N(R^4)-C(=O)$-, $-SO_2$-, -O-, $-SO_2-N(R^4)$- and $-N(R^4)-SO_2$-;

R^4 is selected from:

H, C_{1-4}alkyl and C_{0-4}alkylaryl;.

D is absent or is a member selected from the group consisting of:

(c) aryl, which is independently substituted with 0-2 R^{1a} substituents; and

(d) a monocyclic or fused bicyclic heterocyclic ring system having from 5 to 10 ring atoms, wherein 1-4 ring atoms of the ring system are selected from N, O and S, and wherein the ring system may be subsituted from 0-2 R^{1a} substituents;

R^{1a} is selected from:

Halo, C_{1-4}alkyl, -CN, $-NO_2$, $(CH_2)_mNR^{2a}R^{3a}$, $SO_2NR^{2a}R^{3a}$, SO_2R^{2a}, CF_3, OR^{2a}, and a 5-6 membered aromatic heterocyclic ring containing from 1-4 heteroatoms selected from N, O and S;

R^{2a} and R^{3a} are independently selected from the group consisting of:

H, C_{1-4}alkyl and C_{0-4}alkylaryl;

E is a member selected from the group consisting of:

$-N(R^5)-C(=O)-$, $-C(=O)-N(R^5)-$, $-N(R^5)-C(=O)-N(R^6)-$, $-SO_2-N(R^5)-$, $-N(R^5)-SO_2-N(R^6)-$ and $-N(R^5)-SO_2-N(R^6)-C(=O)-$;

R^5 and R^6 are independently selected from:

H, C_{1-4}alkyl, C_{0-4}alkylaryl, C_{0-4}alkylheteroaryl, C_{1-4}alkylCOOH and C_{1-4}alkylCOOC_{1-4}alkyl;

G is selected from:

$-CR^7R^8-$ and $-CR^{7a}R^{8a}-CR^{7b}R^{8b}-$

wherein R^7, R^8, R^{7a}, R^{8a}, R^{7b} and R^{8b} are independently a member selected from from the group consisting of:

hydrogen, C_{1-4}alkyl, C_{0-4}alkyl-C_{3-8}cycloalkyl, C_{0-4}alkylaryl, $-C_{0-4}$alkylCOOR^9, $-C_{0-4}$alkylC(=O)NR^9R^{10}, $-N(R^9)COR^{10}$, $-N(R^9)C(=O)R^{10}$, $-N(R^9)SO_2R^{10}$, and common amino acid side chains;

R^9 and R^{10} are independently selected from:

H, C_{1-4}alkyl and C_{0-4}alkylaryl;

J is a member selected from the group consisting of:

a direct link, -C(=O)-N(R^{11})-$(CH_2)_{0-2}$, -N(R^{11})-$(CH_2)_{0-2}$-C(=O)-, and -N(R^{11})-$(CH_2)_{0-2}$;

R^{11} is a member selected from the group consisting of:

hydrogen, C_{1-4}alkyl, C_{2-6}alkenyl, C_{2-6}alkynyl, C_{3-8}cycloalkyl, C_{0-4}alkylaryl, C_{0-4}alkylheterocyclics, CH_2COOC_{1-4}alkyl, CH_2COOC_{1-4}alkylaryl;

G and J together can form a cyclic ring systems.

Z is a member selected from the group consisting of:

(a) aryl, which is independently substituted with 0-2 R^{1b} substituents;and

(b) a monocyclic or fused bicyclic heterocyclic ring system having from 5 to 10 ring atoms, wherein 1-4 ring atoms of the ring system are selected from N, O and S, and wherein the ring system may be subsituted from 0-2 R^{1b} substituents;

R^{1b} is selected from:

halo, C_{1-4}alkyl, -CN, -NO_2, $NR^{2b}R^{3b}$, $SO_2NR^{2b}R^{3b}$, SO_2R^{2b}, CF_3, OR^{2b}, O-CH_2-CH_2-OR^{2b}, O-CH_2-$COOR^{2b}$, N(R^{2b})-CH_2-CH_2-OR^{2b}, N(-CH_2-CH_2-OR^{2b})$_2$, N(R^{2b})-C(=O)R^{3b}, N(R^{2b})-SO_2-R^{3b}, and a 5-6 membered aromatic heterocyclic ring containing from 1-4 heteroatoms selected from N, O and S;

R^{2b} and R^{3b} are independently selected from the group consisting of:

H, C_{1-4}alkyl and C_{0-4}alkylaryl;

L is selected from:

H, -CN, C(=O)$NR^{12}R^{13}$, $(CH_2)_nNR^{12}R^{13}$, C(=NR^{12})$NR^{12}R^{13}$, OR^{12}, -NR^{12}C(=NR^{12})$NR^{12}R^{13}$ and NR^{12}C(=NR^{12})-R^{13};

R^{12} and R^{13} are independently selected from:

hydrogen, -OR^{14}, -$NR^{14}R^{15}$, C_{1-4}alkyl, C_{0-4}alkylaryl $COOC_{1-4}$alkyl, and $COO\text{-}C_{0-4}$alkylaryl; and

R^{14} and R^{15} are independently selected from H and C_{1-4}alkyl.

3. A compound of claim 1, wherein:

A is selected from:

(a) phenyl, which is independently substituted with 0-2 R^1 substituents;

(b) a monocyclic or fused bicyclic heterocyclic ring system having from 5 to 10 ring atoms, wherein 1-4 ring atoms of the ring system are selected from N, O and S, and wherein the ring system may be subsituted from 0-2 R^1 substituents; and

(c) -NR^2R^3, $R^3C(=NR^2)$-, $R^2R^3N\text{-}C(=NR^2)$-, $R^2R^3N\text{-}C(=NR^2)\text{-}N(R^3)$-

R^1 is selected from:

halo, $R^2\text{-}C(=NR^3)$-, $R^2R^3N\text{-}C(=NR^2)$-, $(CH_2)_mNR^2R^3$, $SO_2NR^2R^3$ and SO_2R^2;

R^2 and R^3 are independently selected from the group consisting of:

H and C_{1-4}alkyl;

Y is a member selected from the group consisting of:

a direct link, -CH2-, -C(=O)-, -$N(R^4)$-, -$N(R^4)$CH2-, and -C=N(R4)-,

D is a member selected from the group consisting of:

(a) phenyl, which is independently substituted with 0-2 R^{1a} substituents; and

(b) a monocyclic or fused bicyclic heterocyclic ring system having from 5 to 10 ring atoms, wherein 1-4 ring atoms of the ring system are selected from N, O and S, and wherein the ring system may be subsituted from 0-2 R^{1a} substituents;

R^{1a} is selected from:

Halo and C_{1-4}alkyl;

R^{2a} and R^{3a} are independently selected from the group consisting of:

H, C_{1-4}alkyl, C_{0-4}alkylaryl;

E is a member selected from the group consisting of:

$-N(R^5)-C(=O)-$

R^5 is independently selected from:

H, C_{1-4}alkyl, C_{0-4}alkylaryl and C_{0-4}alkylheteroaryl;

G is selected from:

$-CR^7R^8-$ and $-CR^{7a}R^{8a}-CR^{7b}R^{8b}-$

wherein R^7, R^8, R^{7a}, R^{8a}, R^{7b} and R^{8b} are independently a member selected from from the group consisting of:

hydrogen, C_{1-4}alkyl, C_{0-4}alkyl-C_{3-8}cycloalkyl, C_{0-4}alkylaryl, $-C_{0-4}$alkylCOOR^9, $-C_{0-4}$alkylC(=O) NR^9R^{10}, $-C_{0-4}$alkylC(=O)NR^9-CH_2-CH_2-O-R^{10}, $-C_{0-4}$alkylC(=O)$NR^9(-CH_2-CH_2-O-R^{10}-)_2$, $-N(R^9)COR^{10}$, $-N(R^9)C(=O)R^{10}$, $-N(R^9)SO_2R^{10}$, and common amino acid side chains;

R^9 and R^{10} are independently selected from:

H and C_{1-4}alkyl, wherein the NR^9R^{10} group of R^7, R^8, R^{7a}, R^{8a}, R^{7b} and R^{8b} is optionally cyclized to form a 5-8 membered heterocyclic group;

J is a member selected from the group consisting of:

$-N(R^{11})-C(=O)-(CH_2)_{0-2}$, and $-N(R^{11})-(CH_2)_{0-2}$; own where $(CH_2)_4$

R^{11} is a member selected from the group consisting of:

hydrogen, C_{1-4}alkyl, C_{2-6}alkenyl, C_{0-4}alkylaryl and a C_{0-4}alkylheterocyclic ring;

G and J together can form a cyclic ring systems.

Z is a member selected from the group consisting of:

(a) phenyl, which is independently substituted with 0-2 R^{1b} substituents;

(b) an aromatic heterocyclic ring having from 5 to 10 ring atoms, wherein 1-4 ring atoms are selected from N, O and S, and wherein the ring may be subsituted independently by from 0-2 R^{1b} substituents; and

(c) a fused aromatic bicyclic heterocyclic ring system having from 5 to 10 ring atoms, wherein 1-4 ring atoms of the ring system are selected from N, O and S, wherein the bicyclic ring system may be subsituted from 0-2 R^{1b} substituents;

R^{1b} is selected from:

halo, C_{1-4}alkyl, OH, OBn, O-CH_2-CH_2-OH, O-CH_2-CH_2-OCH_3, O-CH_2-COOH, O-CH_2-C(=O)-O-CH_3, NH_2, NH-CH_2-CH_2-O-CH_3, NH-C(=O)-O-CH_3, and NH-SO_2-CH_3;

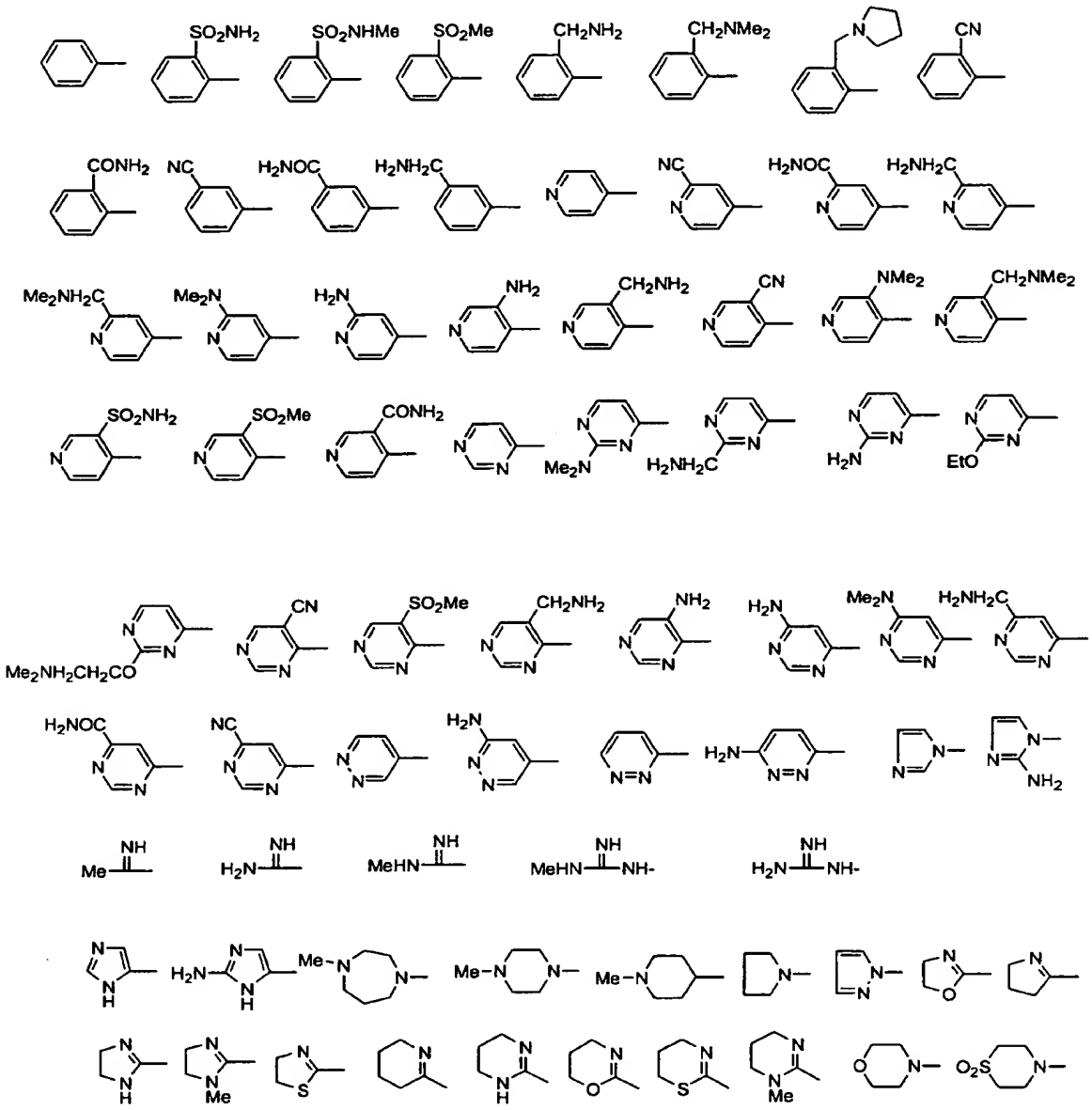
L is selected from:

H, $C(=O)NR^{12}R^{13}$, $(CH_2)_nNR^{12}R^{13}$ and $C(=NR^{12})NR^{12}R^{13}$; and

R^{12} and R^{13} are independently selected from hydrogen and C_{1-4}alkyl.

4. A compound of claim 1, wherein:

A is a member selected from the groups consisting of:



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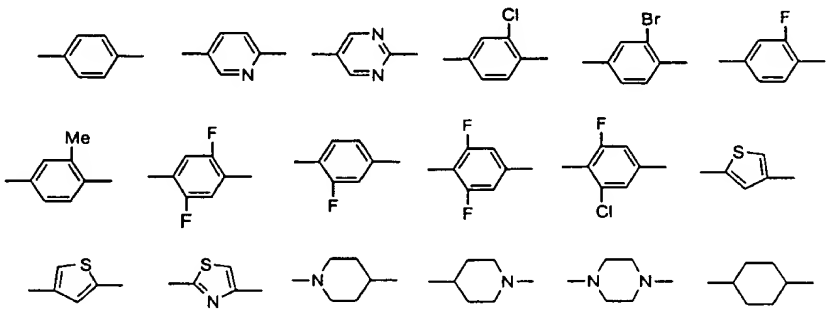
Y is selected from the group consisting of:

a direct link, -CO-, -SO2-, -N(Me)-, -N(Me)-CH2-, CH2-, C(=NH)-, and –C(=NMe)-

D is a direct link or a member selected from the group consisting of:



E is a member selected from the group consisting of::

-C(=O)-NH-, -NH-C(=O)-,

G is selected from:

-CH-(-NH_2)-CH_2-, -CH-(-NH(C(=O)-CH_3))-CH_2-,
-CH-(-NH(C(=O)-Ph))-CH_2-, -CH-(C(=O)-OR^8)-, -CH(-R^7)-,
-CH_2-CH(C(=O)-OR^8)-, and -CH_2-CH(C(=O)-N(-R^8, -R^8))-;

R^7 is a member selected from the group consisting of :

H, C-14alkyl, phenyl, Bn, and cycohexyl;

R^8 is a member selected from the group consisting of:

H, C_{1-6}alkyl, and C_{3-6}cycloalkyl;

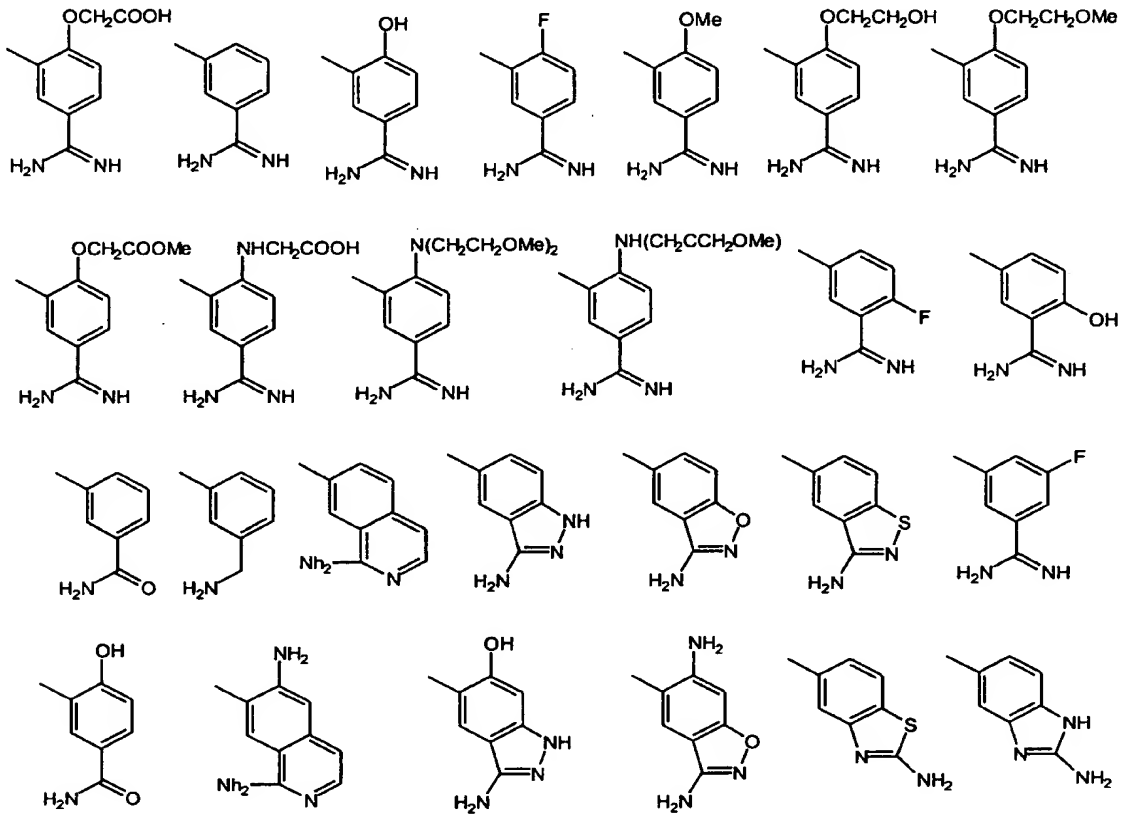
J is a member selected from the group consisting of;

-C(=O)-N(R^{11})-$(CH_2)_{0-2}$, -N(R^{11})-$(CH_2)_{0-2}$-C(=O)-, and -N(R^{11})-$(CH_2)_{0-2}$;

R^{11} is a member selected from the group consisting of:

H, methyl, phenyl and benzyl; and

Z and L taken together are a member selected from the group consisting of:



5. A compound of claim 1, wherein:

A is selected from:

(a) phenyl, which is independently substituted with 0-2 R^1 substituents;

(b) naphthyl, which is independently substituted with 0-2 R^1 substituents;and

(c) a monocyclic or fused bicyclic heterocyclic ring system having from 5 to 10 ring atoms, wherein 1-4 ring atoms of the ring system are selected from N, O and S, and wherein the ring system may be subsituted from 0-2 R^1 substituents;

Y is a direct link;

D is a member selected from the group consisting of:

(a) phenyl, which is independently substituted with 0-2 R^{1a} substituents;

(b) naphthyl, which is independently substituted with 0-2 R^{1a} substituents; and

(c) a monocyclic or fused bicyclic heterocyclic ring system having from 5 to 10 ring atoms, wherein 1-4 ring atoms of the ring system are selected from N, O and S, and wherein the ring system may be subsituted from 0-2 R^{1a} substituents;

E is NH-C(=O)-;

G is $-CHR^{7a}-CHR^{7b}-$;

J is a member selected from the group consisting of:

$-C(=O)-N(R^{11})-(CH_2)_{0-2}$, $-N(R^{11})-(CH_2)_{0-2}-C(=O)-$, and $-N(R^{11})-(CH_2)_{0-2}$;

Z is a member selected from the group consisting of:

(a) phenyl, which is independently substituted with 0-2 R^{1b} substituents;

(b) naphthyl, which is independently substituted with 0-2 R^{1b} substituents; and

(c) a monocyclic or fused bicyclic heterocyclic ring system having from 5 to 10 ring atoms, wherein 1-4 ring atoms of the ring system are selected from N, O and S, and wherein the ring system may be subsituted from 0-2 R^{1b} substituents;

R^{1b} is $N(R^{2b})-C(=O)R^{3b}$; and

L is H.

6. A pharmaceutical composition for preventing or treating a condition in a mammal characterized by undesired thrombosis comprising a pharmaceutically acceptable carrier and a compound of claim 1.

7. A pharmaceutical composition for preventing or treating a condition in a mammal characterized by undesired thrombosis comprising a pharmaceutically acceptable carrier and a compound of claim 2.

8. A pharmaceutical composition for preventing or treating a condition in a mammal characterized by undesired thrombosis comprising a pharmaceutically acceptable carrier and a compound of claim 3.

9. A pharmaceutical composition for preventing or treating a condition in a mammal characterized by undesired thrombosis comprising a pharmaceutically acceptable carrier and a compound of claim 4.

10. A method for preventing or treating a condition in a mammal characterized by undesired thrombosis comprising the step of administering to said mammal a therapeutically effective amount of a compound of claim 1.

11. The method of claim 10, wherein the condition is selected from the group consisting of acute coronary syndrome, myocardial infarction, unstable angina, refractory angina, occlusive coronary thrombus occurring post-thrombolytic therapy or post-coronary angioplasty, a thrombotically mediated cerebrovascular syndrome, embolic stroke, thrombotic stroke, transient ischemic attacks, venous thrombosis, deep venous thrombosis, pulmonary embolus, coagulopathy, disseminated intravascular coagulation, thrombotic thrombocytopenic purpura, thromboangiitis obliterans, thrombotic disease associated with heparin-induced thrombocytopenia, thrombotic complications associated with extracorporeal circulation, thrombotic complications associated with instrumentation such as cardiac or other intravascular

catheterization, intra-aortic balloon pump, coronary stent or cardiac valve, and conditions requiring the fitting of prosthetic devices.

12. A method for preventing or treating a condition in a mammal characterized by undesired thrombosis comprising the step of administering to said mammal a therapeutically effective amount of a compound of claim 2.

13. The method of claim 12, wherein the condition is selected from the group consisting of acute coronary syndrome, myocardial infarction, unstable angina, refractory angina, occlusive coronary thrombus occurring post-thrombolytic therapy or post-coronary angioplasty, a thrombotically mediated cerebrovascular syndrome, embolic stroke, thrombotic stroke, transient ischemic attacks, venous thrombosis, deep venous thrombosis, pulmonary embolus, coagulopathy, disseminated intravascular coagulation, thrombotic thrombocytopenic purpura, thromboangiitis obliterans, thrombotic disease associated with heparin-induced thrombocytopenia, thrombotic complications associated with extracorporeal circulation, thrombotic complications associated with instrumentation such as cardiac or other intravascular catheterization, intra-aortic balloon pump, coronary stent or cardiac valve, and conditions requiring the fitting of prosthetic devices.

14. A method for preventing or treating a condition in a mammal characterized by undesired thrombosis comprising the step of administering to said mammal a therapeutically effective amount of a compound of claim 3.

15. The method of claim 14, wherein the condition is selected from the group consisting of acute coronary syndrome, myocardial infarction, unstable angina, refractory angina, occlusive coronary thrombus occurring post-thrombolytic therapy

or post-coronary angioplasty, a thrombotically mediated cerebrovascular syndrome, embolic stroke, thrombotic stroke, transient ischemic attacks, venous thrombosis, deep venous thrombosis, pulmonary embolus, coagulopathy, disseminated intravascular coagulation, thrombotic thrombocytopenic purpura, thromboangiitis obliterans, thrombotic disease associated with heparin-induced thrombocytopenia, thrombotic complications associated with extracorporeal circulation, thrombotic complications associated with instrumentation such as cardiac or other intravascular catheterization, intra-aortic balloon pump, coronary stent or cardiac valve, and conditions requiring the fitting of prosthetic devices.

16. A method for preventing or treating a condition in a mammal characterized by undesired thrombosis comprising the step of administering to said mammal a therapeutically effective amount of a compound of claim 4.

17. The method of claim 16, wherein the condition is selected from the group consisting of acute coronary syndrome, myocardial infarction, unstable angina, refractory angina, occlusive coronary thrombus occurring post-thrombolytic therapy or post-coronary angioplasty, a thrombotically mediated cerebrovascular syndrome, embolic stroke, thrombotic stroke, transient ischemic attacks, venous thrombosis, deep venous thrombosis, pulmonary embolus, coagulopathy, disseminated intravascular coagulation, thrombotic thrombocytopenic purpura, thromboangiitis obliterans, thrombotic disease associated with heparin-induced thrombocytopenia, thrombotic complications associated with extracorporeal circulation, thrombotic complications associated with instrumentation such as cardiac or other intravascular catheterization, intra-aortic balloon pump, coronary stent or cardiac valve, and conditions requiring the fitting of prosthetic devices.

18. A method for inhibiting the coagulation biological samples, comprising the administration of a compound of claim 1.

19. A method for inhibiting the coagulation biological samples, comprising the administration of a compound of claim 2.

20. A method for inhibiting the coagulation biological samples, comprising the administration of a compound of claim 3.